

DON sources: methods and processes

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Outline:

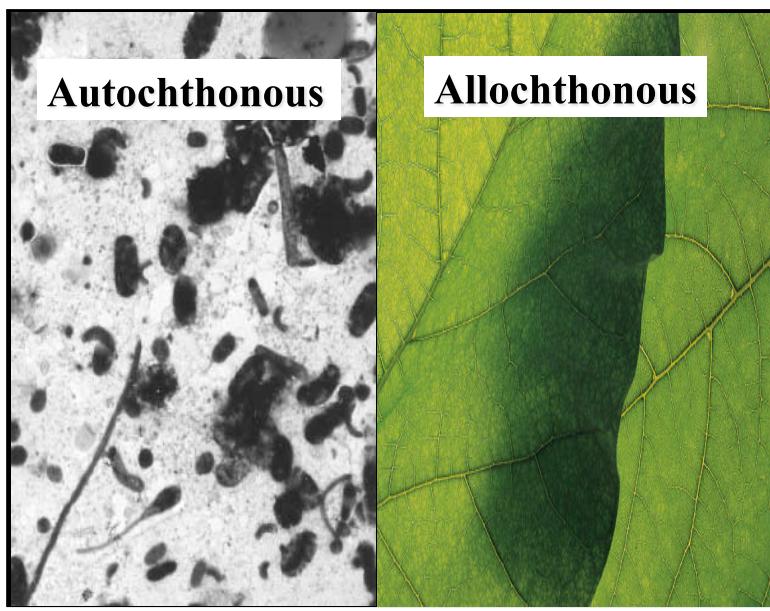
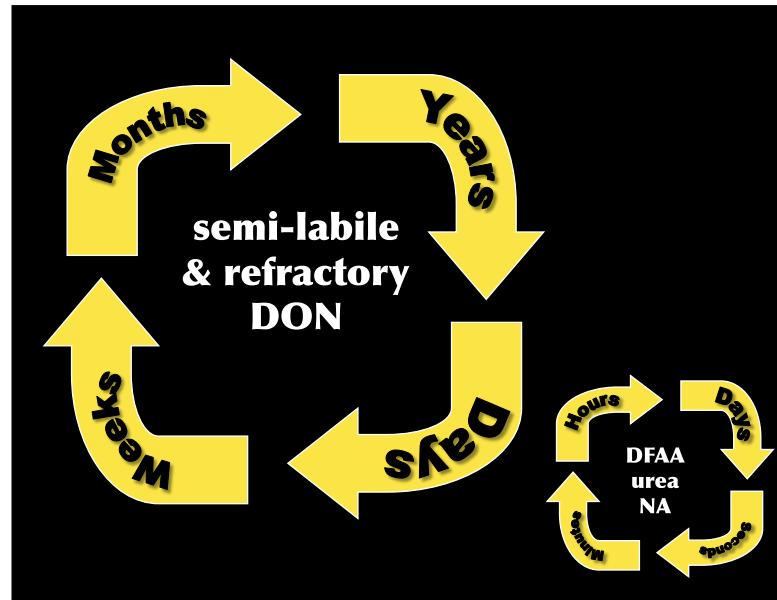
- What is DON?
- Methods to study release
- Autochthonous sources
- Allochthonous sources
- DON as a mode of N delivery

What is DON?

| Labile | Semi-labile | Refractory |
|---------------|---|--------------|
| DFAA | proteins | humic acids |
| urea | DCAA | fulvic acids |
| nucleic acids | amino | porphorins |
| methylamines | polysaccharides (chitins & peptidoglycans) | amides |



DON



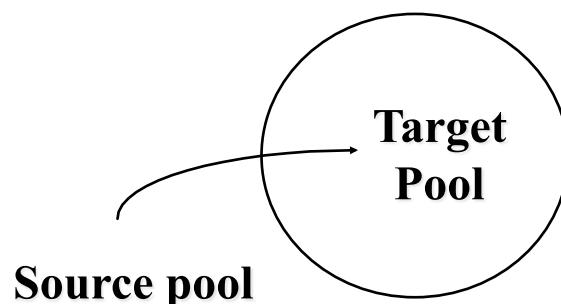
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| Autochthonous | | Allochthonous |

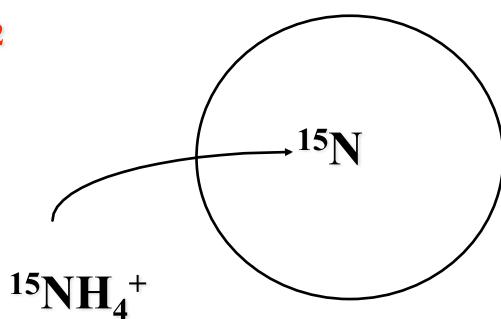
Methods for studying release:

- 1. Bioassays**
 - 2. Radioactive tracers**
 - 3. Stable isotope tracers**
 - a. Direct measures**
 - b. Isotope dilution**

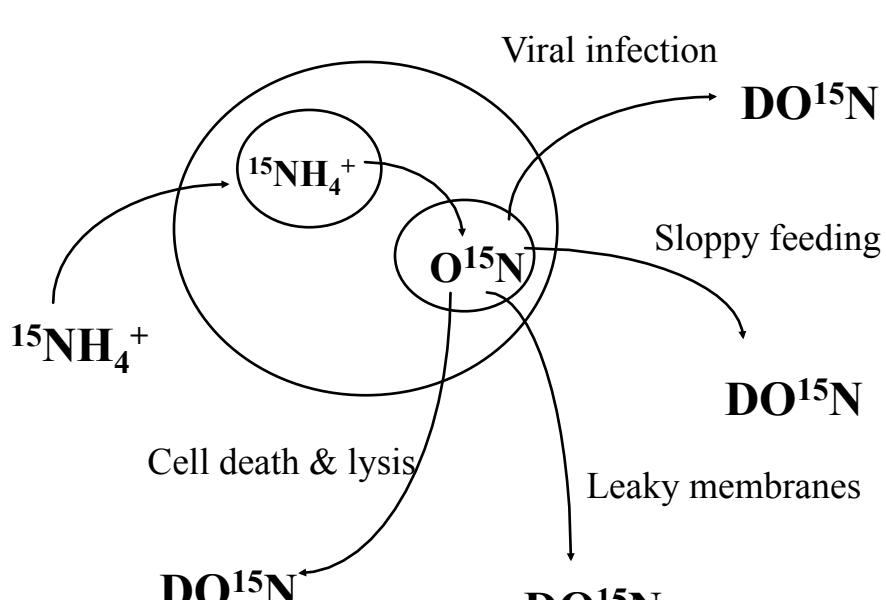
| Bioassays & Radiotracers $(^3\text{H}, ^{14}\text{C}, ^{32}\text{P})$ | Stable isotope tracers $(^{15}\text{N}, ^{13}\text{C})$ |
|---|---|
| Absolute amounts ~Net rates | Ratios $(^{15}\text{N}/^{14}\text{N}$ or $^{13}\text{C}/^{12}\text{C}$) |
| | Gross and net rates Uptake and regeneration simultaneously |



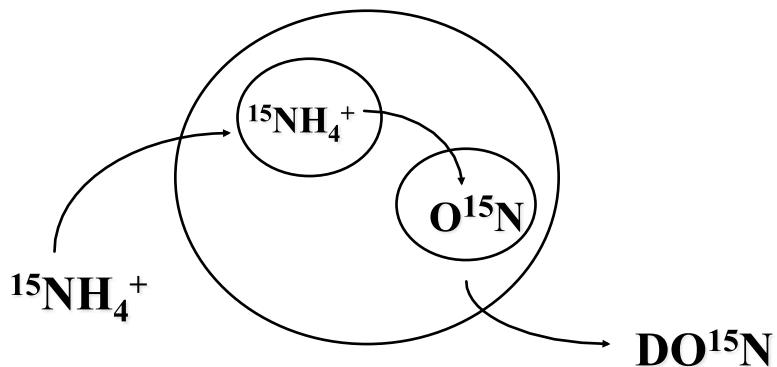
$$\text{Rate} = \frac{\text{atom \% of target}}{\text{atom \%} \times \text{Time}} \times [\text{target}]$$



$$\frac{\text{Net atom\% PN}}{\text{Rate atom \% } \text{NH}_4^+ \times \text{Time}} \times [\text{PN}]$$



Bronk 1999 JPR

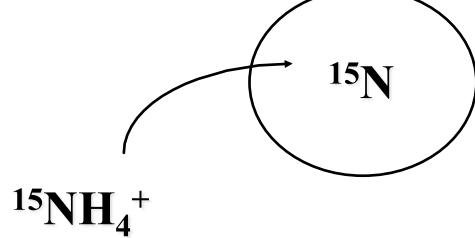


$$\text{Uptake} = \frac{\text{Gross Rate}}{\text{atom \% NH}_4^+ \times \text{Time}} \times [\text{PN}]$$

