

DEEP-SEA RESEARCH PART I

Deep-Sea Research I 52 (2005) 2285-2300

www.elsevier.com/locate/dsr

Inorganic and organic nitrogen cycling in the Southern California Bight

D.A. Bronk^{a,*}, B.B. Ward^b

^aThe College of William and Mary, Virginia Institute of Marine Science, Gloucester Point, VA 23062, USA

^bDepartment of Geosciences, Princeton University, Princeton, NJ 08544, USA

Received 21 January 2004; received in revised form 31 March 2005; accepted 2 August 2005 Available online 5 October 2005

Abstract

On the basis of mass balance calculations performed for nitrogen (N) uptake experiments in the Southern California Bight (SCB), it has been suggested that a significant portion of dissolved inorganic N (DIN) uptake results in the production of dissolved organic N (DON). To investigate this process, the fate of ammonium (NH₄⁺) and nitrate (NO₃⁻) uptake was quantified within the euphotic zone at three coastal stations in the SCB using 15N tracer techniques. Several trends in the fate of DIN and the production of DON were observed. First, production of particulate N (PN), from both NH₄⁺ and NO₃⁻, was quantitatively more important in near surface waters, while DON release dominated within the nitracline. Second, the percentage of gross N uptake released as DON was generally higher when NO₃, rather than NH₄, was the substrate. Third, the percentage of N released as DON was higher at night, relative to the day. Fourth, rates of DON release were significantly correlated to NH₄⁺ regeneration, suggesting that similar mechanisms are responsible for both processes—presumably grazing. The results of this study indicate that the DON pool is a sink for DIN uptake on the time scale of hours. One implication of this finding is that new production estimates based on ¹⁵NO₃ uptake rates will likely underestimate particle flux out of the surface layer because the rate of NO₃ uptake is underestimated due to loss of DO¹⁵N during the incubation. On time scales of months to years, however, the N that is taken up as NO₃ and released as DON will likely contribute to export flux via incorporation of the dissolved phase during seasonal mixing into sinking particles or transport. The export of DON on these time scales argues for the use of gross uptake rates to calculate f-ratios. © 2005 Elsevier Ltd. All rights reserved.

Keywords: Nitrogen; Uptake; DON; Phytoplankton; Bacteria

1. Introduction

The Southern California Bight (SCB) has been the site of research into the marine nitrogen (N) cycle for decades (Eppley et al., 1979a, b; Ward, 1987; reviewed

*Corresponding author. Tel.: +18046847779; fax: +18046847786.

E-mail addresses: bronk@vims.edu (D.A. Bronk),

bbw@princeton.edu (B.B. Ward).

by Eppley, 1992). In the SCB, nutrient regimes range from eutrophic to oligotrophic with intermittent zones of upwelling, making it an ideal area to pursue research questions on new and regenerated production. New production is of particular interest because of its relationship to export flux and the sequestration of carbon (C) in marine sediments over long time-scales (Chester, 2003).

One common way to estimate new production in the SCB and elsewhere is to use ¹⁵N tracer

techniques to quantify NO₃ uptake rates. Implicit in this approach is the assumption that the uptake of ¹⁵N-label results in the production of particulate N (PN). However, a significant flux into dissolved organic N (DON) during ¹⁵N tracer studies has been hypothesized for decades based on deficits in ¹⁵N mass balances (Glibert et al., 1982; Laws, 1984; Ward et al., 1989; Slawyk et al., 1990). In an extreme case, Eppley and Renger (1992) documented missing ¹⁵N seven-fold greater than the N incorporated into biomass. More recently, several researchers, in a range of environments, have found that an important fate of NO₃ uptake is DON release, in addition to PN production (Bronk and Glibert, 1991; reviewed by Bronk, 2002; Varela et al., 2003). Prior research findings in the SCB provide evidence that DON release may be quantitatively important in the SCB as well. Ward et al. (1989) used ¹⁵N tracer techniques to measure uptake of NO₃ and NO₂ in the SCB and consistently observed a gradual loss of up to 98% of the ¹⁵N-label, added as either NO₃ or NO₂, over the course of a 24 h incubation. They hypothesized that a N pool, other than the PN, NO₃, and NO₂ pools measured, was a sink for the missing ¹⁵N—a likely candidate was the DON pool. The study presented here was undertaken to address this hypothesis, and to quantify the significance of DON release by planktonic assemblages in the SCB.

1.1. Site description

Two offshore stations, 205 (April 1994) and 305 (October 1992), and one nearshore station, 303 (October 1992), were occupied. The depth of the water at the offshore station occupied in April (33°18.7′ N, 118°9.6′ W, 53 km from shore) and the offshore station occupied in October (33°45' N, 118°47′ W, 46 km from shore) was approximately 920 m. The stations were under the influence of the California Current; however, water temperatures at these stations are typically higher than those observed in the main flow of the Current. The depth of the water at the nearshore station (33°35' N, 118°31′ W, 5.6 km from shore) was approximately 50 m, and was located over the narrow continental shelf; this shelf region typically has higher chlorophyll a (chl a) and particulate matter concentrations relative to the offshore regions of the SCB (Mullin, 1986).

All of these stations were located within the California Cooperative Oceanic Fisheries Investiga-

tions (CalCOFI) program grid, which has included measurements of chl a and primary production since 1969 and measurements of plankton standing stocks since 1974 as part of the Southern California Bight Study. All of the stations were within the region of the "Northern Inshore" grouping defined by Venrick (1998). Venrick (1998) found that diatoms dominated the phytoplankton community in the region of our stations during April 1993 and 1995; we assume that the community composition was similar during our cruises.

The Inner Bight region of the SCB, where these experiments were conducted, has been extensively studied and is characterized annually by oligotrophic conditions as defined by biomass and nutrient data (reviewed in Azam, 1986; Eppley and Holm-Hansen, 1986; Williams, 1986). The water column within the euphotic zone at these sites has two layers; a surface layer, where inorganic N concentrations are near the limit of detection, and a nitracline layer, where NO₃ concentrations increase significantly with increasing depth.

1.2. Research objectives

The specific objectives of this study were to quantify rates of (1) net and gross NH₄⁺ and NO₃⁻ uptake and (2) DON release resulting from uptake of both NH₄⁺ and NO₃⁻. We use "DON release" to refer to the production of DON from labeled dissolved inorganic N (DIN), regardless of the mechanism involved. Used in this way, the term "release" does not simply imply passive release by phytoplankton but includes both exudation and grazing losses. These rates were characterized within the two layers of the euphotic zone, the surface layer and the nitracline. To accomplish these objectives, vertical profiles were performed at the three stations during two cruises. Ambient nutrient and chl a concentrations were quantified and N flux rates were measured using 15N tracer techniques. Additional NH₄ and NO₃ uptake experiments were also carried out in 20 L carboys, at an offshore and a nearshore site on the October cruise, to allow a larger set of variables to be measured simultaneously.

2. Methods

2.1. Field sampling

Water was collected at each station using 10 or 30 L Niskin or Go-Flo bottles, with five to six

sample depths chosen to span the range of N and light environments within the euphotic zone (Table 1). The euphotic zone was defined as the surface down to the 1% light depth, which was assumed to be 2.7 times the Secchi depth. Day incubations were initiated approximately at midday, and night incubations were initiated at dusk.

2.2. Ambient conditions

Water from each depth was filtered through precombusted (450 °C for 2h) Whatman GF/F filters. The filter was retained and used to measure the concentration of chl a after grinding the filter in acetone and allowing the ground filter to extract in acetone overnight (Parsons et al., 1984). The filtrate was frozen for later determination of nutrient concentrations; all samples for the project were run within 18 months of collection and generally within 2–3 months. Concentrations of NO₃ and NO₂ were measured with a Technicon AutoAnalyzer and concentrations of NH₄⁺ were measured manually with the phenol/hypochlorite method (Grasshoff et al., 1999). Concentrations of DON, defined as organic N passing through a 0.2 µm Supor filter, were measured with UV oxidation using a 1200 watt Hg vapor lamp, H₂O₂ as an oxidant, and 18h of irradiation (Armstrong and Tibbitts, 1968; Bronk et al., 2000). Concentrations of PN and particulate C (PC) were measured on precombusted GF/F filters collected at the end of the incubation; PN and PC filters were analyzed with a Control Equipment CHN Analyzer (Grasshoff et al., 1983).

