Phosphorus cycling in the sea: new stories and enduring questions

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Please note this has been edited to remove unpublished examples, feel free to contact me with any questions.

The take home message

- The basics:
 - P is important
 - It is rapidly cycled
 - It comes in organic forms that are poorly characterized
- Tech advances
 - Magic tells us that the inorganic form is at very low levels
 - Distribution of bond classes in high molecular weight DOP
- Adaptation to low P is common
 - Frequency and expression of pstS
 - The emerging importance of polyphosphate
 - Losing your phospholipids
 - Metabolism of phosphonates
- Enduring mysteries
 - Low molecular weight DOP composition
 - P machinery in the agal viruses
 - Microbial metabolism of reduced P

Why phosphorus?

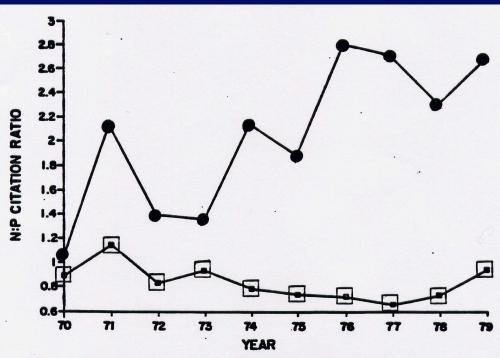


Fig. 16. Ratio of nitrogen citations to phosphorus citations in freshwater (\square) and brackish and marine waters (\bullet) from 1970 to 1979, based on citations in the Bibliographic Retrieval Services Biosis Data Base.

In 1970, equal numbers of N and P citations. By 1980, 4-fold increase in N versus P

The importance of P broadly appreciated

articles

The relative influer in the Ultraoligotrophic

Phosphorus Deficiency in the Atlantic: An Emerging Paradigm

in Oceanography

By James W. Ammerman, Raleigh R. Hood Darin A. Case, and James B. Cotner

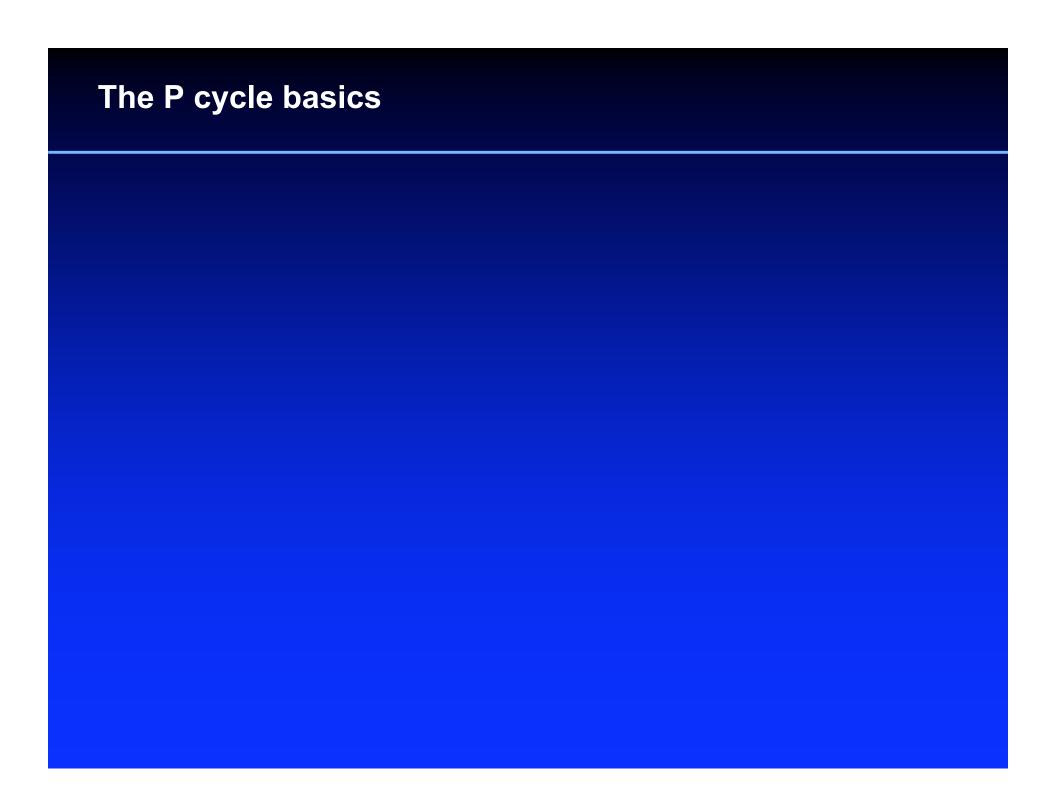
Eos, Vol. 84, No. 18, 6 May 2003

news and views

Aquatic ecology

Phosphorus, the staff of life

David M. Karl



Major P fluxes

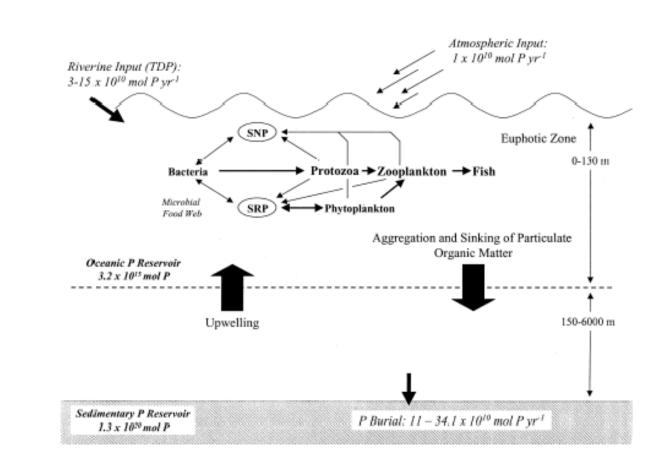
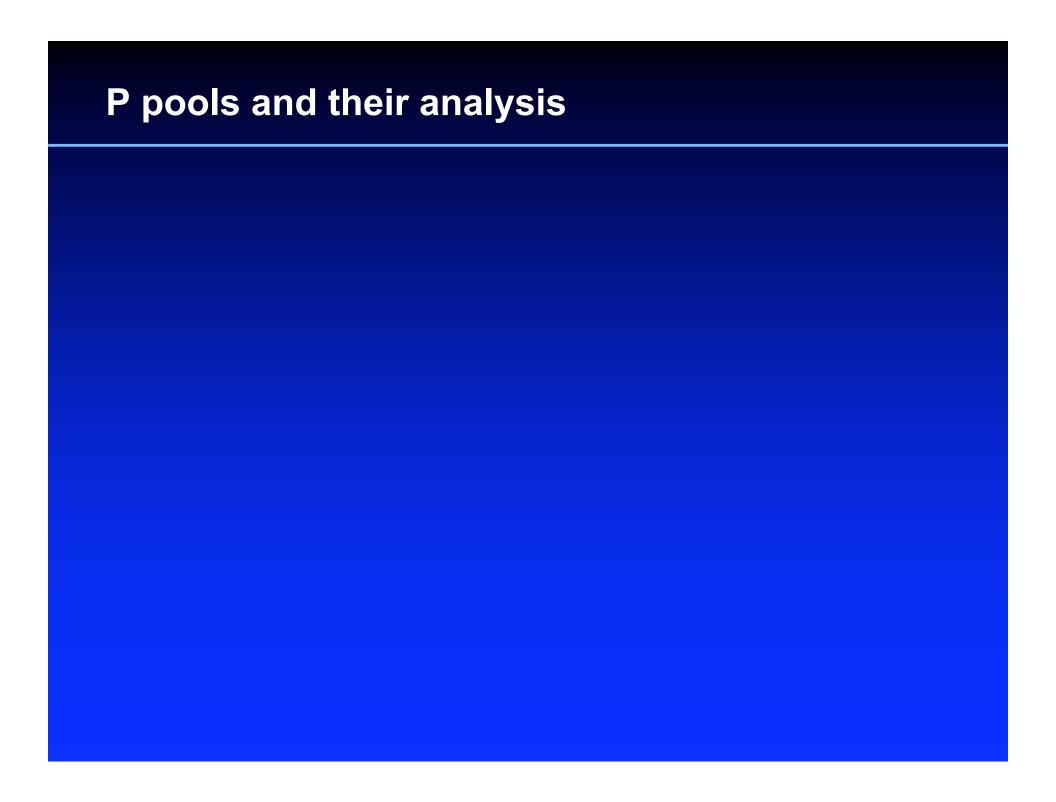


Fig. 3. The pre-anthropogenic marine P cycle. See text for details on fluxes.