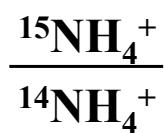
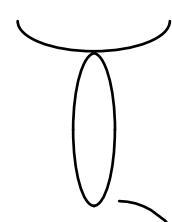
^{^{15}N in PN & DON}

Bronk et al. 1994 Science

$$\text{Gross Uptake} - \text{Net Uptake} = \text{DON Release}$$



$$\frac{\text{Net Uptake}}{\text{Rate}} = \frac{\text{atom \% } ^{15}\text{N in PN}}{\text{atom \% NH}_4^+ \times \text{Time}} \times [\text{PN}]$$



Isotope dilution

$$\frac{\text{Net Uptake}}{\text{Rate}} = \frac{\text{atom \% } ^{15}\text{N in PN}}{\text{atom \% NH}_4^+ \times \text{Time}} \times [\text{PN}]$$

$$P_t = P_0 + (d - u)t$$

$$\ln(R_t - R_a) = \ln(R_0 - R_a) - [d/(d - u)][\ln P_t/P_0]$$

P_t and P_0 = ambient NH_4^+ conc at end and start of incubation

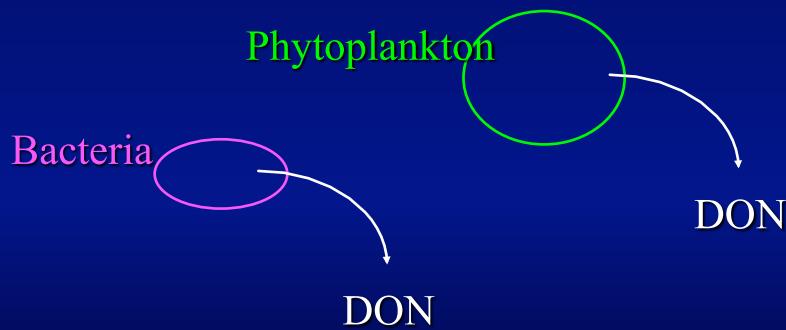
R_t and R_0 = atom % of the NH_4^+ pool at end and start of incub.

u = absolute uptake rate

d = regeneration rate

Glibert et al. 1982 L&O

Autochthonous sources of DON



- Direct release - Passive diffusion
- Cell death and lysis (autolysis)
- Bacterial exoenzyme release

Excretion of organic matter by phytoplankton: Do healthy cells do it?

Sharp 1977 L&O

Active release
outflow model

Fogg 1966 OMBAR

Passive
diffusion model

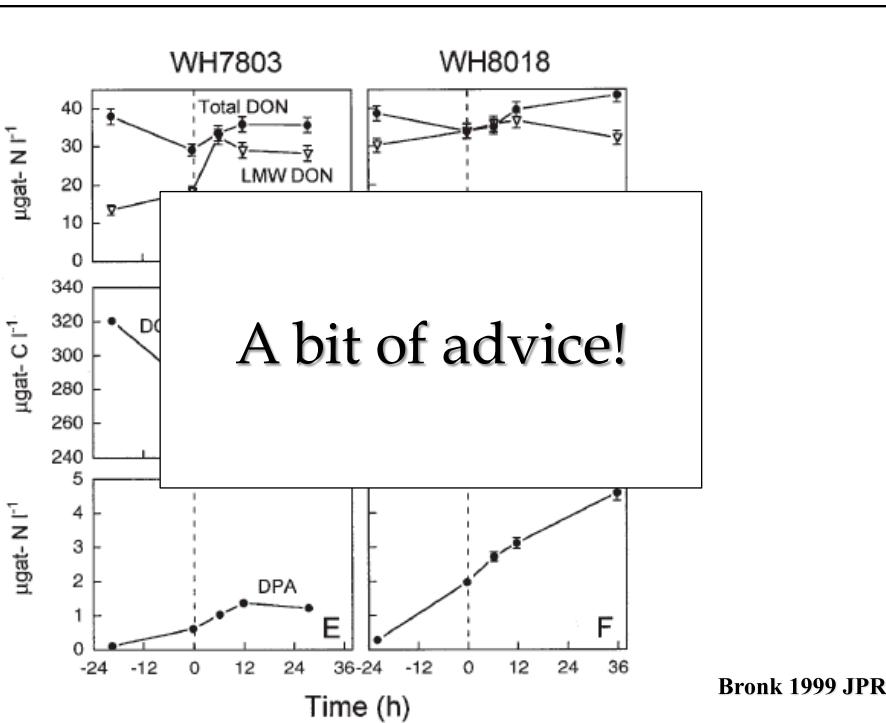
Fogg 1966 OMBAR

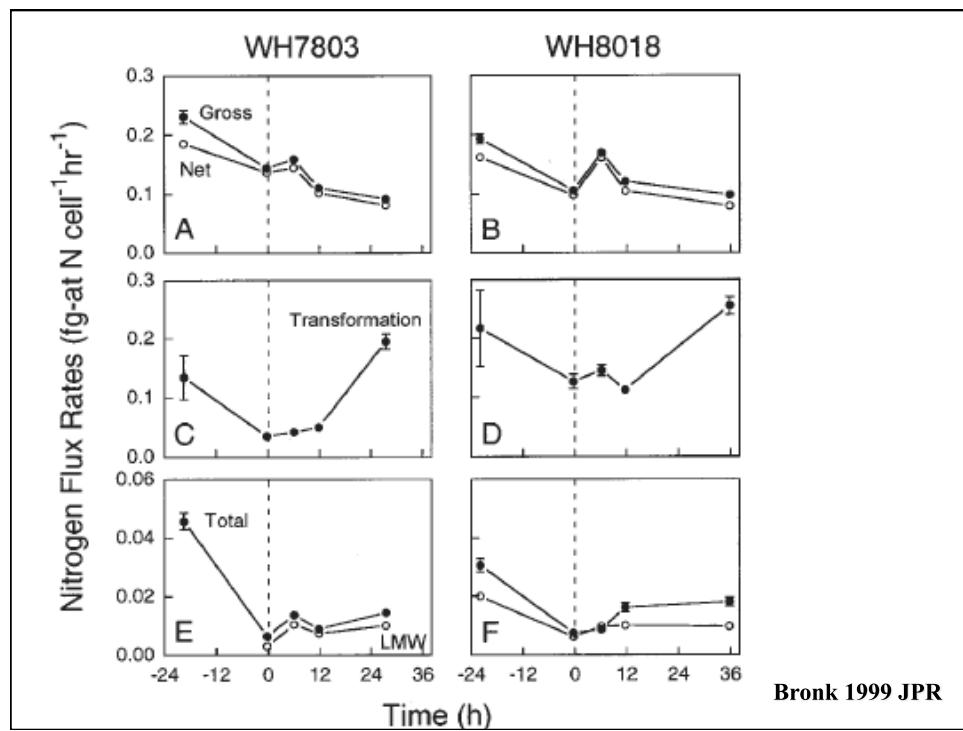
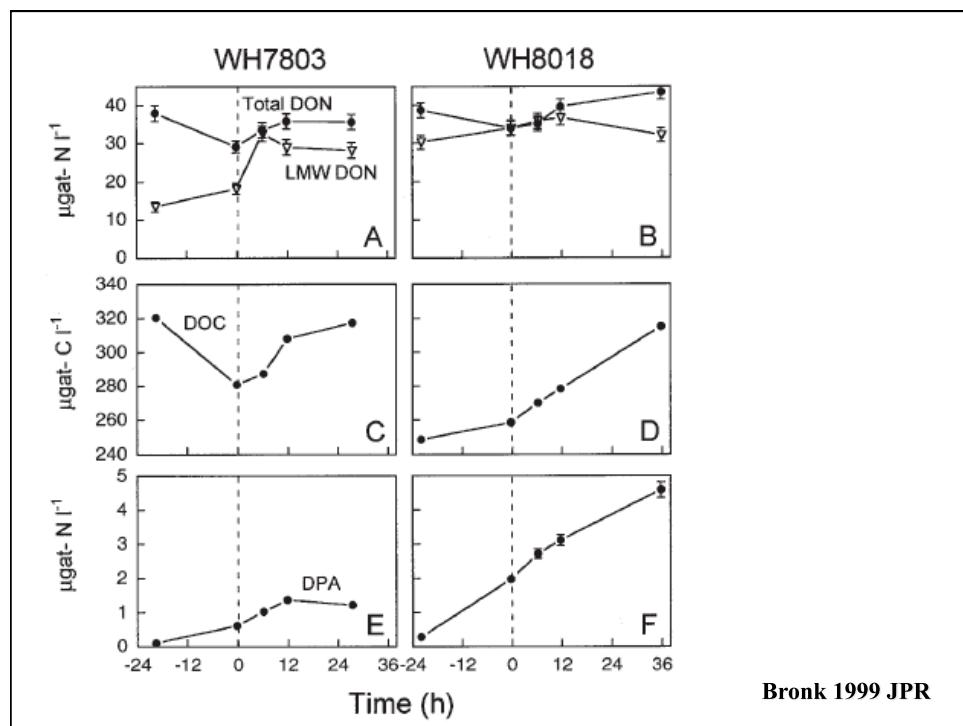
Reduction of
viruses