2.3. N uptake and NH_4^+ regeneration rates

Rates of NH₄⁺ and NO₃⁻ uptake were measured with ¹⁵N tracer techniques using 0.1 μM additions

for both NH₄⁺ and NO₃⁻ incubations (Bronk and Ward, 1999); additions were 5-95% of the ambient pool. Experiments were done in 4L polycarbonate bottles under simulated in situ light conditions, and samples were incubated for 4-6h in on-deck flowthrough incubators (Ward and Bronk, 2001). At the end of each incubation, samples were filtered through precombusted GF/F filters which were subsequently dried at 50 °C and ampoulated using the micro-Dumas method (Barsdate and Dugdale, 1965). The PN atom % enrichments were determined using either a Jasco emission spectrometer (model N-150; Fiedler and Proksch, 1975) or a Europa 20/20 mass spectrometer with an ANCA preparatory unit. A subset of PN samples that were run on both instruments had a mean CV of 7.2% with neither instrument consistently producing higher or lower values; these data include betweensample variability and analytical variability. The filtrate from the NH₄⁺ incubations was collected and frozen for later determination of 15N atom % enrichment of the NH₄⁺ using steam distillation (Glibert et al., 1982).

2.4. DON isolation and DON production rates

The DON pool was isolated using the protocol described in Bronk and Ward (1999). Briefly, at the end of each incubation, an aliquot from each of the ¹⁵NH₄⁺ and ¹⁵NO₃⁻ incubations was passed through a 0.2 µm Supor filter and frozen for later isolation of the DON pool. In the lab, DON was isolated with a series of chemical manipulations designed to remove the ¹⁵N-labeled inorganic NH₄⁺ or NO₃⁻ present in the sample. To isolate the DON pool from NH₄⁺ incubations, vacuum distillation was used (Glibert et al., 1982). In the case of filtrate from NO₃⁻ incubations, filtrate was heated in the presence of

Table I						
Station locations	and	times	and	types	of	experiments

Station	Description	Location	Date	Time	Experiment
205	Offshore	33.17° N 118.09° W	19-04-1994	Night	Vertical profile
205	Offshore	33.17° N 118.09° W	20-04-1994	Day	Vertical profile
305	Offshore	33.45° N 118.47° W	11-10-1992	Day	Vertical profile ^a
305	Offshore	33.45° N 118.47° W	12-10-1992	Day	Vertical profile
305	Offshore	33.45° N 118.47° W	12-10-1992	Night	Vertical profile
305	Offshore	33.45° N 118.47° W	14-10-1992	Day	20 L carboy
303	Nearshore	33.53° N 118.31° W	15-10-1992	Day	Vertical profile
303	Nearshore	33.53° N 118.31° W	15-10-1992	Day	20 L carboy

^aAmbient concentrations were measured but no rate measurements were done.

DeVarda's Alloy to convert NO₃⁻ to NH₄⁺, which was then lost through volatilization. The remaining DON concentrates were UV oxidized for 18 h (Armstrong and Tibbitts, 1968); after the oxidation, all DON in the sample were in the form of NO_3^- . The NO_3^- was reduced to NO_2^- by shaking the sample with spongy cadmium for 1.5 h (Jones, 1984) and then the NO₂ produced was isolated with the organic extraction method of Olson (1981). The isolated DON, now in the form of an azo dye dissolved in methylene chloride, was concentrated by evaporation in a fume hood, spotted onto a precombusted GF/F filter, and the 15N atom % enrichment was measured with an emission spectrometer. Recovery of DON from NH₄⁺ and NO₃⁻ incubations were $89 \pm 35.4\%$ and $62.0 \pm 21.7\%$, respectively. The CV for replicate DON atom % analyses was 3.8 ± 3.3 for NH₄⁺ incubations and 6.6 + 6.7 for NO_3^- incubations.

Atom % enrichments of DON were corrected for possible residual NH_4^+ or NO_3^- present in the final isolated DON fraction. Removal of NH_4^+ and NO_3^- with the method used is 100% as determined using wet chemical analyses of NH_4^+ and NO_3^- . However, these wet chemistry analytical methods have limits of detection of $\sim 0.03-0.05\,\mu M$. To guarantee that rates of DON release were not overestimated due to a small but analytically undetectable amount of labeled inorganic N remaining in the isolated DON fraction, a correction was performed as described in Bronk and Glibert (1991). The ^{15}N atom % enrichment of any calculated residual NH_4^+ or NO_3^- was taken to be the same as that measured at the end of the incubation.

Individual ¹⁵N incubations were not routinely duplicated for the vertical profiles, though all chemical analyses of the various N concentrations and ¹⁵N atom % enrichments were performed in duplicate or better. A propagation of error analysis was done to estimate the error associated with rate measurements in the vertical profiles (Bevington, 1969); this method provides a conservative estimate of the variance.

2.5. Large carboy experiments

One carboy was set up for each treatment (NH_4^+ and NO_3^-) using surface seawater at both the offshore and nearshore stations in October. All rate determinations from the 20 L carboy experiments were run in duplicate and subsamples were analyzed with the same procedures used on the samples from

the vertical profiles, except with respect to DON isolation (see below). In the 20 L carboys, uptake by cells that passed through the GF/F filter (nominal pore size 0.7 µm) was also measured. In this size fraction, ¹⁵N-label can be incorporated through direct uptake of ¹⁵N-labeled NH₄⁺ or NO₃⁻ or via uptake of recently released 15N-labeled DON (Bronk and Glibert, 1994). To estimate the uptake of ¹⁵N-label by plankton that passed through the GF/F filter, DON in the GF/F filtrate was isolated and the 15N atom % enrichment was determined and compared to the DON in the 0.2 µm filtrate as described in Bronk and Glibert (1994). The ¹⁵N in this fraction would include any 15N in the DON pool as well as any 15N in organisms that passed through the GF/F filter. Microscopy confirmed that these organisms were primarily bacteria. To calculate the mass of ¹⁵N present in this bacteria pool, the 15 N present in the DON pool (< 0.2 µm filtrate) was subtracted from the ¹⁵N in the combined DON+ bacteria pool (GF/F filtrate) as discussed in Bronk and Glibert (1994). We note that this does not represent total bacterial uptake but only incorporation of label into those cells that were not trapped on the GF/F filter.

In the 20 L carboy samples, DON was isolated using ion retardation resin as described in Bronk and Glibert (1991, 1993). The ion retardation resin (BioRad AG 11 A8) attracts small charged molecules while allowing DON to pass. We note that the manufacturer has changed the production process of the BioRad AG 11 A8 resin such that it now retains variable amounts of DON; neither the original resin nor the resin we used in this study retained DON (reviewed by Bronk, 2002).