P turnover - can be fast

Table 3	
SRP and SNP	turnover rates

P turnover rates	Coastal	Open ocean	References
SRP	< 1 h–10 days (> 1000 days in Bedford Basin)	Week to several months	Pomeroy, 1960; Duerden, 1973; Taft et al., 1975; Harrison et al., 1977; Perry and Eppley 1981; Smith et al., 1985; Sorokin, 1985; Harrison and Harris, 1986; Björkman and Karl, 1994; Björkman et al., 1999; Benitez-Nelson and Buesseler, 1999a
Total SNP	3->90 days	50-300 days	Jackson and Williams, 1985; Orrett and Karl, 1987; Lal and Lee, 1988; Lee et al., 1992; Karl and Yanagi, 1997; Björkman et al., 1999; Benitez-Nelson and Buesseler, 1999a
Bioavailable SNP (model compounds)	2-30 days	1-4 days	Ammerman and Azam, 1985; Nawrocki and Karl, 1989; Björkman and Karl, 1994; Björkman and Karl, 1999
Microplankton (< 1 μm)	> 1-3 days	NA	Benitez-Nelson and Buesseler, 1999a
Phytoplankton (> 1 μm)	< 1-8 days	< 1 week	Waser et al., 1996; Benitez-Nelson and Buesseler, 1999a
Macrozooplankton (> 280 μm)	14–40 days	30-80 days	Lal and Lee, 1988; Lee et al.,1991, 1992; Waser et al., 1996; Benitez-Nelson and Buesseler, 1999a



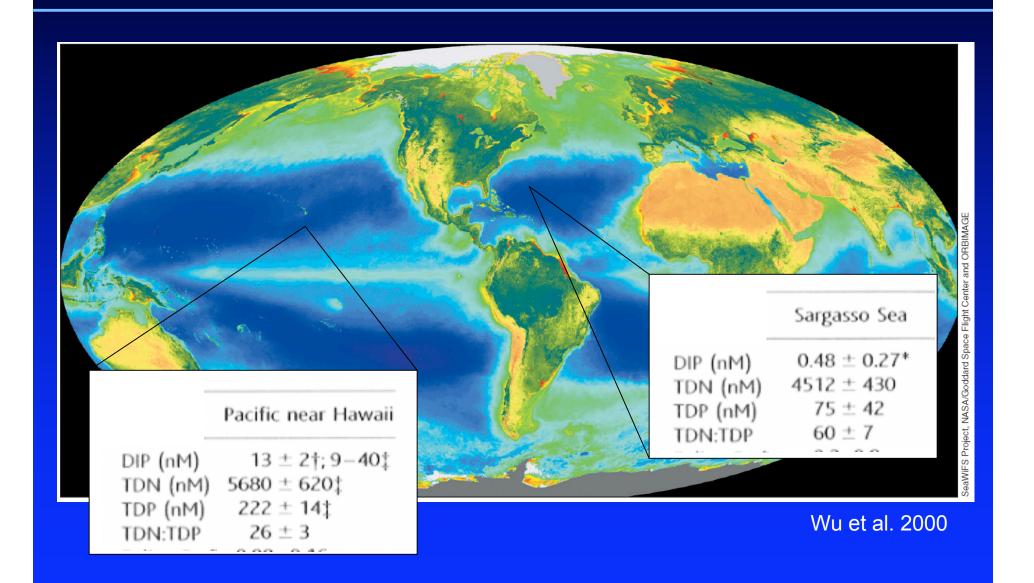
Time series observations pave the way

VOLUME 43, No. 2-3, 1996 ISSN 0967-0645 DEEP-SEA RESEARCH PART II Editor: John D. Milliman TOPICAL STUDIES IN **OCEANOGRAPHY** OCEAN TIME-SERIES: RESULTS FROM THE HAWAII AND BERMUDA RESEARCH PROGRAMS Guest Editors: D. M. Karl and A. F. Michaels Hawaii Bermuda PERGAMON