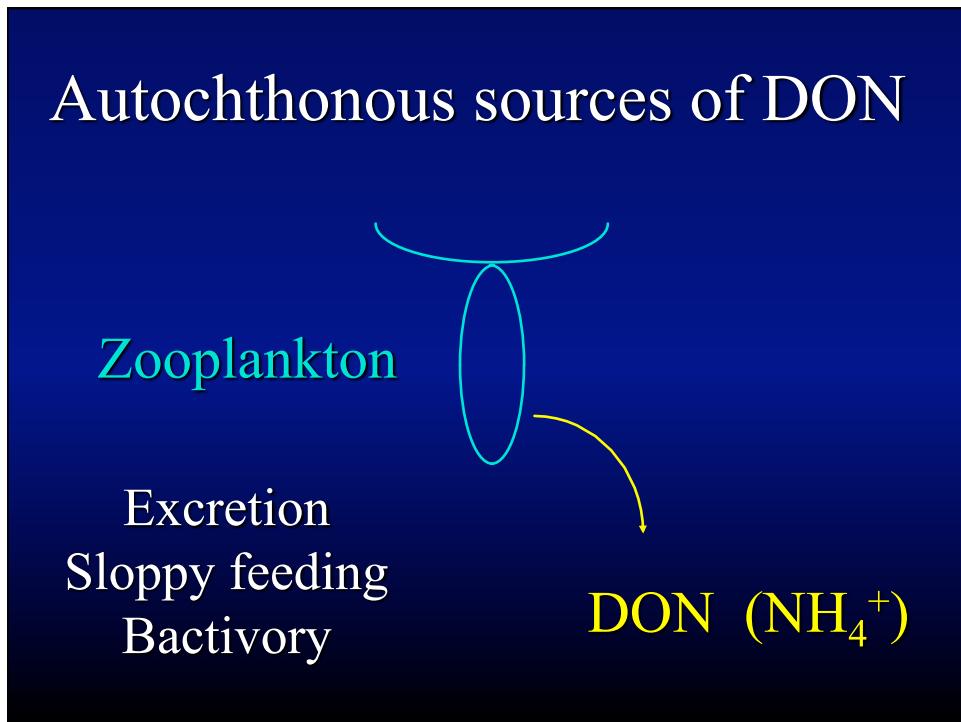
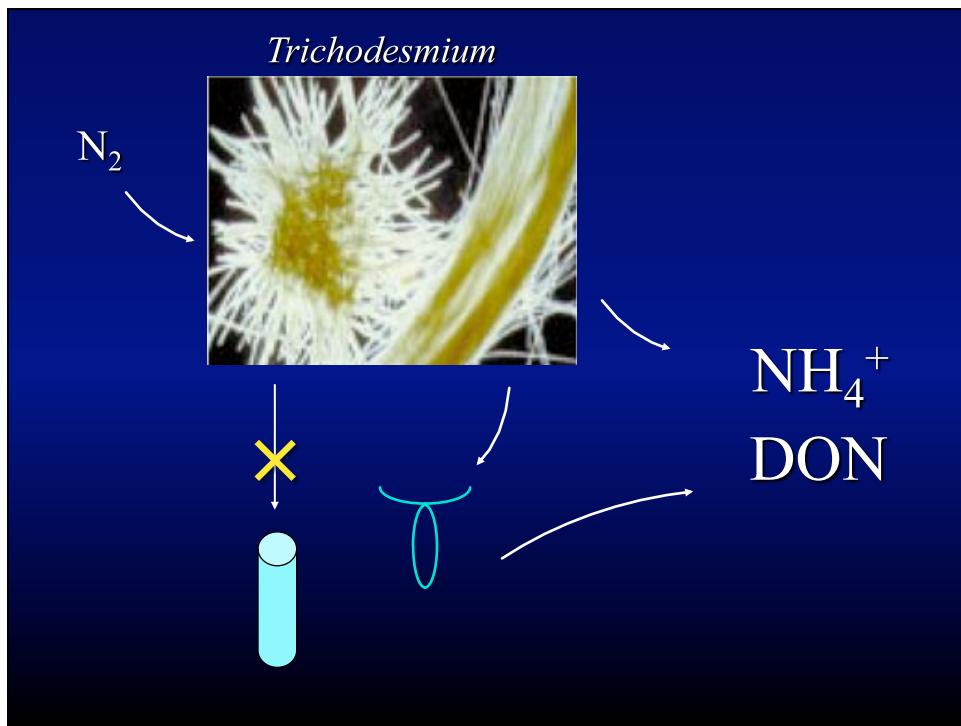
Murray 1995 JPR

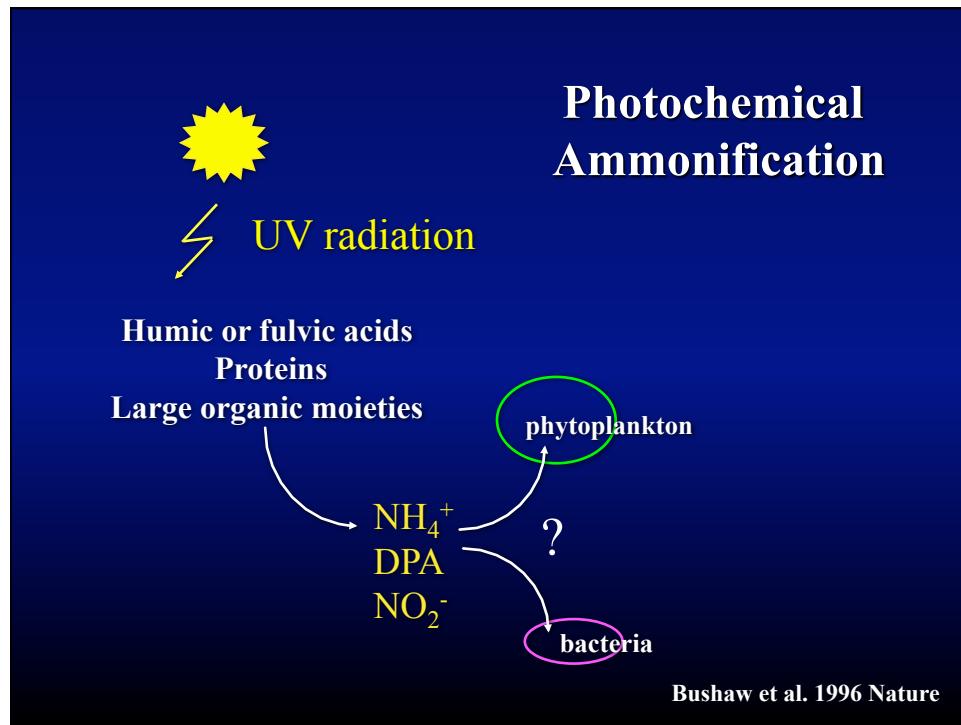
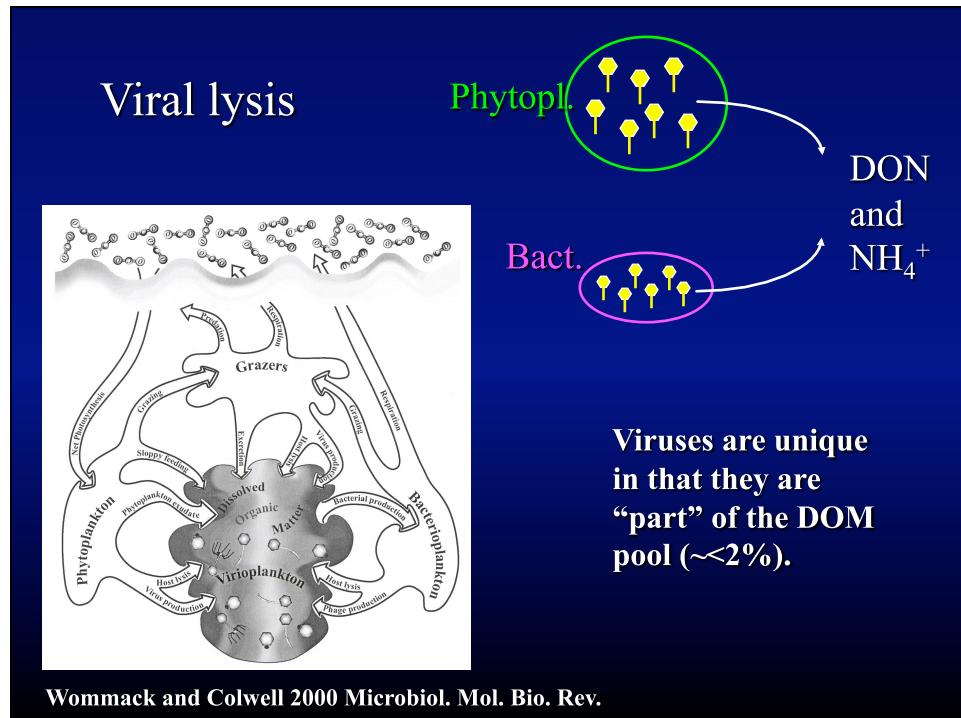
Bacteria as
ectoparasites