2.6. Nitrogen rate calculations

Net uptake rates of NH₄⁺ and NO₃⁻ were calculated according to Dugdale and Goering (1967) with NH₄⁺ uptake rates corrected for isotopic dilution (Glibert et al., 1982). Rates of NH₄⁺ regeneration were calculated according to Glibert et al. (1982). To calculate gross NH₄⁺ and NO₃⁻ uptake rates, the gross atom % enrichment of the PN, which included ¹⁵N measured in both the PN and the extracellular DON pools, was calculated (Bronk et al., 1994). In Bronk et al. (1994) rates of gross N uptake and DON release were corrected for the small amount of N released as DON during the incubation. To make this correction, an iterative process was used to first estimate DON release

without the correction, and then this estimate was used to further refine the DON release rate. This correction was found to be insignificant and so was not done here or on other recently published studies including Bronk et al. (1998), Bronk and Ward (1999), and Ward and Bronk (2001). To calculate the final gross uptake rate, the gross PN atom % enrichment was substituted for the net PN atom % enrichment used in the traditional uptake equation (Bronk et al., 1994, 1998). The rate of DON release was determined as the difference between the gross and net uptake rate of NH₄⁺ or NO₃⁻ (Bronk et al., 1994, 1998). Note that this method of calculation (Bronk et al., 1994) is identical to the later protocol introduced by Slawyk et al. (1998) to calculate the rate of DIN loss ($\rho_{\text{DIN}}^{\text{loss}}$). The following equations were used in the calculations:

Net uptake rate =
$$\rho = \frac{\text{PN at\%}}{\text{DIN at\%} \times \text{Time}} \times [\text{PN}],$$
 (1)

Gross uptake rate

$$= \rho_{\rm G} = \frac{({\rm PN~at\%~\times [PN]}) + ({\rm DON~at\%~\times [DON]})}{{\rm DIN~at\%~\times Time}}, \eqno(2)$$

DON release rate =
$$\rho_{\rm G} - \rho = \frac{\rm DON~at\%~\times [DON]}{\rm DIN~at\%~\times Time}$$
, (3)

where PN, DIN, and DON at% are the ¹⁵N atom % enrichments of PN, DIN and DON pools, respectively. Time is the incubation time. Brackets [] denote concentrations. Daily rates were calculated as the day rate multiplied by 14 h (light period) plus the night rate multiplied by 10 h (dark period).

We investigated the use of ¹⁵N mass balances as a way of estimating rates of gross uptake and DON release as an alternative to the labor intensive, and therefore expensive, process of quantifying the rates directly. In this approach, the ¹⁵N in the PN and substrate (NH₄⁺ or NO₃⁻) pools were summed and compared to the ¹⁵N added at the start of the experiment. The assumption was made that any missing ¹⁵N was transferred to the DON pool, and a DON atom % and a corrected PN atom % enrichment were then calculated in the same fashion as if ¹⁵N in the DON pool had been measured directly (Bronk and Ward, 1999). This procedure produced gross uptake rates that were overestimated by a mean factor of 4.1, relative to gross

uptake rates measured directly, when all incubations were combined in the analysis. Similarly, estimates of DON release rates calculated by mass balance were significantly overestimated in all cases (mean factor of 15). We conclude that ¹⁵N mass balances cannot be used to estimate rates of gross N uptake or DON release.

3. Results

Vertical profiles of ambient N concentrations, rates of net and gross N uptake, DON release, and NH_4^+ regeneration are presented below for all three stations. These results are discussed in relation to characteristics in the surface layer (\sim upper 25 m) in contrast to characteristics within the nitracline, the deepest two sample points at each site. Results are also presented from $^{15}NH_4^+$ and $^{15}NO_3^-$ incubations in 20 L carboys conducted in October.

3.1. Vertical profiles of ambient conditions

Concentrations of NO₃ and NO₂ were depleted in the upper $\sim 20 \,\mathrm{m}$ at all three study sites (Fig. 1A–F). The top of the nitracline was at 27 m at the offshore station in April, 23 m at the offshore station in October, and 20 m at the nearshore station during this study, based on more detailed vertical profiles (Fig. 1 or data not shown). Concentrations of NH₄⁺ were very low but measurable in the near-surface waters in all vertical profiles. At the offshore station in April, NH₄⁺ concentrations increased several-fold within the nitracline coincident with increases in NO₂ concentrations (Fig. 1F and I). Concentrations of DON were relatively constant throughout the water column at all three stations, ranging from 4.7 to 7.3 µM (Fig. 1J–L). A deeper, more detailed profile during the day at the offshore station in October showed additional variability in the DON concentrations above the nitracline and a clear subsurface accumulation of DON (Fig. 2).

3.2. Vertical profiles of N flux rates

At both offshore stations, rates of gross NO_3^- uptake and DON release increased several-fold within the nitracline (Figs. 3 and 4). The dominant N form producing biomass (i.e. PN) in both April and October was NH_4^+ . Net NH_4^+ uptake rates averaged 5 and 12 times higher than parallel net NO_3^- uptake rates at the offshore stations in April

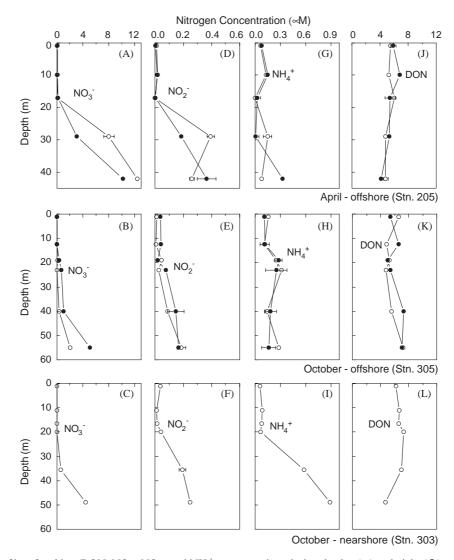


Fig. 1. Vertical profiles of ambient DON, NO_3^- , NO_2^- , and NH_4^+ concentrations during the day (\bigcirc) and night (\bigcirc) at Station 205 (April, upper panels) and Stations 305 and 303 (October, lower panels). Error bars denote standard deviations; where error bars are not seen, errors are smaller than the symbols.

and October, respectively (Figs. 3 and 4). Note that NH_4^+ uptake rates have been corrected for isotope dilution while NO_3^- uptake rates have not; if nitrification were occurring in the water column, the nitrate uptake rate would be underestimated, thereby contributing to the difference between NH_4^+ and NO_3^- uptake. At all sites, the percentage of gross uptake released as DON tended to increase with depth. In contrast to the offshore stations, DON release from NH_4^+ exceeded its release from NO_3^- at the nearshore station, while net uptake of NH_4^+ peaked at 20 m, DON release remained high at depths greater than 20 m (Fig. 5).

Flux rates were integrated within the upper \sim 23 m (depending on the profile as noted above) and then within the nitracline to the base of the euphotic zone, defined as the 1% light depth and represented by the deepest sample collected at each profile. Several trends emerged. First, the production of PN was higher in the surface layer while the production of DON was higher in the nitracline (Fig. 6). Second, a higher percentage of N was released as DON when NO₃ was the substrate, relative to NH₄⁺ (Fig. 6). Third, in seven of eight cases, the percentage of DON released from both NH₄⁺ and NO₃ was higher at night than during the

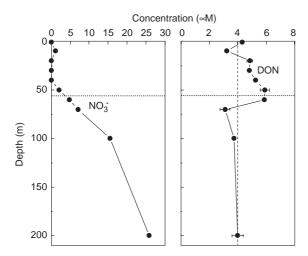


Fig. 2. Vertical profiles of NO_3^- and DON concentrations at Station 305. Error bars denote standard deviations; where error bars are not shown, errors are smaller than the symbols. The dashed line marks the 1% light depth.

day (Fig. 7). Specific uptake rates, for both NH_4^+ and NO_3^- , were also higher during the day than at night, except for the 1 m NH_4^+ uptake sample in April (data not shown). Fourth, with all rates taken together, rates of net NH_4^+ uptake were positively correlated to gross NH_4^+ uptake ($r^2 = 0.92$), but rates of net NO_3^- uptake did not correlate well with gross NO_3^- uptake. However, when only the surface layer was considered, net and gross NO_3^- uptake rates were closely correlated ($r^2 = 0.91$) at the offshore station in April and the nearshore station in October (Figs. 4 and 6).