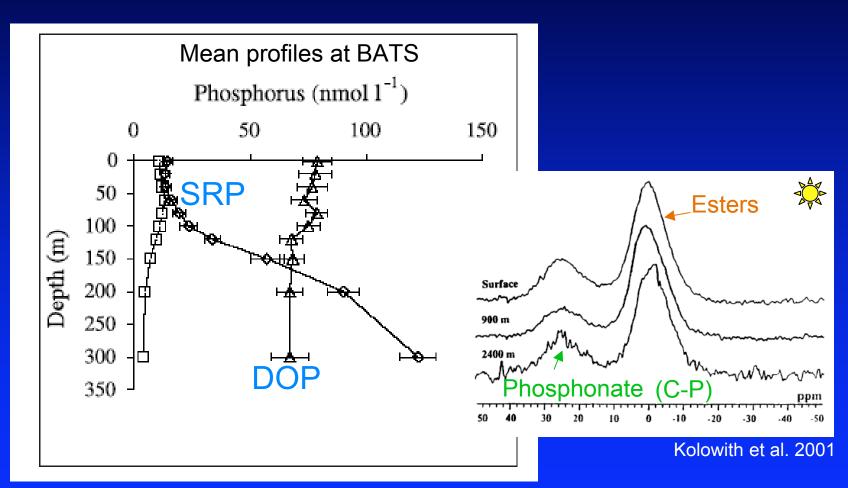
BATS

HOT

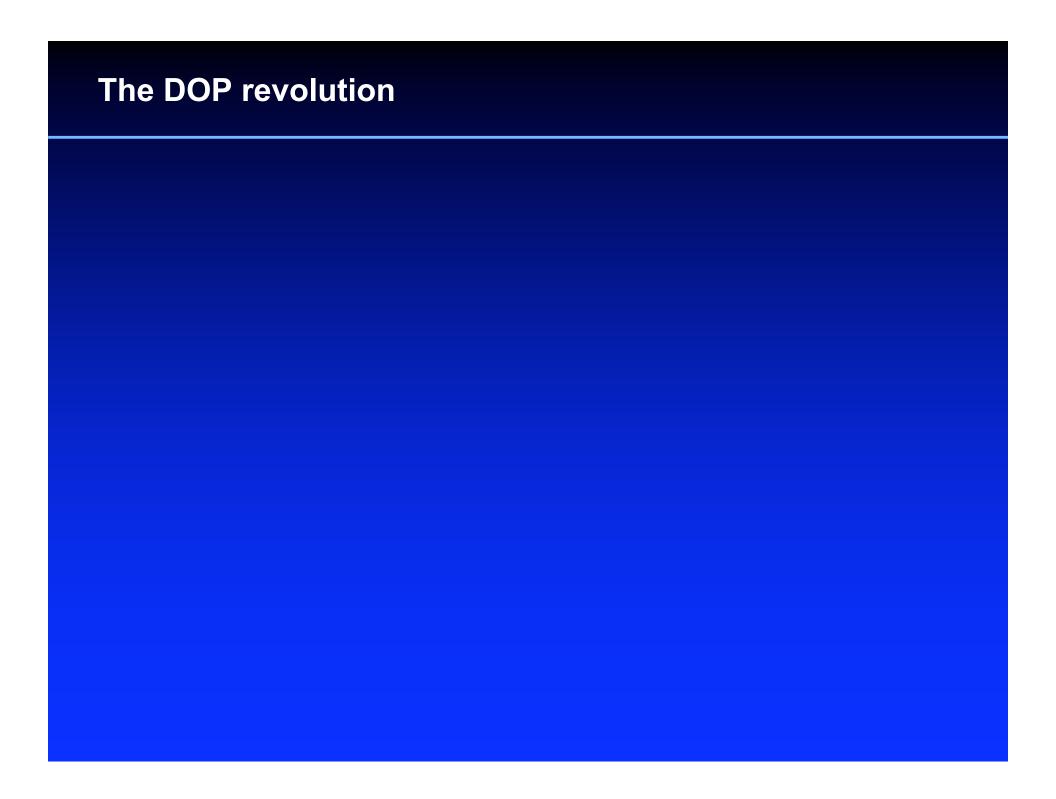
Low P in many ocean biomes



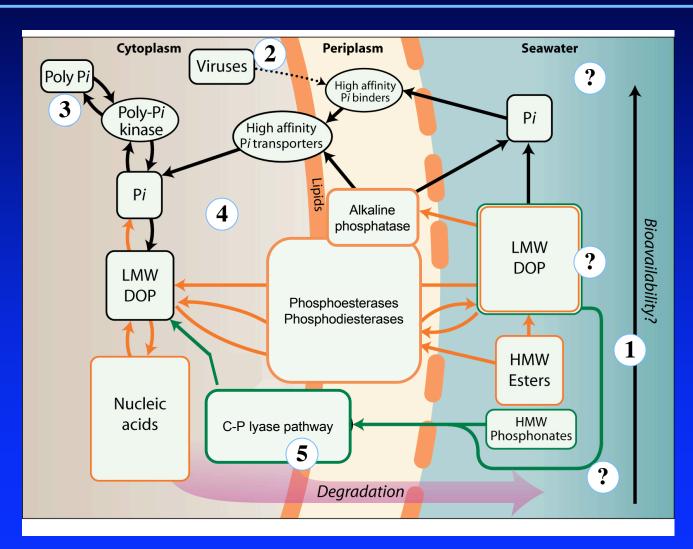
Phosphorus profiles in the Sargasso Sea



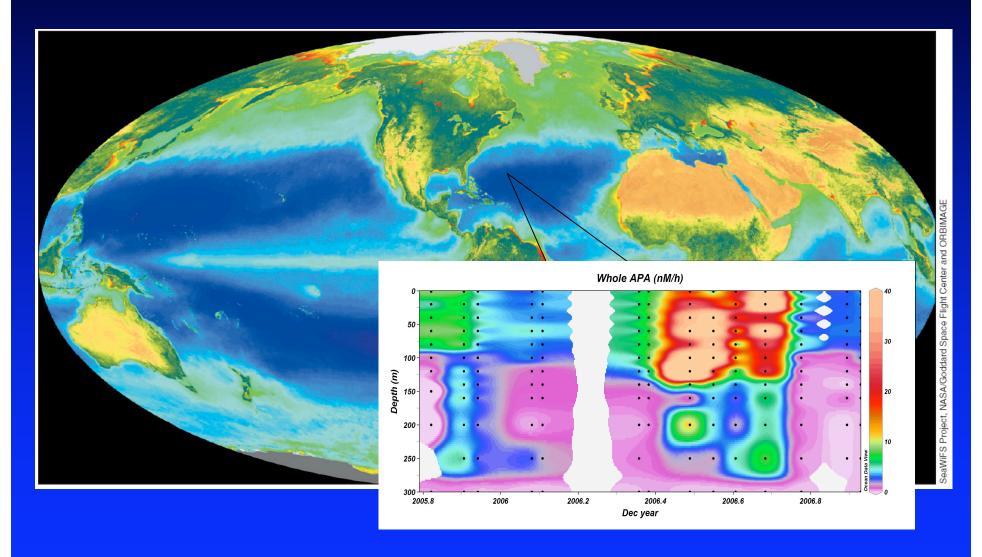
Ammerman et al. 2003



A conceptual diagram of marine microbial interactions with phosphorus

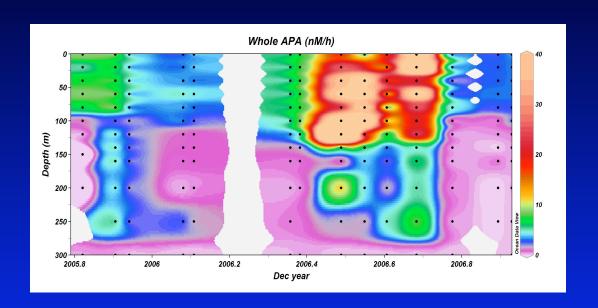


Alkaline phosphatase in the Sargasso Sea

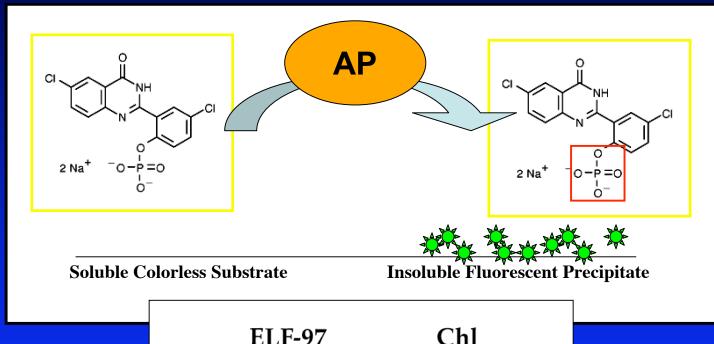


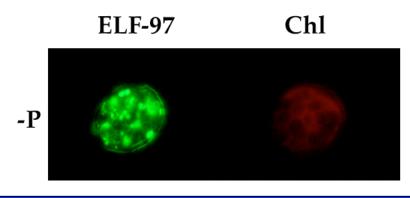
Does DOP support primary production?