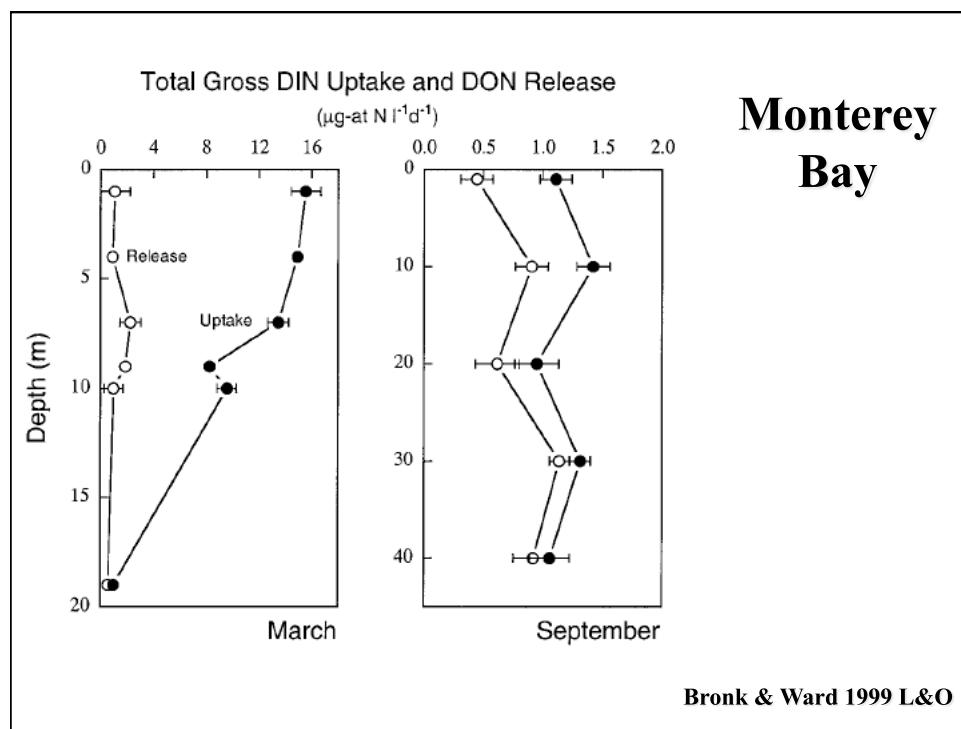
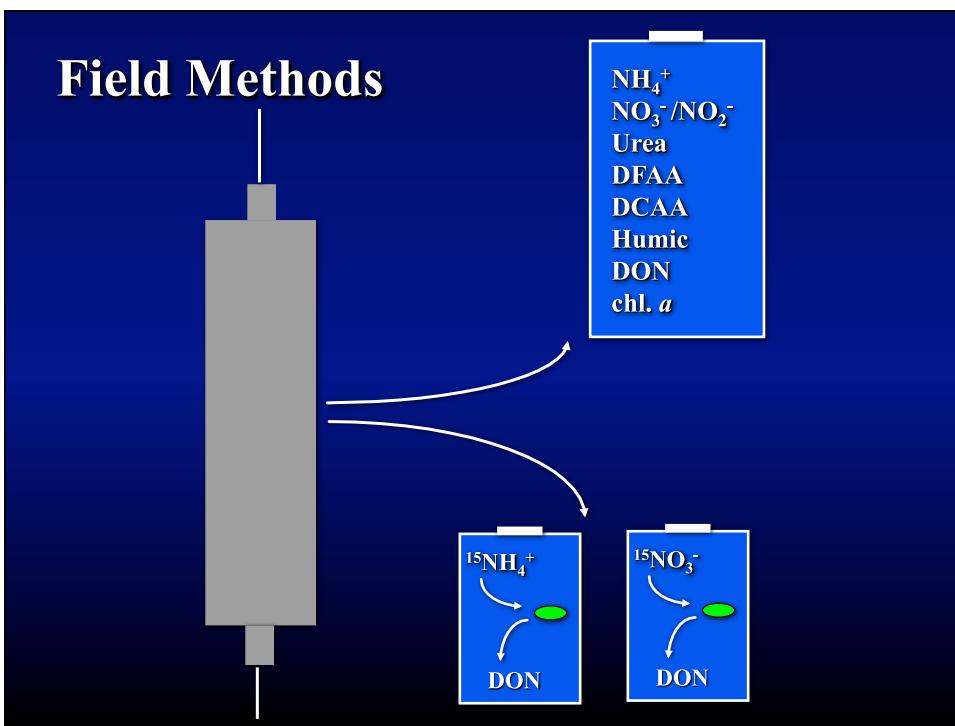
Bjørnsen 1988 L&O