In general, NH_4^+ regeneration exceeded both net and gross NH_4^+ uptake at all depths and the ratio of regeneration to uptake increased with depth (data not shown). Rates of NH_4^+ regeneration were greater than rates of DON release at all depths, representing upwards of 86% of the N released within the euphotic zone (data not shown). The ratio of integrated NH_4^+ regeneration: DON release, however, was higher in the surface waters (19.6–31.6) than within the nitracline (6.5–6.6). Rates of NH_4^+ regeneration and DON release, resulting from NH_4^+ uptake, were significantly correlated ($r^2 = 0.73$; n = 19; p < 0.001).

3.3. Large carboy experiments

Inorganic N concentrations were three times higher in the nearshore carboys than in the offshore carboy (Table 2). Concentrations of DON, chl *a*,

and PN were similar at both stations, but the C:N ratio of the particulate material was significantly higher nearshore (Table 2). Consistent with the vertical profiles (Fig. 4 and 5), gross uptake rates of NH₄⁺ were significantly higher than NO₃⁻ at both sites (Table 3). Rates of NH₄⁺ and NO₃⁻ uptake into the bacterial size fraction, defined as the 0.2–0.7 µm fraction, were over three times higher offshore (Table 3). Bacterial uptake of NH₄⁺ and NO₃⁻, as a percentage of total NH₄⁺ and NO₃⁻ uptake (defined as uptake into cells $> 0.2 \,\mu\text{m}$), was seven times higher offshore (Table 3). More DON was released as a result of NH₄⁺ uptake, relative to NO₃⁻ uptake, at both sites (Table 3). Rates of DON release were similar in magnitude at both sites though the percentage of gross N uptake that was released as DON was two times higher offshore than nearshore. In contrast, rates of NH₄⁺ regeneration were similar at both sites (Table 3).

3.4. Turnover times

Turnover times estimated from depth profile experiments for DIN, PN and DON ranged from less than a day to several weeks. In general, turnover times were shorter in October than in April (Table 4). The NH₄⁺ pool (and NO₃⁻ in October) had the shortest turnover times and the DON pool had the longest (Table 4).

4. Discussion

The objectives of this study were to quantify rates of net and gross NH₄⁺ and NO₃⁻ uptake and DON release in the SCB. The results show that, in all experiments at all depths, DON was an important fate for DIN uptake. In this section, we compare our results to those from other systems, highlight the importance of grazing in DON release, present evidence for bacterial DIN and DON uptake, and discuss how DON release affects export flux and its measurement.

4.1. DON release across systems

In this study, the percentage of N uptake released as DON was generally higher in NO₃ incubations compared to NH₄ incubations, and DON release as a percentage of gross N uptake was relatively low in the surface layer but increased significantly within the nitracline; both observations are consistent with findings in Monterey Bay, CA (Bronk and Ward,

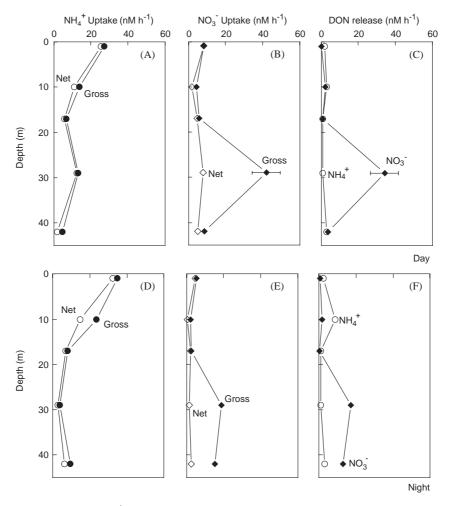


Fig. 3. Vertical profiles of day and night NH_4^+ gross (\bullet) and net (\circ) uptake rates and NO_3^- gross (\bullet) and net (\circ) uptake rates and DON release resulting from uptake of NH_4^+ (\circ) and NO_3^- (\bullet) at Station 205 in April. Error bars denote standard deviations; where error bars are not seen, errors are smaller than the symbols.

1999). In the Gulf of Lyons (NW Mediterranean), however, the magnitude of DON release decreased with depth and did not vary consistently with N substrate—an average of 26% and 24% of NH₄⁺ and NO₃, respectively, was taken up and released as DON (Diaz and Raimbault, 2000). Furthermore, in the Gulf of Lyons, net and gross uptake rates were closely correlated in both NH_4^+ (slope = 0.74; $r^2 = 0.95$) and NO₃ incubations (slope = 0.79; $r^2 = 0.93$; Diaz and Raimbault, 2000). In the SCB, rates of net and gross NH₄⁺ uptake were also closely correlated (slope = 0.82; $r^2 = 0.92$), but a linear relationship was not observed between net and gross NO_3^- uptake. If only the surface waters are considered for the SCB data, however, the relationship was much closer to linear ($r^2 = 0.91$) providing evidence that within the nitracline, NO₃ uptake and biomass production were uncoupled.

At both offshore sites where day and night rates were measured, the percentage of gross N uptake resulting in the production of DON was higher at night than during the day (Fig. 7). A similar trend was also observed during three diel studies in Chesapeake Bay where rates of DON release were higher at dusk than at other times of the day (Bronk et al., 1998). In addition to the vertical profiles and large carboy studies reported here, we also did size-fractionation experiments on different days of each cruise and reported the results previously (Ward and Bronk, 2001). In the size-fractionation studies, DON release rates were higher at night than during the day in seven of 10 experiments in October but in

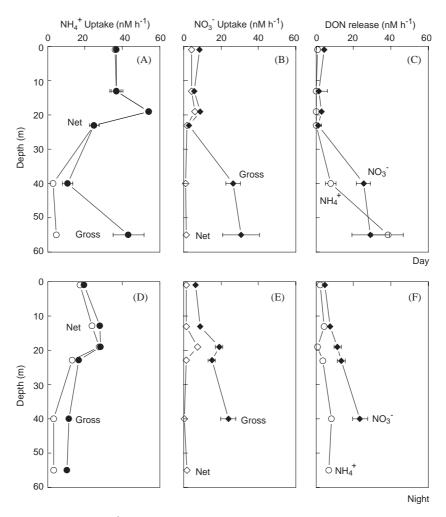


Fig. 4. Vertical profiles of day and night NH_4^+ gross (\bullet) and net (\circ) uptake rates and NO_3^- gross (\bullet) and net (\circ) uptake rates and DON release resulting from uptake of NH_4^+ (\circ) and NO_3^- (\bullet) at Station 305 in October. Error bars denote standard deviations; where error bars are not seen, errors are smaller than the symbols.

only one of seven experiments in April (Ward and Bronk, 2001). We also note dramatic differences in concentrations and rates measured on the days when vertical profiles were performed when compared to days when the size-fractionation experiments were done. The ratio of the concentration or rate measured during the size-fractionation experiment divided by the concentration or rate measured at the closest depth during the vertical profile varied from a mean of 1.2-6.0 for NO₂ concentrations, 1.5-6.3 for NO₃ concentrations, 1.2-3.3 for DIN uptake, 5.2-15.1 for DON release and 0.6-3.0 for NH₄⁺ regeneration. These differences likely reflect changes in the water column, including an apparent shallowing of the nitracline, as well as the large variability inherent in biological rate measurements.

Similar to DON release rates, NH₄⁺ regeneration rates were also higher at night than in the day during the April cruise presented here; an incomplete NH₄⁺ regeneration profile precluded any day/night comparison in October. Ward and Bronk (2001) found that NH₄⁺ regeneration was higher at night than during the day in three of four sizefractionation experiments performed in the SCB in April and five of six experiments in October. DON release, resulting from NH₄⁺ uptake, and NH₄⁺ regeneration were also found to be significorrelated $(r^2 = 0.73; n = 19; p < 0.001).$ cantly The Model II regression slopes suggest that DON release was approximately 40% of the measured NH₄⁺ regeneration rate. Similar trends were also observed during our size-fractionation

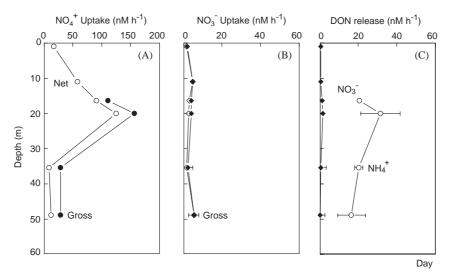


Fig. 5. Vertical profiles of day NH_4^+ gross (\bullet) and net (\circ) uptake rates and NO_3^- gross (\bullet) and net (\diamond) uptake rates and DON release resulting from uptake of NH_4^+ (\circ) and NO_3^- (\bullet) at nearshore Station 303 in October; uptake rates were not measured at night. Error bars denote standard deviations; where error bars are not seen, errors are smaller than the symbols.