Seasonal DOP inventory declines at BATS



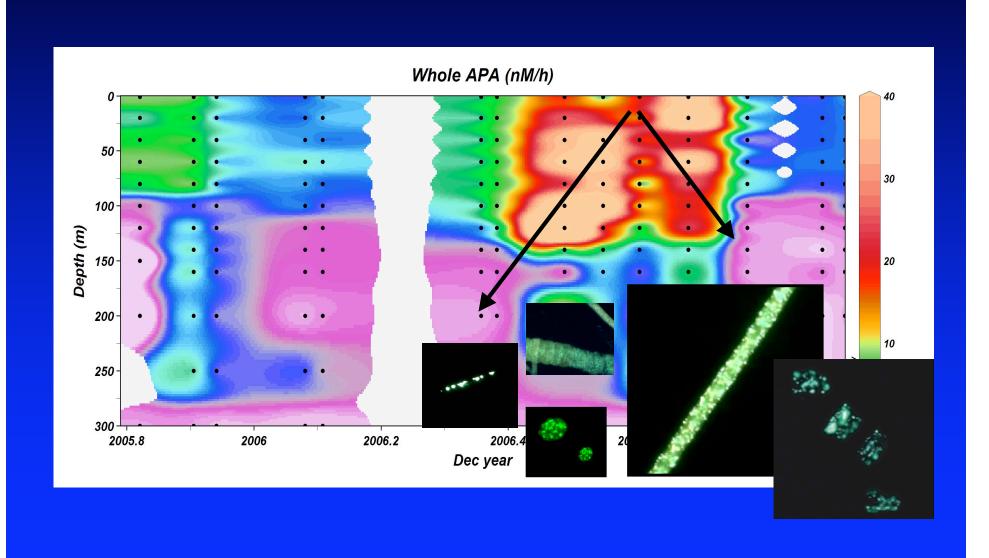
Enzyme labeled fluorescence

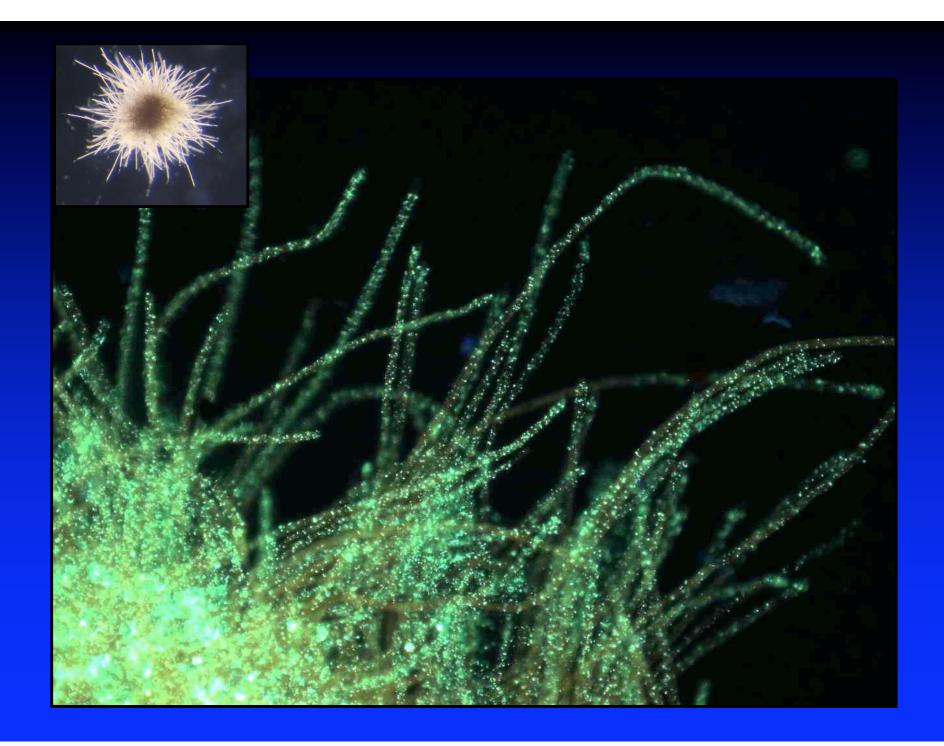




Dyhrman and Palenik 1999

Many microbes contributing to APA



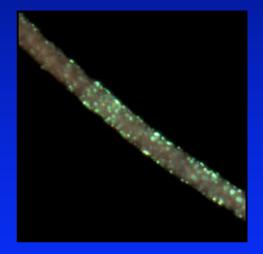


Seasonal APA in the Sargasso Sea

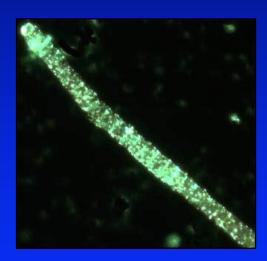
June DIP: 10.2 nM



August
DIP: 25.0 nM



October DIP: 9.8 nM



P dynamics in the NASG

- DIP and DOP depletion in the NASG
- During the Spring up to 30% of the production in the NASG may be supported by DOP - corresponds to APA
- The bioavailability of DOP is important!!

LETTERS

Phosphorus cycling in the North and South Atlantic Ocean subtropical gyres

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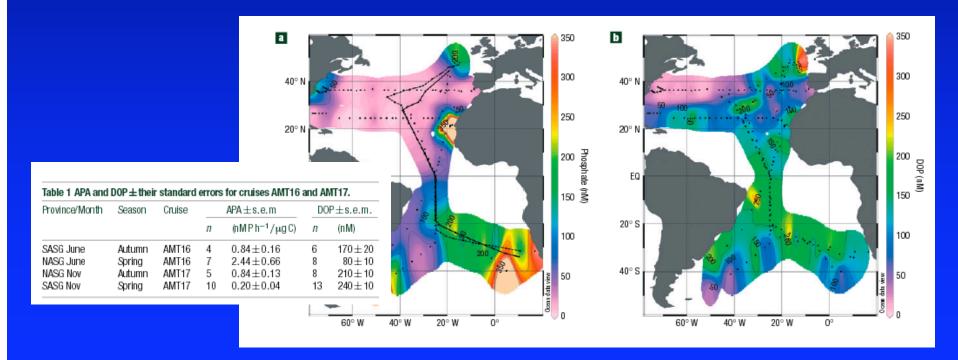
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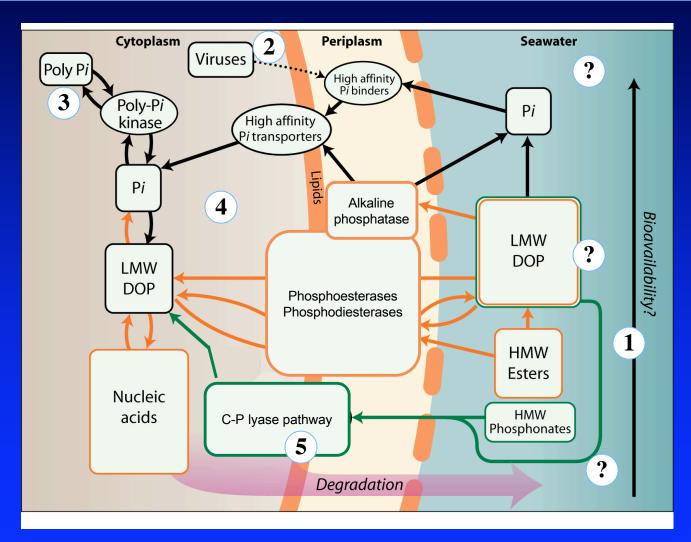
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für Meereswissenschaften, Marine Biogeochemical Modelling, Düsternbrooker Weg 20, D-24105 Kiel, Germany (A.L.)