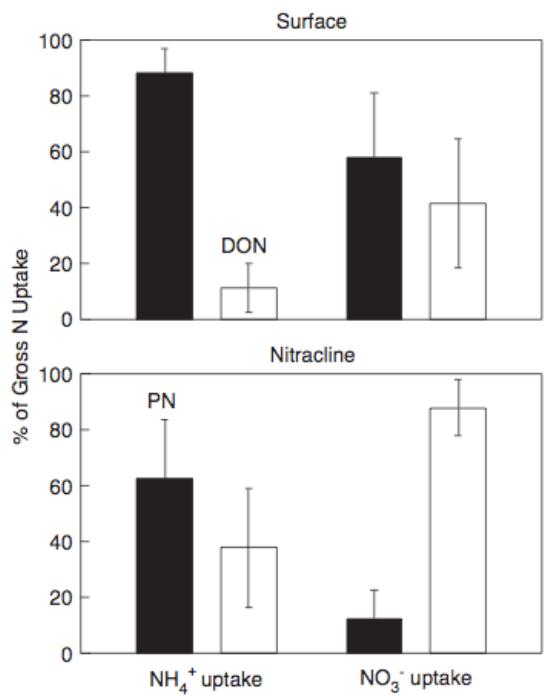




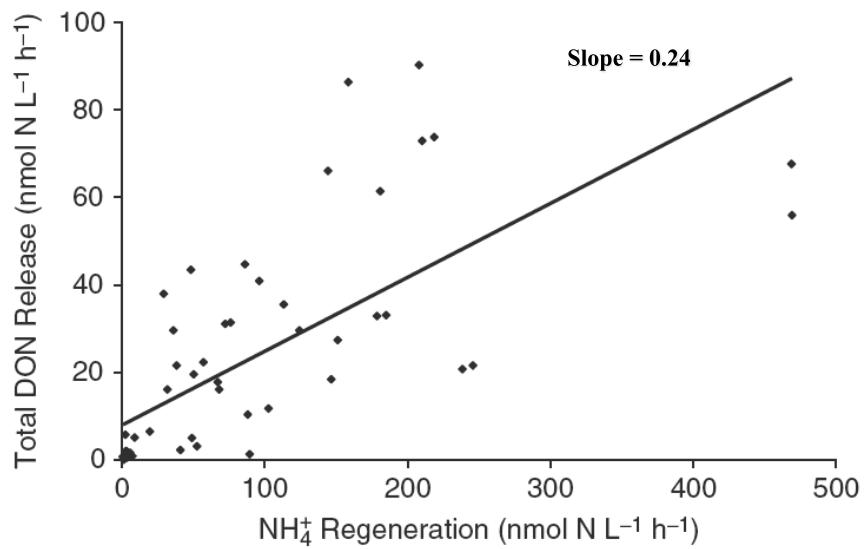


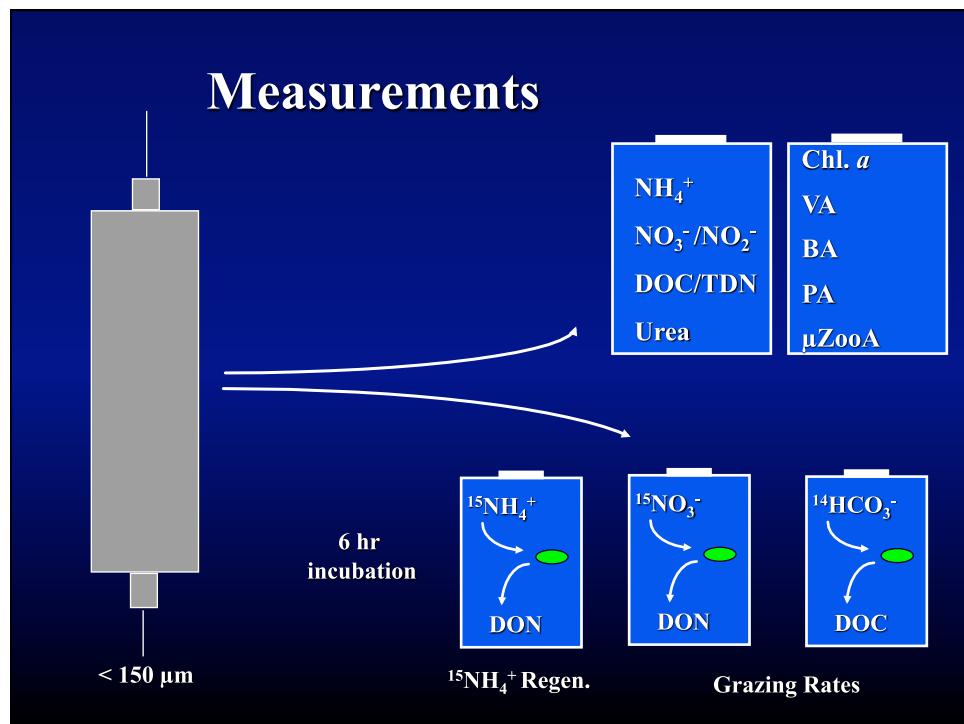
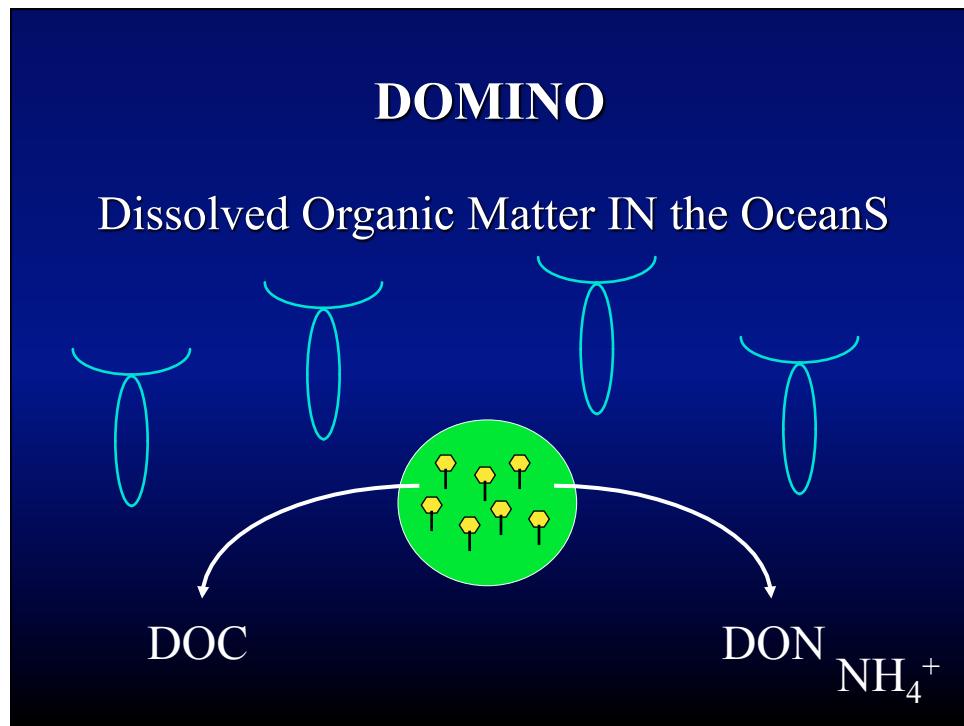
Southern California Bight

Bronk & Ward 2005 DSRI



Bronk & Steinberg 2008 N Bible





Ambient Conditions



$\text{NH}_4^+ \sim 9 \mu\text{M}$

$\text{NO}_3^- \sim 10 \mu\text{M}$

$\text{DOC} \sim 185 \mu\text{M}$

$\text{DON} \sim 10 \mu\text{M}$ (C:N 18)

urea $\sim 0.6 \mu\text{M}$

Treatments



Control



Acartia tonsa

+ Grazers



0 Virus



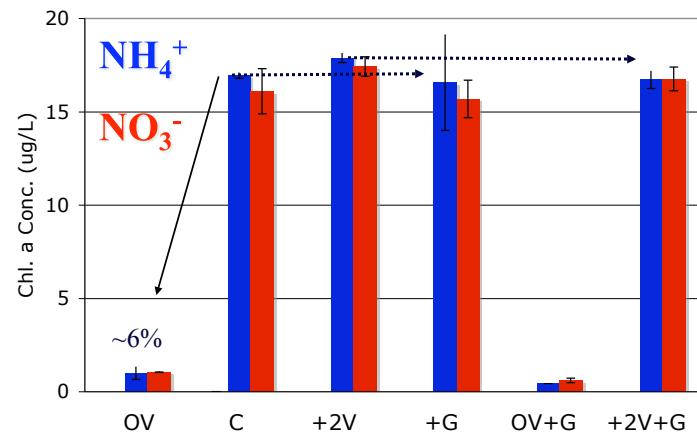
0 Virus + Grazers



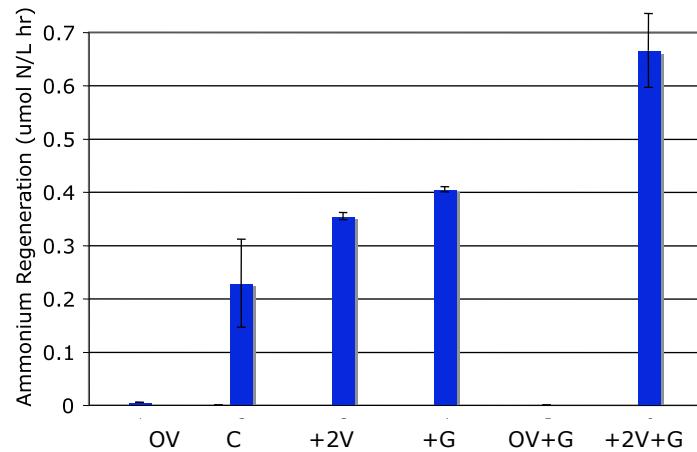
2X Virus



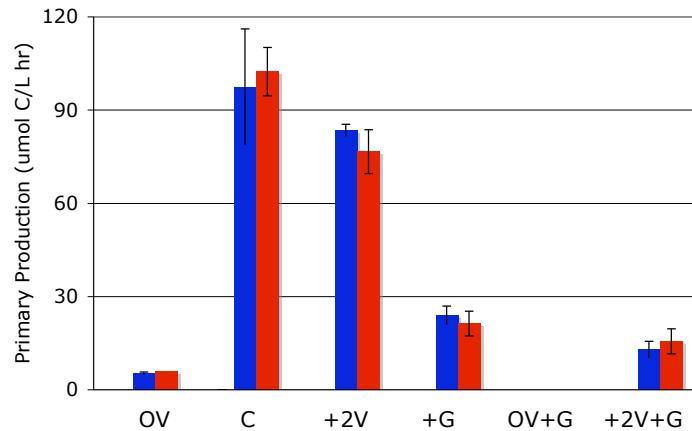
2X Virus + Grazers



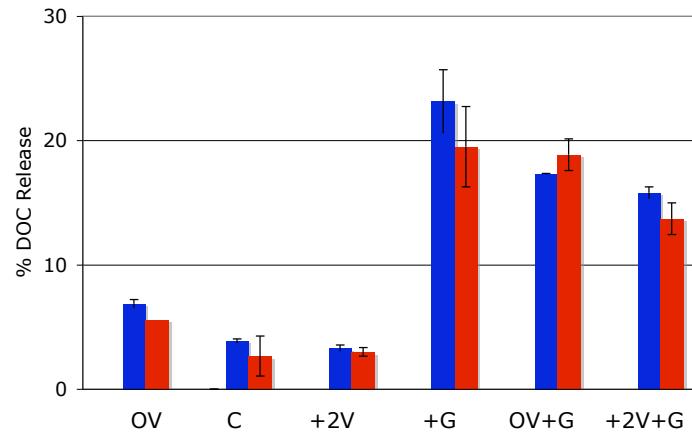
- Slight decrease in Chl. a when grazers are added.