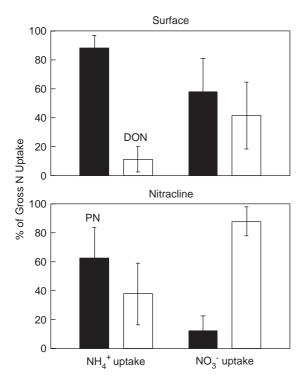


Fig. 6. The percentage of integrated gross nitrogen uptake that resulted in the production of PN (\blacksquare) or DON (\square) in the surface layers and within the nitracline layer for the offshore vertical profiles in April and October.



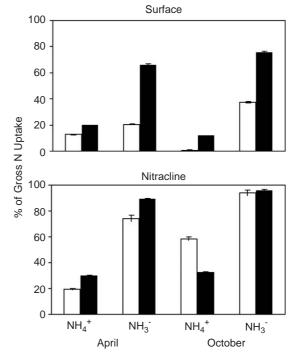


Fig. 7. The percentage of integrated gross nitrogen uptake that resulted in the production of DON in the surface layers and within the nitracline during the day (\square) and at night (\blacksquare). Error bars represent a propagation of error analysis.

coastal waters, DON release was approximately 59% of NH₄⁺ regeneration (Hasegawa et al., 2000b).

Table 2 Ambient conditions in the 20 L carboys containing offshore and nearshore water from 10 m in the Southern California Bight in October

Concentration	Offshore	Nearshore
$NH_4^+ (\mu M) \\ NO_3^-/NO_2^- (\mu M) \\ DON (\mu M)$	0.08 ± 0.00 0.03 ± 0.01 7.1 ± 0.4	0.21 ± 0.01 0.10 ± 0.01 6.7 ± 0.05
chlorophyll a (µg L^{-1}) PC (µmol C L^{-1}) PN (µmol N L^{-1}) C:N of particulate	0.23 ± 0.01 9.17 ± 0.50 1.14 ± 0.11 8.0	0.24 ± 0.02 15.4 ± 0.50 1.25 ± 0.13 12.3

Table 3 Gross and net nitrogen uptake and DON release rates for the offshore and nearshore 20 L carboy experiments performed in the Southern California Bight in October

	Offshore	Nearshore
Gross uptake (>0.7 μm)	$(nM h^{-1})$	71.2 + 1.0
NH ₄ ⁺ NO ₃ ⁻	_	71.3 ± 1.0 8.7 ± 0.1
Gross bacterial uptake (0.2–0.7 μm)	(nMh^{-1})	
NH ₄	1.9 ± 0.9	0.4 ± 0.7
NO_3^-	0.7 ± 0.7	0.2 ± 0.1
Bacterial uptake: total uptake		
NH ₄ ⁺	4.7%	0.6%
NO_3^-	14.7%	2.4%
DON release	(nMh^{-1})	
from NH ₄ ⁺ uptake	10.3 ± 0.4	11.9 ± 0.6
from NO ₃ uptake	2.4 ± 0.2	2.2 ± 0.1
DON release : Gross uptake in the 0.2 µm fraction		
NH_4^+	25.1%	16.6%
NO_3	50.0%	24.7%
NH ₄ ⁺ regeneration	23.4 ± 2.8	27.8 ± 3.4

Standard deviations come from replicate bottles.

4.2. Evidence for the importance of grazing to DON release

The robust correlations between DON release and NH₄⁺ regeneration, and the tendency for higher DON production rates at night, suggest that both NH₄⁺ regeneration and DON release are likely influenced, on some level, by the same process—the most likely of which is grazing (reviewed in Nagata, 2000 and Bronk, 2002; Hasegawa et al., 2000a,

2001). The importance of grazing in DOM production was predicted in a study by Jackson and Eldridge (1992) where inverse modeling was used to estimate the rates of N and C uptake and release, DON and DOC production, and grazing in the microbial and protozoan size fractions. The model indicates that grazing is responsible for a major flux of material from phytoplankton through the detrital pool and from there into DOC and DON. Jackson and Eldridge (1992) conclude that exudation or passive DOM release is not important, but that grazing-induced fluxes into DOM are at least as large as direct phytoplankton biomass consumption by grazers. Unfortunately, there are few direct measurements of DON release during grazing though there are a number of studies that measure grazer mediated release of DOC (Dagg, 1974; Lampert, 1978; Urban-Rich, 1999). One study in Japanese coastal waters that was specific to DON showed that, when chl a concentrations are high $(>6 \,\mu g \, chl \, L^{-1})$, the addition of copepods increases the rate of DON release, but the relationship does not hold at lower chl a levels (Hasegawa et al., 2000a). In another study, Hasegawa et al. (2000b) used an isotope dilution approach to demonstrate that grazing by microzooplankton was an important source of DON.

It is important to stress that DON release during grazing is separated from release during excretion where dissolved organic products are released after ingestion. In our experiments the flow of ¹⁵N-label was traced into the phytoplankton and then released from the cells either directly or as a result of sloppy feeding. We assumed that the shorter incubations times used (\sim 3 h) would largely prevent us from including excretion in our release rates because there would not be enough time for the ¹⁵N-label to be taken up by a phytoplankton cell, consumed by a grazer, and then metabolized and excreted; we are unaware of data supporting or refuting this assumption, however. If this assumption is not correct, then a portion of the release we measured may be due to excretion. Indeed, results from excretion studies are similar to what we observed. At the BATS station, for example, increases in concentrations of DON and NH₄⁺ were monitored over time to estimate net release rates. In experiments with copepods, DON release is 21% of the total N excreted (DON and NH₄⁺ release combined). Combining all data from the SCB study presented here, DON was $11.5 \pm 10.0\%$ of total N regeneration.

Table 4 Turnover times of NH_4^+ , NO_3^- , and dissolved organic nitrogen (DON) estimated during two cruises in the Southern California Bight

Cruise	Depth (m)	Turnover times (days)					
		NO ₃	NH ₄ ⁺	DON	PN		
April	1	0.75	0.09	89.7	0.94		
	10	1.50	0.29	30.2	1.51		
	17	1.50	0.00	156.9	2.61		
	29	6.90^{a}	0.65	7.1 ^a	0.57^{a}		
	42	11.17	0.45	19.2	1.16		
	$mean \pm std$	4.37 ± 4.53	0.30 ± 0.26	60.6 ± 62.5	1.36 ± 0.78		
October	1	< 0.01	0.21	46.8	0.93		
	13	< 0.01	0.14	35.1	1.19		
	19	0.37	0.22	32.1	0.87		
	23	< 0.01	0.54	24.8	1.63		
	40	0.53	0.40	7.1	0.73		
	55	0.45	0.22	3.3	0.28		
	mean ± std	0.30 ± 0.35	0.31 ± 0.15	24.9 ± 16.8	0.94 ± 0.45		

Turnover times of NH_4^+ and NO_3^- were estimated using gross uptake rates. DON turnover times were estimated by combining rates of DON release estimated in incubations with NH_4^+ and NO_3^- . Turnover times for particulate nitrogen (PN) were calculated using ambient PN concentrations and the combined gross NH_4^+ and NO_3^- daily uptake rates.