†e-mail: Wolff@liv.ac.u

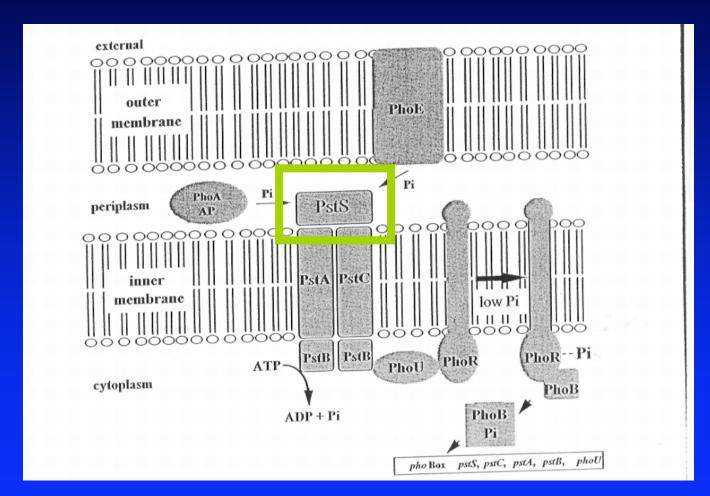


A conceptual diagram of marine microbial interactions with phosphorus



Dyhrman et al. 2007

High affinity phosphate binding (pstS)

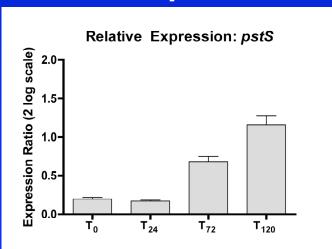


Scanlan and Wilson 2000

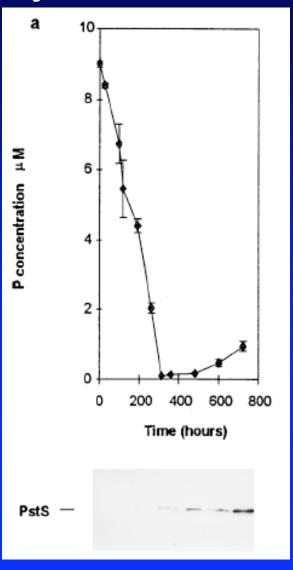
PstS - Synechococcus and Crocosphaera

- Multiple copies of pstS are present in many of the cyanobacterial genomes
- In most but not all cases it is part of the the pstSCAB system for the high affinity transport of phosphate
- The gene and protein are typically upregulated by P deficiency as a part of the pho regulon

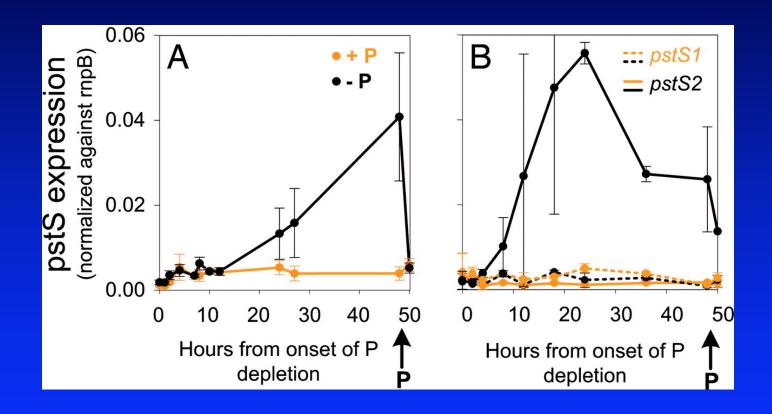
Crocosphaera



Synechococcus

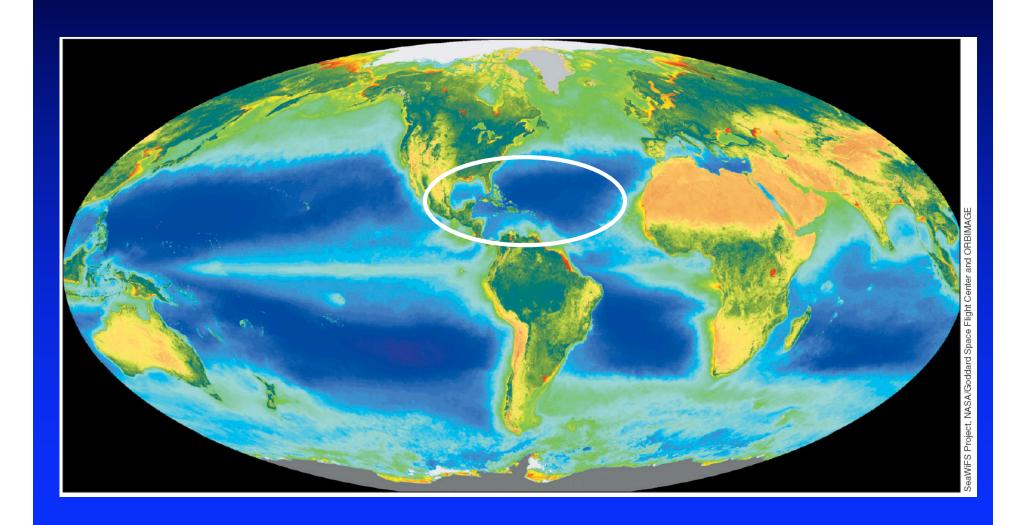


pstS - Prochlorochoccus



Martiny, Adam C. et al. (2006) Proc. Natl. Acad. Sci. USA 103, 12552-12557

NASG: SAR11 pstS



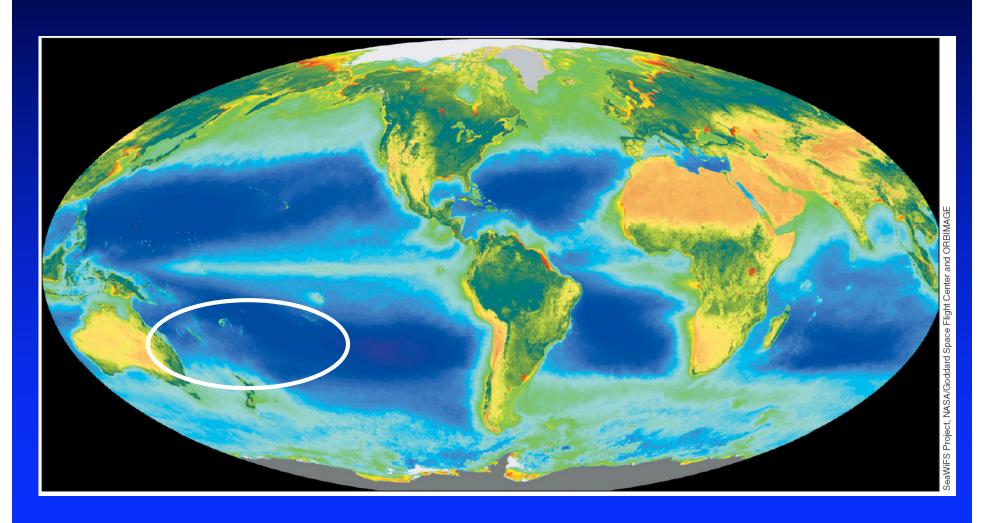
PstS - in the Sargasso Sea

Sowell et al. 2008 ASM General Meeting

PstS - dominant in the Sargasso Sea

Sowell et al. 2008 ASM General Meeting

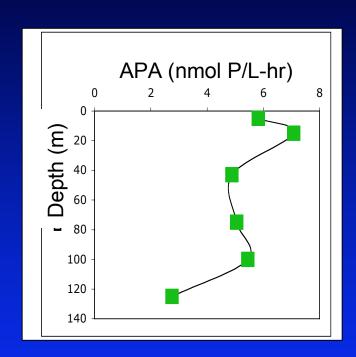
South Pacific: Crocosphaera pstS

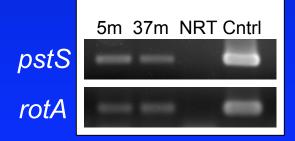


Thanks to Jon Zehr for supporting our cruise participation and for samples!