- Grazers and viruses increased the rate of NH_4^+ regeneration.
- Additive effect.



- Both grazers and viruses depressed primary production.
 - Additive effect.

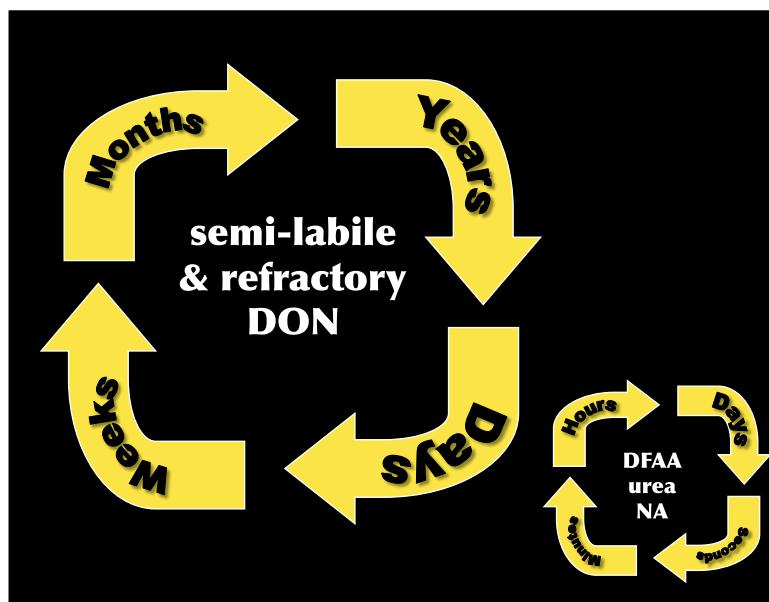


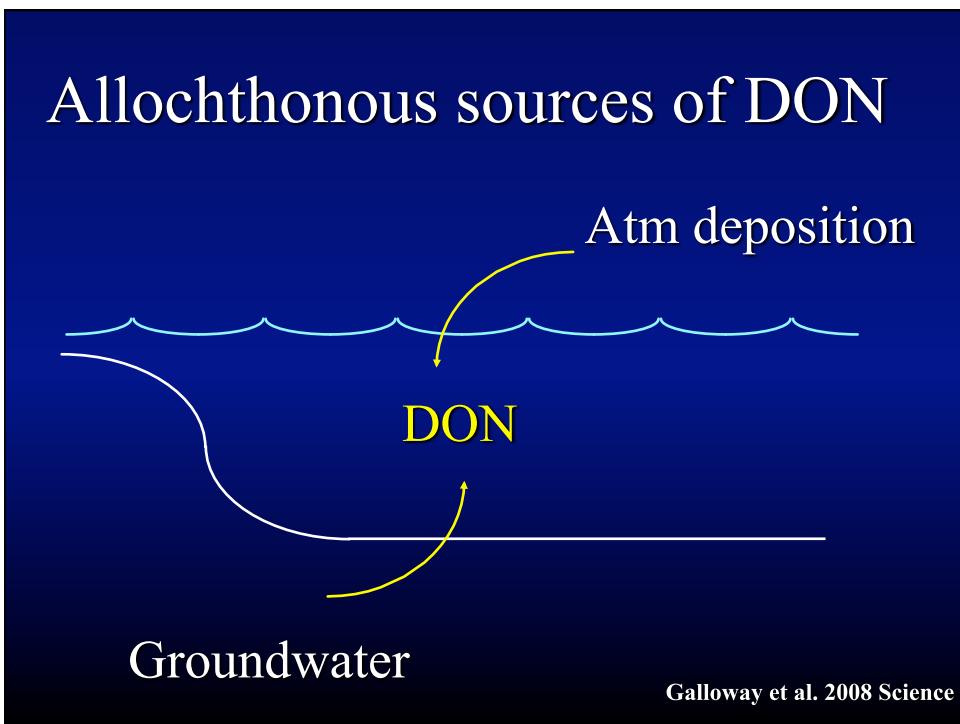
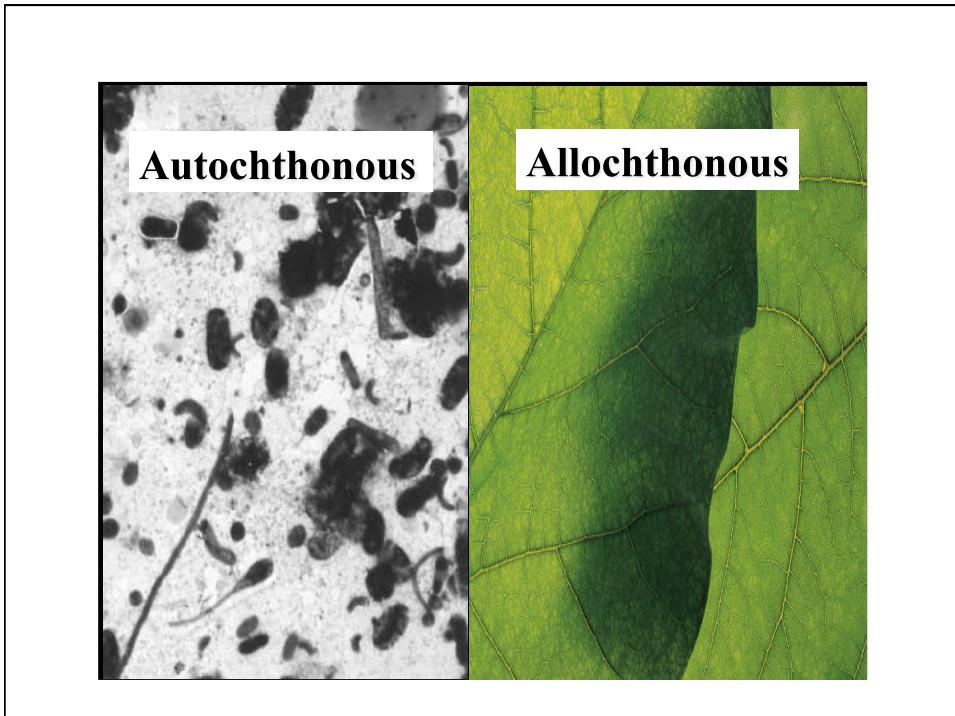
- Grazers increased the % of primary production released as DOC.
 - Viruses tended to decrease it.