4.3. Bacterial N uptake

In this study, the amount of ¹⁵N recovered in the bacterial fraction was used to estimate rates of N uptake into that fraction. Bronk and Glibert (1994) showed that a significant portion of ¹⁵N-label can be incorporated by the $\sim 0.2-0.7 \,\mu \text{m}$ size fraction, assumed to include primarily heterotrophic bacteria. The incorporation can occur via direct uptake of inorganic N or via uptake of recently released ¹⁵N-labeled DON (Bronk and Glibert, 1994). In general, the amount of N that accumulates in the bacterial fraction is usually larger when NH₄⁺, rather than NO₃, is the substrate (Bronk and Glibert, 1994). This trend likely reflects the preference by bacteria for NH₄⁺ over NO₃⁻ that has been observed in a number of studies (Kirchman, 1994; reviewed in Kirchman, 2000). In the offshore large carboy experiment presented here, only slightly more ¹⁵N-label was incorporated into the bacterial fraction when NH₄⁺ was the substrate $(2.1\% \text{ for NH}_4^+ \text{ versus } 1.8\% \text{ for NO}_3^-)$. Bacterial uptake of NH₄⁺ is also suggested by the observation that NH₄⁺ regeneration exceeded NH₄⁺ uptake in some experiments. If bacteria < 0.7 µm were taking up NH₄, this uptake would not be included in our measured NH₄⁺ uptake rates resulting in an underestimate of the uptake rate. At the nearshore station, however, slightly more 15N was recovered

in the bacterial fraction when NO₃⁻ was the substrate relative to NH₄⁺; the recovered ¹⁵N could have come from direct uptake of ¹⁵NO₃⁻ or of recently released DO¹⁵N. DON release rates were five times higher when NH₄⁺ was the substrate resulting in higher concentrations of recently released DON in NH₄⁺ incubations. Therefore, the larger amount of ¹⁵N that appeared in the bacterial fraction when NO₃⁻ was the substrate likely reflects higher rates of bacterial utilization of NO₃⁻ rather than recently released DON. Bacterial uptake is increasingly recognized as an important sink for NO₃⁻ in marine surface waters (Wheeler and Kirchman, 1986; Horrigan et al., 1988; Kirchman and Wheeler, 1998; Allen et al., 2002).

4.4. DON release and export flux

The recognition that appreciable amounts of N uptake result in the production of DON rather than PN is important when considering export production. With the exception of isolated sites of downwelling and during seasonal overturn, N must be packaged into particles of sufficient size and density to sink out of the euphotic zone. Transfer of N to a dissolved fraction, particularly one with longer turnover times such as DON (Table 4), will retain N within the more biologically active surface waters for a longer period of time. Here we focus on three

^aNO₃ uptake and DON release from NO₃ uptake were from night experiments only.

aspects of DON release that will affect export production and its estimation: the relative magnitudes of DON release from NH₄⁺ versus NO₃⁻, the fate of recently released DON, and the location of DON release in the water column.

The *f*-ratio is the ratio of new production (commonly assumed to be the rate of NO₃⁻ uptake) divided by the sum of new and regenerated production (commonly assumed to be the rate of NO₃⁻ uptake plus NH₄⁺ uptake and, at times, urea uptake; Dugdale and Goering, 1967; Eppley and Peterson, 1979). Though the common application of the *f*-ratio tends to oversimplify a very complex system it is still a useful index for quickly describing the reliance of a system on different N forms and as a relative indicator of export flux. DON release is important to consider when calculating the *f*-ratio because it will affect the balance of new versus regenerated production.

In this study, more biomass was produced as a result of NH₄⁺ utilization, while NO₃⁻ uptake yielded relatively more DON production at all stations. At our offshore sites, f-ratios increased by up to a factor of four when gross uptake rates were used, reflecting the substantial loss of DON observed when NO₃ was the substrate. Three earlier studies focused on NO₃ uptake rates and new production at the same sites occupied in this study: Eppley and Renger (1986), Ward et al. (1989), and Small et al. (1989). The rates measured using ¹⁵N tracers and 24 h time-courses were based on the accumulation of 15N-label in the cells and are therefore approximations of net NO₃ uptake rates (Eppley and Renger, 1986; Ward et al., 1989). Rates in these studies that were based on NO₃ concentration changes measured with chemiluminscent detection, however, measured NO₃ disappearance regardless of the ultimate fate of the N and are, therefore, approximations of gross NO₃ uptake rates, in the absence of nitrification (Eppley and Renger, 1986; Small et al., 1989). New production estimates made using net uptake rates were $8.3-10.5 \,\mathrm{mg}\,\mathrm{N}\,\mathrm{m}^{-2}\,\mathrm{d}^{-1}$ compared to new production estimates of $17.8-55.5 \,\mathrm{mg} \,\mathrm{N} \,\mathrm{m}^{-2} \,\mathrm{d}^{-1}$ made using gross uptake rates (data pooled from Eppley and Renger, 1986; Ward et al., 1989; Small et al., 1989; this study).

The question then becomes which *f*-ratio is the appropriate one to use in discussions of export flux. We submit that the answer depends on the fate of the released DON and the time scale under consideration. The primary fate of DON is gen-

erally thought to be bacterial utilization. Bacteria do not have appreciable sinking rates such that any N incorporated into their biomass would be retained within the surface layer. On the order of days to weeks, therefore, it would be unlikely that the N taken up as NO₃ and released as DON would contribute to export flux. As a result, the traditionally determined net uptake rates, which measure only PN production, would be the appropriate rates to use in calculating the f-ratio. On the order of months to years, however, it is much more likely that N, released as DON during NO₃ uptake, would be exported to depth. In specific, DON could be transported to depth in three ways-diffusiondriven flux, seasonal overturn and uptake into sinkable particles.

In our deeper more detailed casts, we observed a surface accumulation of DON particularly at the base of the euphotic zone (Fig. 2). This accumulation is similar to the "semi-labile" pool observed in DOC profiles (Kirchman et al., 1993; Carlson and Ducklow, 1995) and its presence suggests that DON uptake and production are uncoupled. A surface enrichment of DON has been observed in a large number of environments (reviewed in Bronk, 2002). and elevated DON concentrations at the surface suggest that DON can be exported to depth during seasonal mixing (Toggweiler, 1989; Hopkinson et al., 1997; Hansell, 2002). Vidal et al. (1999) calculated vertical gradient-driven fluxes of DON in the equatorial Atlantic using vertical profiles of DON concentrations and found that surface DON did appear to be transported to depth.

Incorporation of DON into sinkable particles is another important mechanism for NO₃ uptake released as DON to reach the deep ocean. Historically, it was believed that phytoplankton production was fueled by inorganic N and that DON did not contribute significantly to phytoplankton N nutrition. Although many phytoplankton species were known to have the ability to take up a variety of organic compounds (Berg et al., 1997; reviewed by Antia et al., 1991, Bronk, 2002, and Berman and Bronk, 2003) most studies were done using cultures and large (mM) substrate additions (reviewed in Bronk and Flynn, in press). At the concentrations of DON found in nature, however, it was thought that phytoplankton could not compete with bacteria for the organic substrates. More recent work, however, has shown that many phytoplankton can and do obtain N from organic substrates (reviewed in Bronk, 2002; Berman and Bronk, 2003). If recently

released DON is reincorporated into autotrophic biomass than export out of the surface layer via direct sinking or repackaging into fecal pellets becomes much more likely. In this regard, DON differs markedly from DOC because C can be respired, significantly reducing the likelihood it will be packaged into sinkable particles.