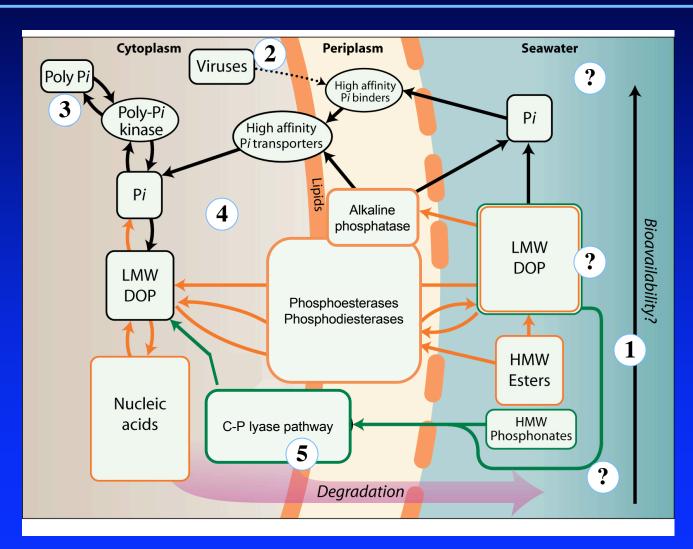
Crocosphaera pstS expression

- St. 25 depth profile:
 - Alkaline phosphatase activity indicates Pmonoester hydrolysis
 - Expression of pstS
 - Both are possible markers of P physiology



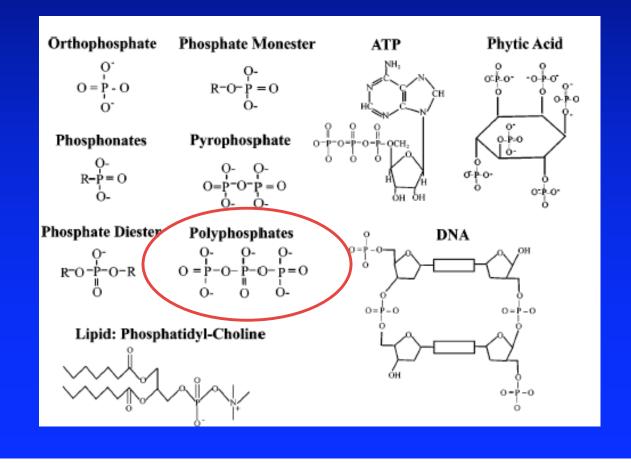


A conceptual diagram of marine microbial interactions with phosphorus



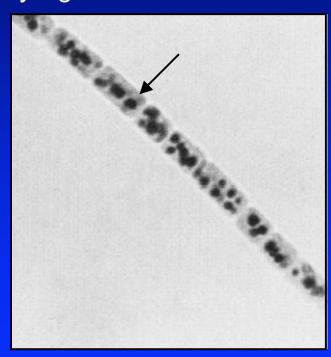
Importance of polyphosphate in P cycling

- Is polyphosphate (polyP) common?
- How is polyphosphate accumulation and degradation regulated?
- Is polyphosphate important to overall P cycling?



Polyphosphate (polyP)

PolyP granules in Anabaena spp.



(Bolier et al 1992)

- Linear chains of P_i ranging in length from 3-1000s residues
- Can form large granules
- Functions:
 - energy reserve
 - ballasting
 - metal chelator
 - osmotic regulation
 - short and long term reservoir for P
- PolyP relatively understudied in marine systems

Key questions

- PolyP dynamics not comprehensively addressed in marine models
- Do Trichodesmium and Crocosphaera have the capacity to synthesize and degrade PolyP?
- Do Trichodesmium and Crocosphaera metabolize PolyP in culture?
- Does *Trichodesmium* produce PolyP *in situ?*
 - Luxury uptake, P overplus, P mining?

PPK

Polyphosphate

Kinase

polyP

Presence of genes in cultured models



Extract DNA



Look for presence of gene with PCR

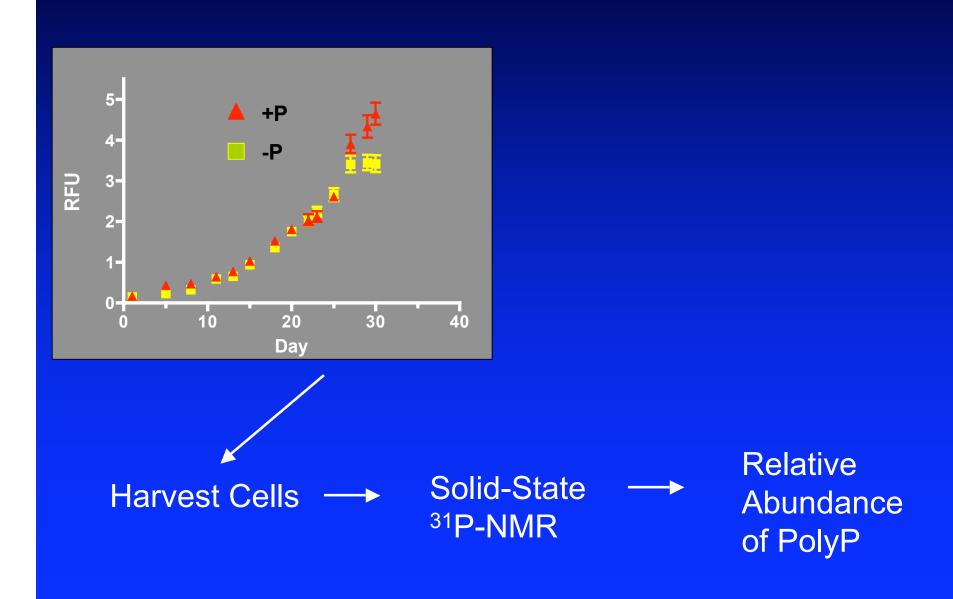


Sequence gene products

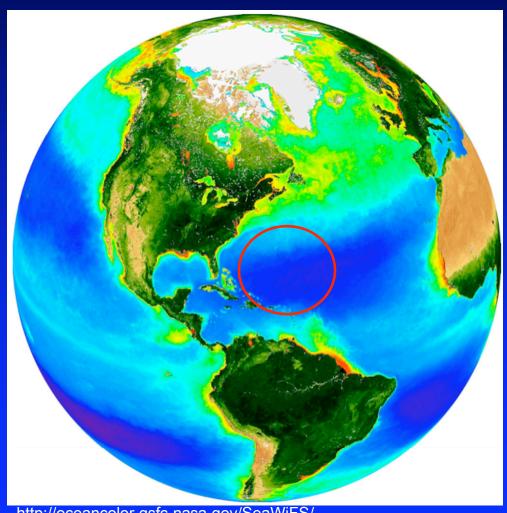
	ррК	ррА	ррХ
Crocosphaera watsonii WH8501	~	✓	>
Trichodesmium erythraeum IMS101	V	>	>
T. spiralis	~	94%*	93%
T. tenue	~	94%	93%
T. thiebautii	V	95%	94%

Trichodesmium and Crocosphaera have the molecular machinery to produce and degrade PolyP