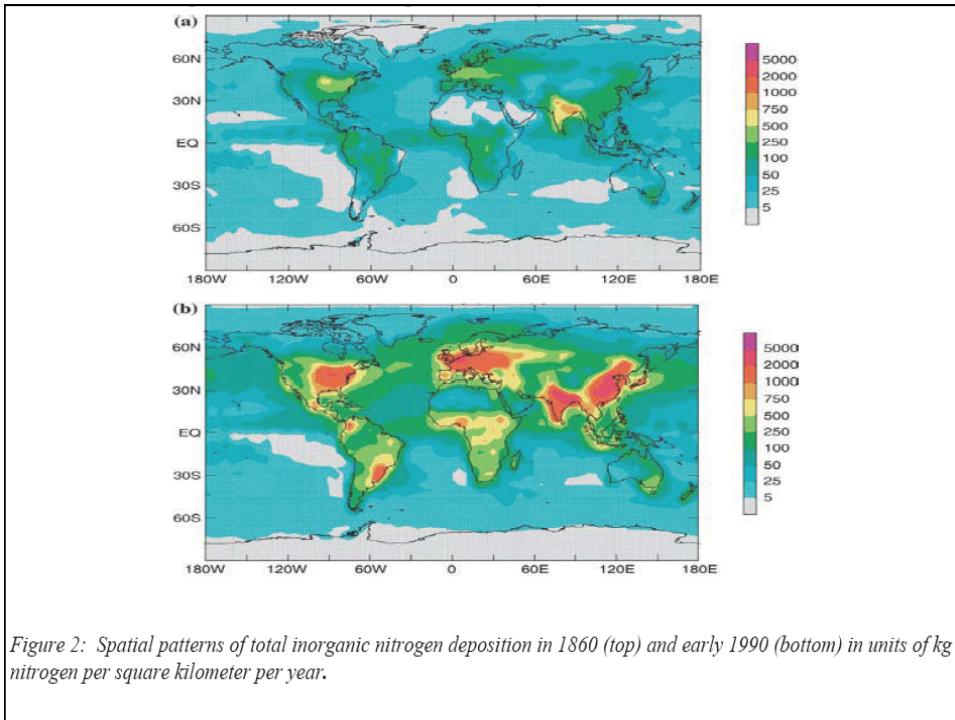
Turnover time = conc/rate

| Location | Compound Considered | Turnover Time | Units |
|-------------------------------------|-------------------------------|--------------------------|-------|
| Oceanic | | | |
| Northeastern Pacific | DON | 0.91 | years |
| Equatorial Atlantic (15N-25N) | DON | 0.4 to 13.2 ^c | years |
| Equatorial Atlantic (15S-15N) | DON | 12.7 ± 26.1 ^c | years |
| Equatorial Atlantic (35S-15S) | DON | 2.1 to 300 ^c | years |
| Caribbean Sea | DON | 40.7 ± 10.4 | days |
| Southern California Bight where? | DON | 11 to 62 | days |
| | HMW DON >1kD | ~238 ^a | days |
| Northern Sargasso Sea | Protein | 0.38 to 3.42 | days |
| Northern Sargasso Sea | Modified protein ^d | 9.04 to 32.71 | days |
| Northern Sargasso Sea | Modified protein ^d | 9.04 to 32.71 | days |
| Northern Sargasso Sea | DFAA | 0.03 to 0.29 | days |
| Central Arctic | DFAA | ~2.72 | days |

Bronk 2002 Book chapter







DON in atmospheric deposition

Refractory??

| Source/Location | % org N | Reference |
|---------------------|---------|-------------------------------------|
| Walker Branch, TN | 34 | Kelly and Meagher 1986 |
| Coastal plain, FL | 40–63 | Rickert 1983 |
| Cascade Mtns., OR | 46–72 | Fredriksen 1976 |
| Coastal plain, SC | 49 | Richter et al. 1983 |
| Philadelphia, PA* | 19–52 | This study |
| Chesapeake Bay | 57 | Smullen et al. 1982 |
| | | |
| Rhode River, MD | 18–44 | Jordan et al. 1995 |
| New Brunswick, NJ*† | 2–44 | Seitzinger and Sanders unpubl. data |
| Narragansett, RI | 19 | Nixon et al. 1995 |
| U.K. | 21 | Cornell et al. 1995 |
| Czech Rep. | 27 | " |
| N. Carolina | 21 | " |
| Amazonia | 22 | " |
| Recife, Brazil | 25 | " |
| Bermuda | 59 | " |
| Tahiti | 84 | " |
| NE Atlantic | 62 | " |
| NE Atlantic | 67 | " |
| Cape Cod, MA | 43 | Valielia et al. 1997 |
| Lewes, DE | 23 | Scudlark et al. 1998 |

Seitzinger & Sanders 1999 L&O

Allochthonous sources of DON

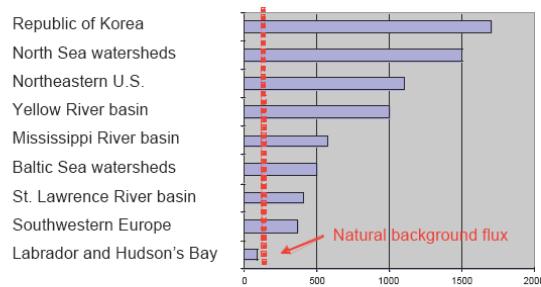
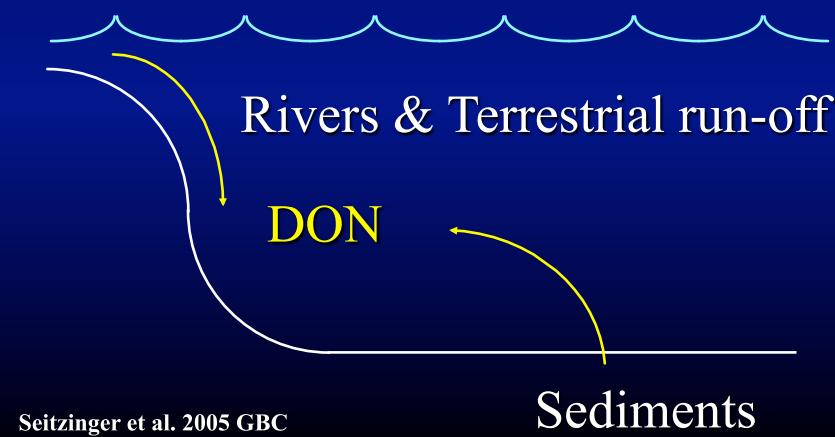


Figure 3: Flux of reactive nitrogen from the landscape to coastal oceans in rivers for key contrasting regions of the world in the temperate zone, in units of kg nitrogen per square kilometer of watershed area per year.

DON in rivers

Refractory??

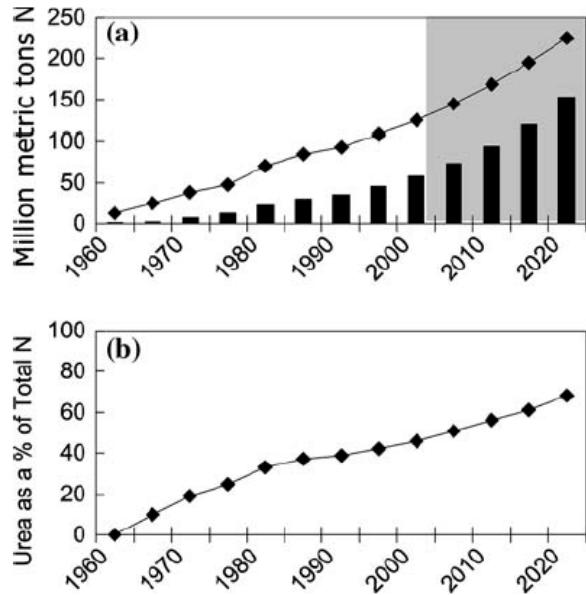
Table 5. Percent composition of inorganic and organic N in various rivers

| River | % DIN | % DON | % PN | % ON | Reference |
|---------------------------------|-----------------|-------|------|-----------------|----------------------------|
| Mullica, NJ, USA | 30 | 61 | 9 | 70 | Durand (1988) |
| Mississippi, USA | 53 ^a | | | 43 | Turner & Rabalais (1991) |
| Amazon | 25 | 29 | 46 | 75 | Richey et al. (1991) |
| Mackenzie, Canada | 10 | 15 | 75 | 90 | Telang et al. (1991) |
| Unnamed stream, NH, USA | 39 | 51 | 10 | 61 | Likens (1985) |
| Blackstone, RI, USA | 73 | 23 | 4 | 27 | Nixon et al. (1995) |
| Pocono Mountain Stream, PA, USA | 4 | | | 96 | Seitzinger (unpubl. data) |
| Delaware, PA, USA | 74 | 20 | 5 | 25 | Culberson et al. (1987) |
| Rhode, MD, USA | 30 | 14 | 56 | 70 | Peterjohn & Correll (1984) |
| Como, CO, USA | 10 | 90 | | 90 ^b | Meybeck (1982) |
| Lindaret, France | 78 | 22 | | 22 ^b | Meybeck (1982) |
| Brevon, France | 77 | 23 | | 23 ^b | Meybeck (1982) |
| Danube, Rumania | 55 | 45 | | 45 ^b | Meybeck (1982) |
| Aare, Switzerland | 73 | 27 | | 27 ^b | Meybeck (1982) |
| Reuss, Switzerland | 56 | 44 | | 44 ^b | Meybeck (1982) |
| Rhine, Switzerland | 74 | 26 | | 26 ^b | Meybeck (1982) |
| Rhône, Switzerland | 56 | 44 | | 44 ^b | Meybeck (1982) |
| Ticino, Switzerland | 57 | 43 | | 43 ^b | Meybeck (1982) |
| Missouri, USA | 46 | 54 | | 54 ^b | Meybeck (1982) |
| Windrush, England | 60 | | | 40 | Heathwaite (1993) |
| Eastern Note, Poland | 29 | | | 71 | Taylor et al. (1986) |
| Wda, Poland | 70 | | | 30 | Taylor et al. (1986) |
| Kullarna, Sweden | 25 | | | 75 | Lepistö et al. (1995) |
| Dånersta, Sweden | 22 | | | 78 | Lepistö et al. (1995) |
| Myllypuro, Finland | 6 | | | 94 | Lepistö et al. (1995) |
| Pahkajoja, Finland | 21 | | | 79 | Lepistö et al. (1995) |
| Avg world rivers | 33 | | | 67 | Meybeck (1982) |