DON production would be especially significant if it occurred deeper in the water column. Diatoms, which have appreciable sinking rates, are common at the base of the euphotic because of the abundant NO₃ (Goldman, 1988). If diatoms could also use recently released DON as a N source, then this is another mechanism for getting DON exported from the surface layer. In contrast, if diatoms are losing a substantial amount of the N they take up as DON, then the affect on the magnitude of N flux out of the euphotic could be great because this N is not packaged into sinkable particles (i.e. diatom biomass) but instead accumulates at the base of the euphotic zone (e.g. Fig. 2). The accumulated DON could, however, be transported to depth during seasonal overturn (Hansell, 2002). The high rates of DON release in the nitracline measured (Fig. 4) and the apparent DON accumulation (Fig. 2) observed supports the contention that DIN uptake at the base of the photic zone does not translate wholly into sinkable PN.

In summary, on the time scale of months to years, DON release, which occurs as a result of new N uptake, could contribute to export flux via incorporation into sinkable particles or seasonal mixing. As a result, gross N uptake rates should be used when calculating the *f*-ratio.

5. Conclusions

A number of studies investigating N uptake and new production in the SCB found circumstantial evidence for substantial DON release including missing ¹⁵N in isotope mass balances (Eppley and Renger, 1986; Ward et al., 1989). Here we provide direct evidence for DON release and its relative magnitude throughout the euphotic zone. When we incorporate our findings into what is known about N cycling in the SCB, the picture emerging is one where N uptake in the surface waters is dominated by NH₄⁺ uptake (Eppley et al., 1979a, b; this study) and the primary fate of DIN uptake is PN production. NH₄⁺ is the most important regenerated N form (the ratio of NH₄⁺ regeneration to DON release was 20–30) and the DON pool is turning

over on the order of weeks to months. Deeper in the water column, approaching the nitracline, the importance of NO₃ increases as a N source, and the dominant fate of NO₃ uptake is DON production resulting in an accumulation of DON, which has a turnover time on the order of days. NH₄⁺ was still the most important regenerated N form but the ratio of NH₄⁺ regeneration to DON release decreased to six. The strong correlation between NH₄⁺ regeneration and DON release and the higher rates of DON release measured point to grazing as an important mechanism for DON release. Finally, in a broader sense, significant production of DON during standard incubation experiments implies that uptake rates based solely on ¹⁵N accumulation in PN underestimate the rate of phytoplankton N assimilation.

Acknowledgments

We thank M. Hogan for excellent technical assistance, and Sandy Moore (OSU) for analyzing CHN samples. We thank the Captains and crews of the R/V Sproul. This research was supported by NSF grant OCE-9115940 to BBW and the writing was supported by OCE-0221825 to DAB. This paper is VIMS contribution No. 2690 from the Virginia Institute of Marine Science, The College of William and Mary.

References

Allen, A.E., Howard-Jones, M.H., Booth, M.G., Frischer, M.E., Verity, P.G., Bronk, D.A., Sanderson, M.P., 2002. Importance of heterotrophic bacterial assimilation of ammonium and nitrate in the Berents Sea during summer. Journal of Marine Systems 38, 93–108.

Antia, N.J., Harrison, P.J., Oliveira, L., 1991. Phycological reviews: the role of dissolved organic nitrogen in phytoplankton nutrition, cell biology, and ecology. Phycologia 30, 1–89.

Armstrong, F.A.J., Tibbitts, S., 1968. Photochemical combustion of organic matter in sea water for nitrogen, phosphorus, and carbon determination. Journal of the Marine Biological Association of the United Kingdom 48, 143–152.

Azam, F., 1986. Nutrient cycling and food web dynamics in the Southern California Bight: the microbial food web. In: Eppley, R.W. (Ed.), Plankton Dynamics of the Southern California Bight. Springer, New York, pp. 274–288.

Barsdate, R.J., Dugdale, R.C., 1965. Rapid conversion of organic nitrogen to N₂ for mass spectrometry: an automated Dumas procedure. Analytical Chemistry 13, 1–5.

Berg, G.M., Glibert, P.M., Lomas, M.W., Burford, M.A., 1997. Organic nitrogen uptake and growth by the chrysophyte *Aureococcus anophagefferens* during a brown tide event. Marine Biology 129, 377–387.

- Berman, T., Bronk, D.A., 2003. Dissolved organic nitrogen: a dynamic participant in aquatic ecosystems. Aquatic Microbial Ecology 31, 279–305.
- Bevington, P.R., 1969. Data Reduction and Error Analysis for the Physical Sciences. McGraw-Hill, New York, pp. 56–65.
- Bronk, D.A., 2002. Dynamics of organic nitrogen. In: Hansell, D.A., Carlson, C.A. (Eds.), Biogeochemistry of Marine Dissolved Organic Matter. Academic Press, New York, pp. 153–247.
- Bronk, D.A., Flynn, K.J., in press. Algal cultures as a tool to study the cycling of dissolved organic nitrogen. In: Durvasula, S.R.V. (Ed.), Algal Cultures, Analogues of Blooms and Applications. Oxford & IBH Publishing Co. Pvt. Ltd., pp. 301–341.
- Bronk, D.A., Glibert, P.M., 1991. A ¹⁵N tracer method for the measurement of dissolved organic nitrogen release by phytoplankton. Marine Ecology Progress Series 77, 171–182.
- Bronk, D.A., Glibert, P.M., 1993. Contrasting patterns of dissolved organic nitrogen release by two size fractions of estuarine plankton during a period of rapid NH₄⁺ consumption and NO₂⁻ production. Marine Ecology Progress Series 96, 291–299.
- Bronk, D.A., Glibert, P.M., 1994. The fate of the missing ¹⁵N differs among marine systems. Limnology and Oceanography 39, 189–194.
- Bronk, D.A., Ward, B.B., 1999. Gross and net nitrogen uptake and DON release in the euphotic zone of Monterey Bay, California. Limnology and Oceanography 44, 573–585.
- Bronk, D.A., Glibert, P.M., Ward, B.B., 1994. Nitrogen uptake, dissolved organic nitrogen release, and new production. Science 265, 1843–1846.
- Bronk, D.A., Glibert, P.M., Malone, T.C., Banahan, S., Sahlsten,
 E., 1998. Inorganic and organic nitrogen cycling in Chesapeake Bay: autotrophic versus heterotrophic processes and relationships to carbon flux. Aquatic Microbial Ecology 15, 177–189
- Bronk, D.A., Lomas, M., Glibert, P.M., Schukert, K.J., Sanderson, M.P., 2000. Total dissolved nitrogen analysis: comparisons between the persulfate, UV and high temperature oxidation method. Marine Chemistry 69, 163–178.
- Carlson, C.A., Ducklow, H.W., 1995. Dissolved organic carbon in the upper ocean of the central equatorial Pacific Ocean, 1992: daily and fine scale vertical variations. Deep-Sea Research II 42, 639–656.
- Chester, R., 2003. Marine Geochemistry. Blackwell Science Ltd, Malden, MA 506 pp.
- Dagg, M.J., 1974. Loss of prey body contents during feeding by an aquatic predator. Ecology 55, 9903–9906.
- Diaz, F., Raimbault, P., 2000. Nitrogen regeneration and dissolved organic nitrogen release during spring in a NW Mediterranean coastal zone (Gulf of Lions): implications for the estimation of new production. Marine Ecology Progress Series 197, 51–65.
- Dugdale, R.C., Goering, J.J., 1967. Uptake of new and regenerated forms of nitrogen in primary productivity. Limnology and Oceanography 12, 196–206.
- Eppley, R.W., 1992. Chlorophyll, photosynthesis and new production in the Southern California Bight. Progress in Oceanography 30, 117–150.
- Eppley, R.W., Holm-Hansen, O., 1986. Primary production in the Southern California Bight. In: Eppley, R.W. (Ed.),