Detecting PolyP in cultures



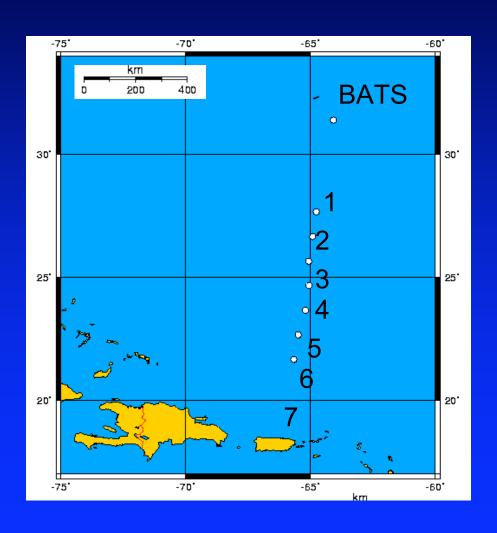
PolyP in oligotrophic systems

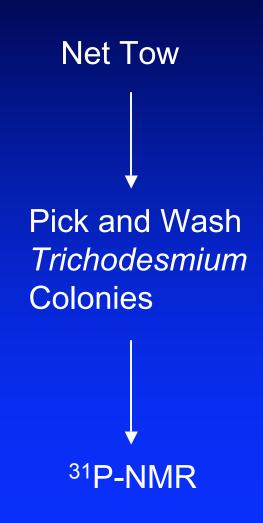


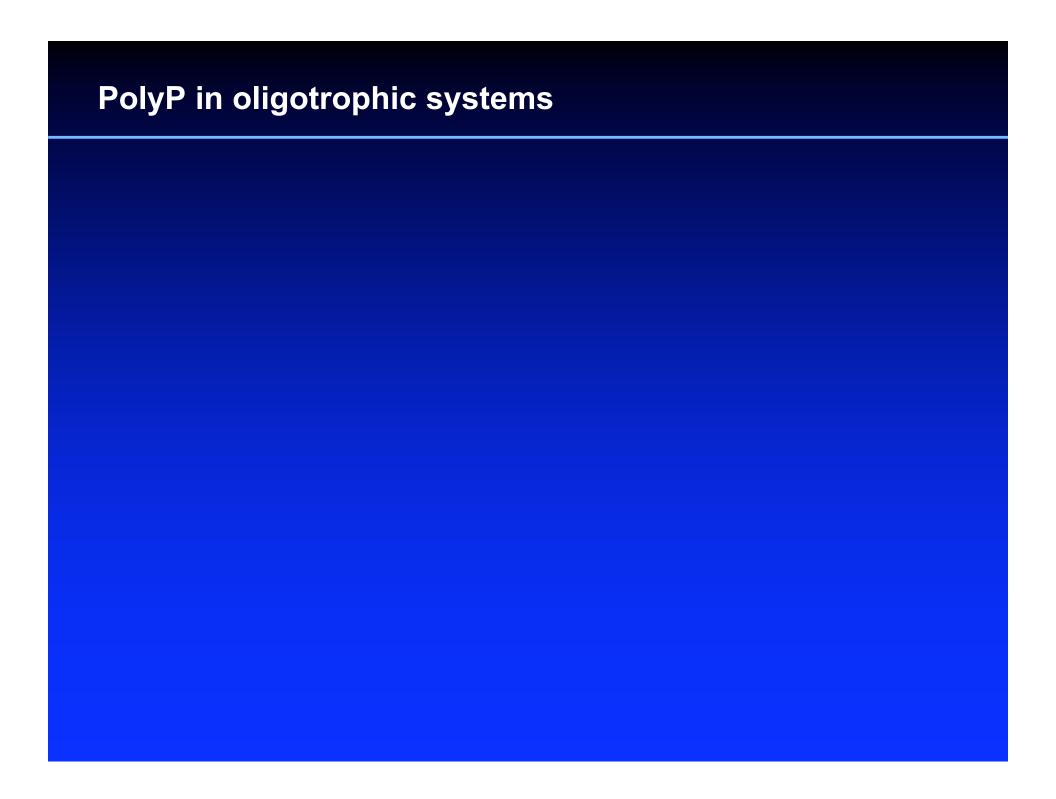
http://oceancolor.gsfc.nasa.gov/SeaWiFS/

- Few studies have looked at PolyP in oligotrophic environments
- NASG is a very low P environment
- Is PolyP detectable in this environment?

PolyP in oligotrophic systems







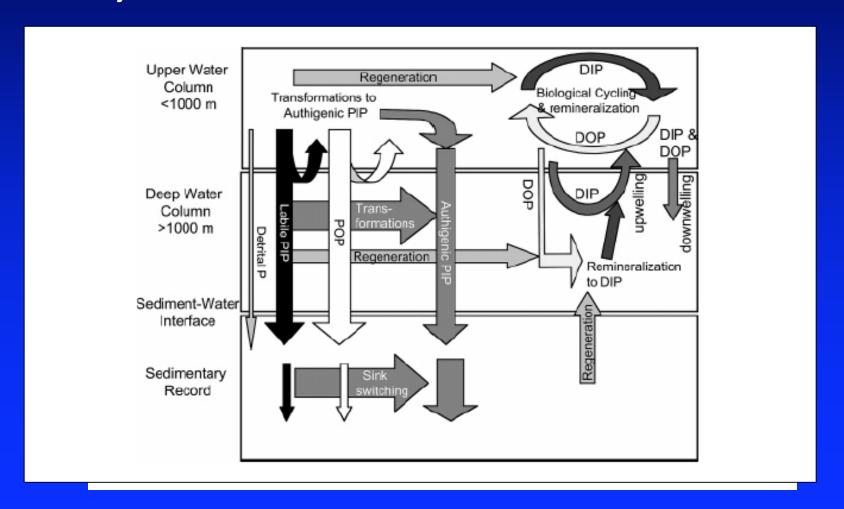
PolyP in oligotrophic systems PolyP may be more important to P cycling and bioavailability - even in oligotrophic environments than previously thought

Geologic P sequestration

•

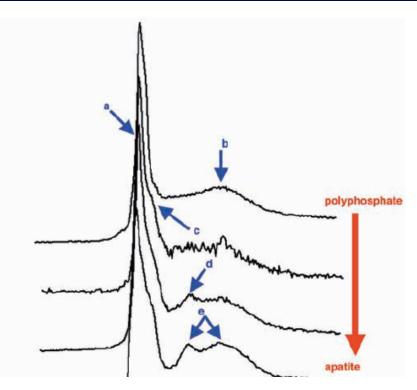
•

 Calcium phosphate minerals (e.g. apatite) are common in marine sediments, and a major P sink



Diagenic transformation of polyP to apatite

Fig. 2. Diagenetic transformation of polyphosphate to apatite. An overlay of phosphorus x-ray fluorescence spectra collected from micrometer-sized phosphorusrich regions in Effingham Inlet sediment illustrates the diagenetic transition from polyphosphate (top) to apatite (bottom). The primary phosphorus fluorescence peak occurs at 2150 eV (a). Spectral features above the primary peak reflect the local bonding environment of phosphorus. Polyphosphate, a simple linear polymer associated with calcium in cells, is characterized by a single peak 18 eV above the primary



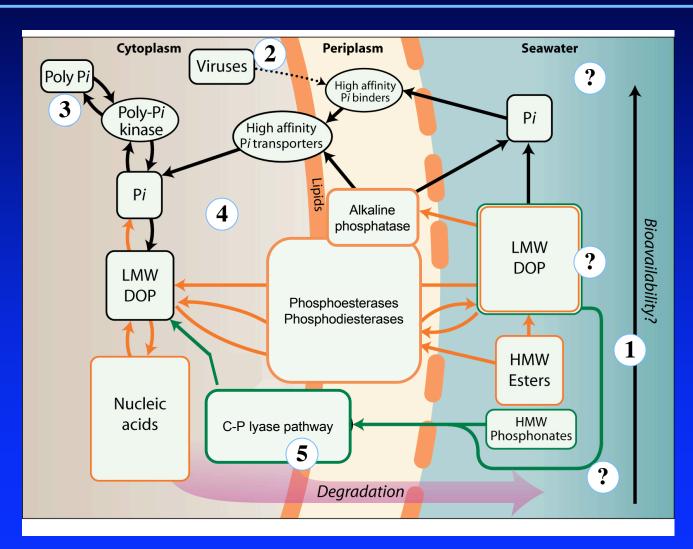
A very recent - and previously unknown transformation of P - that has cycling implications.

crystalline, which may account for the appearance of

Energy (eV)

a primary peak "shoulder" (c). As the crystalline mineral matrix develops further, a peak 11 eV above of the primary peak appears (d), and secondary peaks become more defined (e). The spectra presented in this figure were collected from a single Effingham Inlet sediment sample <3 years of age. Thus, the relative ages of the particles that yielded these spectra are not known.