^aNO₃+NO₂ only. ^bMinimum estimates as based on DON only; no PN data

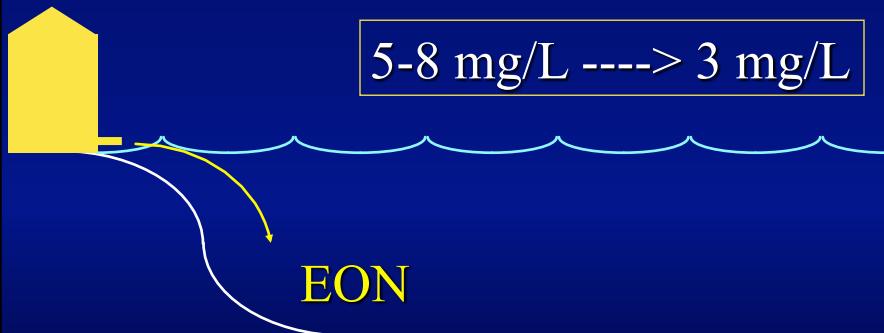
Seitzinger & Sanders 1997 MEPS

The rise of urea



Glibert et al. 2006 Biogeosciences

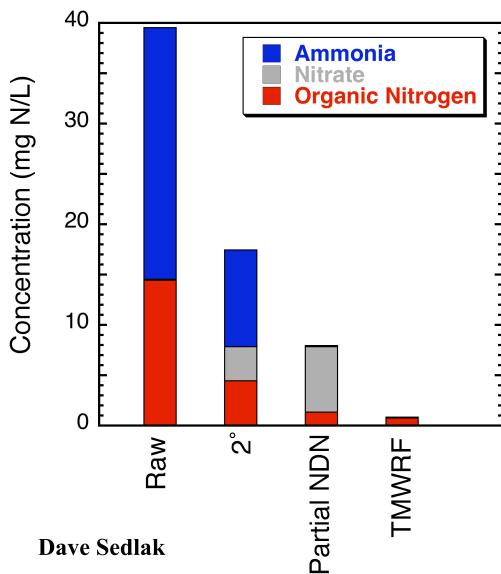
Allochthonous sources of DON



Effluent Organic Nitrogen

Mulholland et al. 2007 STAC

Nitrogen in Wastewater



Compounds poorly removed by treatment
Humics in source
Recalcitrant organics

Compounds formed during treatment

Composition of EON

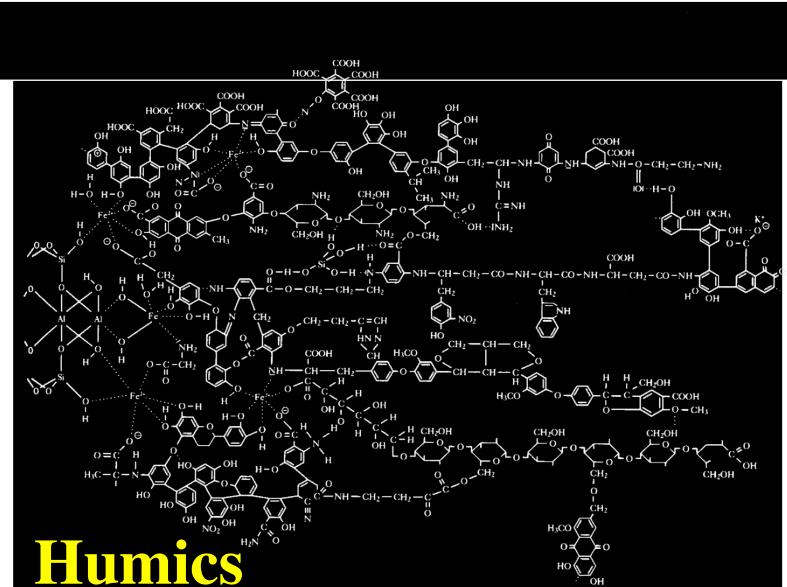
TABLE 1. Total Amino Acid Concentrations in the Secondary Treated Wastewater Effluents (Scully et al., 1988b; Confer et al., 1995; Grohmann et al., 1998; Pehlivanoglu and Sedlak, in preparation)

| | Wagott | Parkin | Elsässer | Hejzlar | Scully ^a | Scully | Confer | Pehlivanoglu |
|-----------|--------|--------|----------|---------|---------------------|--------|--------|--------------|
| mg N/L | 0.017 | 0.025 | 0.013 | 0.034 | 0.042–0.084 | 0.02 | 0.26 | 0.14–0.17 |
| μM N | 1.23 | 1.79 | 0.93 | 2.45 | 3–6 | 1.43 | 18.76 | 10–12 |

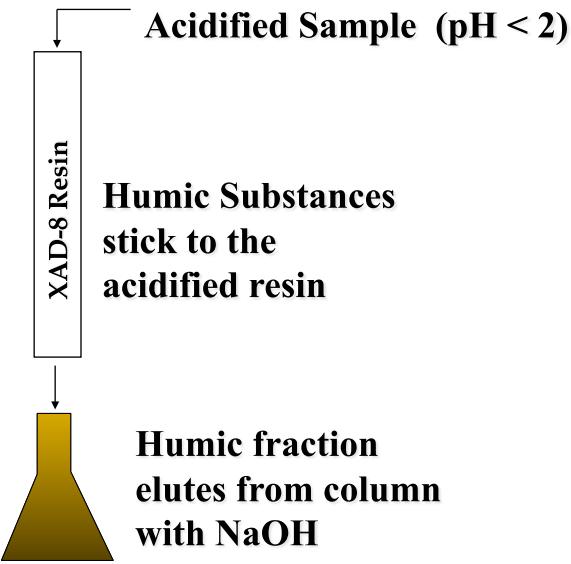
Note. Wagott, Parkin, Elsässer, Hejzlar, and Scully data are obtained from Grohmann et al. (1998).

^aPrimary effluent.

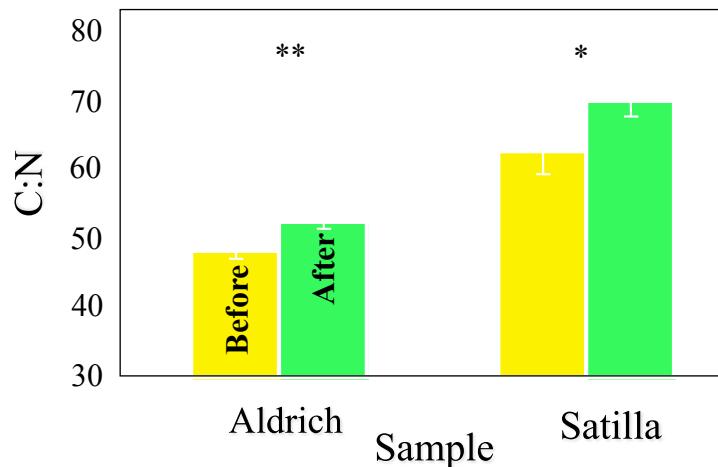
**Organics as a mode of
inorganic N delivery!**



Humic Extraction Method

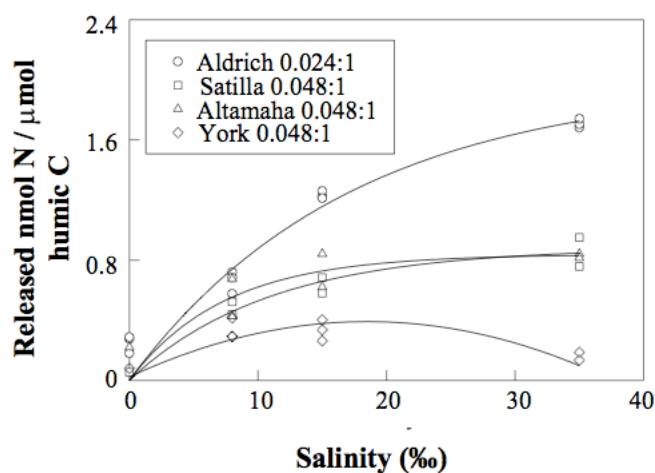


C:N Ratio of Saturated Humics Before and After XAD Extraction



Samples Saturated at 4 $\mu\text{mol NH}_4^+$
(mg humic-C) $^{-1}$

See & Bronk 2005 Mar Chem



See 2003 Dissertation

When humics hit ~15 %o they
dump NH_4^+



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