- Plankton Dynamics of the Southern California Bight. Springer, New York, pp. 176–215.
- Eppley, R.W., Peterson, B.J., 1979. Particulate organic matter flux and planktonic new production in the deep ocean. Nature 282, 677–680.
- Eppley, R.W., Renger, E.H., 1986. Nitrate-based primary production in nutrient-depleted surface waters off California. Oceanographic Topics 2, 229–268.
- Eppley, R.W., Renger, E.H., 1992. Nitrate utilization by plankton in the Equatorial Pacific March 1988 along 150°W. Journal of Geophysical Research 97, 663–668.
- Eppley, R.W., Renger, E.H., Harrison, W.G., 1979a. Nitrate and phytoplankton production in southern California coastal waters. Limnology and Oceanography 24, 483–494.
- Eppley, R.W., Renger, E.H., Harrison, W.G., Cullen, J.J., 1979b. Ammonium distribution in southern California coastal waters and its role in the growth of phytoplankton. Limnology and Oceanography 24, 495–509.
- Fiedler, R., Proksch, G., 1975. The determination of N15 by emission and mass spectrometry in biochemical analysis: a review. Analytica Chimica Acta 78, 1–62.
- Glibert, P.M., Lipschultz, F., McCarthy, J.J., Altabet, M.A., 1982. Isotope dilution models of uptake and remineralization of ammonium by marine plankton. Limnology and Oceanography 27, 639–650.
- Goldman, J.C., 1988. Spatial and temporal discontinuities of biological processes in pelagic surface waters. In: Rothschild, B.J. (Ed.), Toward a Theory on Biological-Physical Interactions in the World Ocean. Kluwer Academic Publishers, Dordrecht, pp. 273–296.
- Grasshoff, K., Ehrhardt, M., Kremling, K., 1983. Methods of Seawater Analysis. Verlag Chemie, New York, p. 419.
- Grasshoff, K., Kremling, K., Ehrhardt, M. (Eds.), 1999. Methods of Seawater Analysis. Wiley–VCH, Weinheim p. 599.
- Hansell, D.A., 2002. DOC in the global ocean carbon cycle. In: Hansell, D.A., Carlson, C.A. (Eds.), Biogeochemistry of Marine Dissolved Organic Matter. Academic Press, San Diego, pp. 685–726.
- Hasegawa, T., Koike, I., Mukai, H., 2000a. Dissolved organic nitrogen dynamics in coastal waters and the effect of copepods. Journal of Experimental Marine Biology and Ecology 244, 219–238.
- Hasegawa, T., Koike, I., Mukai, H., 2000b. Estimation of dissolved organic nitrogen release by micrograzers in natural plankton assemblages. Plankton Biology and Ecology 47, 23–30.
- Hasegawa, T., Koike, I., Mukai, H., 2001. Fate of food nitrogen in marine copepods. Marine Ecology Progress Series 210, 167–174.
- Hopkinson Jr., C.S., Fry, B., Nolin, A.L., 1997. Stoichiometry of dissolved organic matter dynamics on the continental shelf of the northeastern USA. Continental Shelf Research 17, 155–166.
- Horrigan, S.G., Hagström, Å., Koike, I., Azam, F., 1988. Inorganic nitrogen utilization by assemblages of marine bacteria in seawater culture. Marine Ecology Progress Series 50, 147–150.
- Jackson, G.A., Eldridge, P.M., 1992. Food web analysis of a planktonic system off Southern California. Progress in Oceanography 30, 223–251.
- Jones, M.N., 1984. Nitrate reduction by shaking with cadmium. Water Research 18, 643–646.

- Kirchman, D.L., 1994. The uptake of inorganic nutrients by heterotrophic bacteria. Microbial Ecology 28, 255–271.
- Kirchman, D.L., 2000. Uptake and regeneration of inorganic nutrients by marine heterotrophic bacteria. In: Kirchman, D.L. (Ed.), Microbial Ecology of the Oceans. Wiley, New York, pp. 261–288.
- Kirchman, D.L., Wheeler, P.A., 1998. Uptake of ammonium and nitrate by heterotrophic bacteria and phytoplankton in the sub-Arctic Pacific. Deep-Sea Research I 45, 347–365.
- Kirchman, D.L., Lancelot, C., Fasham, M., Legendre, L., Radach, G., Scott, M., 1993. Dissolved organic matter in biogeochemical models of the ocean. In: Towards a Model of Ocean Biogeochemical Processes. Springer, New York, pp. 209–225.
- Lampert, W., 1978. Release of dissolved organic carbon by grazing zooplankton. Limnology and Oceanography 23, 831–834.
- Laws, E., 1984. Isotope dilution models and the mystery of the vanishing N-15. Limnology and Oceanography 29, 379–386.
- Mullin, M.M., 1986. Spatial and temporal scales and patterns. In: Eppley, R.W. (Ed.), Plankton Dynamics of the Southern California Bight. Springer, New York, pp. 216–273.
- Nagata, T., 2000. Production mechanisms of dissolved organic matter. In: Kirchman, D.L. (Ed.), Microbial Ecology of the Oceans. Wiley, New York, pp. 121–152.
- Olson, R.J., 1981. ¹⁵N tracer studies of the primary nitrite maximum. Journal of Marine Research 39, 203–225.
- Parsons, T.R., Maita, Y., Lalli, C., 1984. A Manual of Chemical and Biological Methods for Seawater Analysis. Pergamon Press, Oxford 173 p.
- Slawyk, G., Raimbault, P., Gentilhomme, V., 1990. On the discrepancies between a colorimetric and isotopic method for measuring nitrate utilization in nutrient-depleted waters: implications for the design of experimental protocols in new production studies. Hydrobiologia 207, 333–339.
- Slawyk, G., Raimbault, P., Garcia, N., 1998. Measuring gross uptake of ¹⁵N-labeled nitrogen by marine phytoplankton without particulate matter collection: evidence for low ¹⁵N losses to the dissolved organic nitrogen pool. Limnology and Oceanography 43, 1734–1739.

- Small, L.F., Landry, M.R., Eppley, R.W., Azam, F., Carlucci, A.F., 1989. Role of plankton in the C and N budget of Santa Monica Basin, Calf. Marine Ecology Progress Series 56, 57–74.
- Toggweiler, J.R., 1989. Is the downward dissolved organic matter (DOM) flux important in carbon transport? In: Berger, W.H., Smetacek, V.S., Wefer, G. (Eds.), Productivity in the Ocean: Past and Present. Wiley, New York, pp. 65–83.
- Urban-Rich, J., 1999. Release of dissolved organic carbon from copepod fecal pellets in the Greenland Sea. Journal of Experimental Marine Biology and Ecology 232, 107–124.
- Varela, M.M., Barquero, S., Bode, A., Fernandez, E., Gonzalez, N., Teira, E., Varial, M., 2003. Microplanktonic regeneration of ammonium and dissolved organic nitrogen in the upwelling area of the NW of Spain: relationships with dissolved organic carbon production and phytoplankton size structure. Journal Plankton Research 25, 719–736.
- Venrick, E.L., 1998. Spring in the California Current: the distribution of phytoplankton species, April 1993 and April 1995. Marine Ecology Progress Series 167, 73–88.
- Vidal, M., Duarte, C.M., Agusti, S., 1999. Dissolved organic nitrogen and phosphorus pools and fluxes in the central Atlantic Ocean. Limnology and Oceanography 44, 106–115.
- Ward, B.B., 1987. Nitrogen transformations in the Southern California Bight. Deep-Sea Research I 34, 785–805.
- Ward, B.B., Bronk, D.A., 2001. Net nitrogen uptake and DON release in surface waters: importance of trophic interactions implied from size-fractionation experiments. Marine Ecology Progress Series 219, 11–24.
- Ward, B.B., Kilpatrick, K.A., Renger, E.H., Eppley, R.W., 1989.Biological nitrogen cycling in the nitracline. Limnology and Oceanography 34, 493–513.
- Wheeler, P.A., Kirchman, D.L., 1986. Utilization of inorganic and organic nitrogen by bacteria in marine systems. Limnology and Oceanography 31, 998–1009.
- Williams, P.M., 1986. Chemistry of the dissolved and particulate phases in the water column. In: Eppley, R.W. (Ed.), Plankton dynamics of the Southern California Bight. Springer, New York, pp. 53–83.