A conceptual diagram of marine microbial interactions with phosphorus



Intracellular P distribution in microbes

Table II. Percentage of P in Different Chemical Pools in Some Marine Organisms and E. coli

For *E. coli*, content of the components was from Neihardt (Neidhardt, 1987), while total P was estimated as 3.2% of dry weight from Luria (Luria, 1960); average mol wts were from Neihardt for all calculations. Data for *Synechococcus* are from Cuhel and Waterbury (1984). For *Isochrysis* (T-ISO), the carbon (8.09 pg cell-1) and DNA (0.28 pg cell-1) content was obtained from Veldhuis et al., 1997; a Redfield ratio of 106:1 was used to calculate P content. RNA was estimated as three times DNA on a weight basis; lipid at 5.9 pg cell-1 was from Brown, 1991, and 20% (12–25% range) was estimated to be phospholipid (Ben-Amotz et al., 1985). Data for *Chaetoceros* are from Oku and Kamatani (1995).

Component	E. coli	Synechococcus	Isochrysis	Chaetoceros
DNA	9.7	3.2	14	16,4
RNA	61.0	43.8	41	*
Lipid	12.5	0.4	26	1.1
Other				
LPS	5.7			
LMW		51.4		19 (sugar-P)
ATP, NADPH	1.8			1.9
Orthophosphate			•	45.0
Undescribed	9.2	1.2	19	16.6
Total P/cell	0.29 fm	0.21 fm	6.36 fm	180.3 fm

^{*} Nucleic acid pool is 16.4% and is assumed to be RNA and DNA.

Redfield - an average

Living organism (particulate debris) in seawater have similar overall compositions

Average plankton compositions determined by Redfield et al., 1963 ⇒ "famous" Redfield-ratio for the production of **generic** marine organic matter:

106
$$CO_2$$
+16 $HNO_3 + H_3PO_4 + 122 H_2O \Rightarrow \Rightarrow \Rightarrow$

$$(CH_2O)_{106} (NH_3)_{16} H_3PO_4 + 138 O_2$$

Atomic Ratios of the Principal Elements Present in Plankton

	C	N	P
Zooplankton	103	16.5	1
Phytoplankton	108	15.5	1
Average	106	16	1

Considerable plasticity among the phytoplankton

Table 4. Optimum N:P atomic ratios for some freshwater and marine phytoplankton (from Smith 1982 and Kilham and Kilham 1984).

N:P	Species
87	Scenedesmus quadricauda
39	Cryptomonas erosa
30	Scenedesmus obliquus
28	Oscillatoria agardhii
25	Fragilaria crotonensis
24	Chaetoceros affinis
23	Selenastrum capricornutum
21	Ankistrodesmus falcatus
21	Pseudoanabaena catenata
12	Skeletonema costatum
12	Asterionella formosa
10	Synedra ulna
9	Microcystis sp.
7	Melosira binderana



Redfield plasticity

RKR is an "average" value

Many phytoplankton are not average and show considerable C:N:P placticity

For example: C:N:P Ratios of *Chlorella* growing under different nutrient regimes (Ketchum and Redfield, 1949)

	С	N	Р	N:P
Normal	47	5.6	1	6
P-deficient	231	30.9	1	31
N-deficient	75	2.9	1	3

Flexible stoichiometry in *Trichodesmium*

Limnol. Oceanogr., 51(4), 2006, 1777-17.

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Flexible elementa

AQUATIC MICROBIAL ECOLOGY Aquat Microb Ecol

Published March 29

Angelicque E. White College of Oceanic and

David M. Karl School of Ocean and Ea

Ricardo M. Letelier College of Oceanic and

Plasticity of N:P ratios in laboratory and field populations of *Trichodesmium* spp.

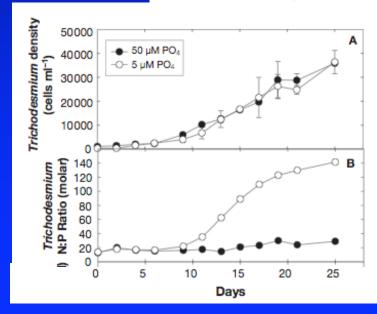
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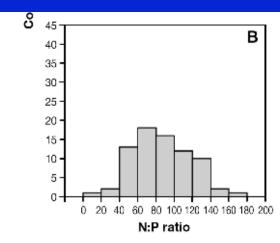


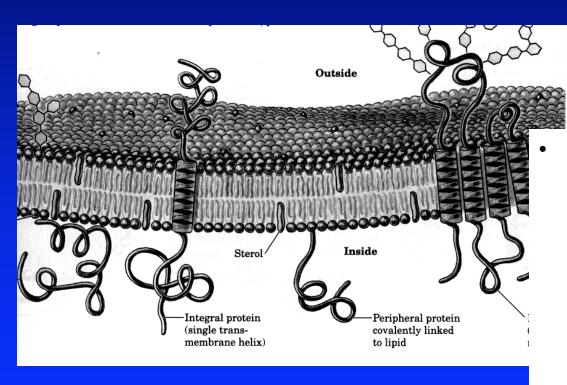
Fig. 5. Trichodesmium spp.. N:P ratios of (A) floating and (B) sinking colonies (Gulf of Mexico)

Flexible stoichiometry - how?

- P is locked up in DNA, RNA, lipids, and other molecules which may or may not be flexible
- One long term strategy would be to have a small genome, given the P locked up in nucleic acids
- But that doesn't explain short-term physiological plasticity
- Variable stores of polyP are a possibility
- Another possibility is alternative molecules serving to conserve P

Flexible stoichiometry - how?

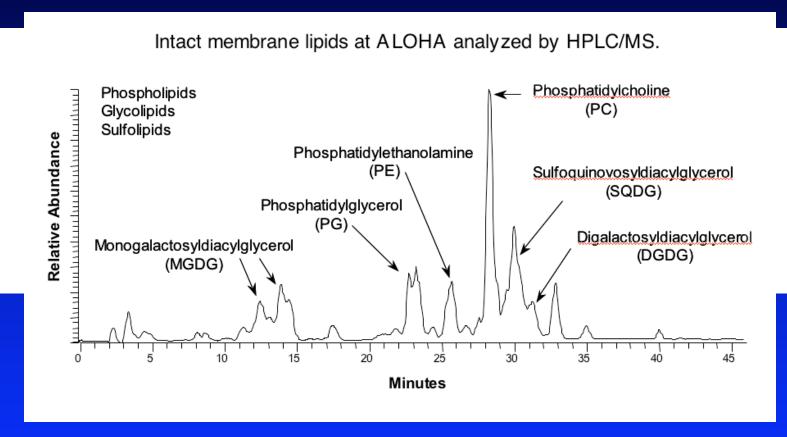
- Phospholipids
 - Are found primarily in cell membranes.
 - Composed of a polar "head" group and a nonpolar "tail" group.



- Three basic types of phospholipids in phytoplankton.
 - Phosphatidylglycerol (PG).
 - Phosphatidylethanolamine (PE).
 - Phosphatidylcholine (PC).

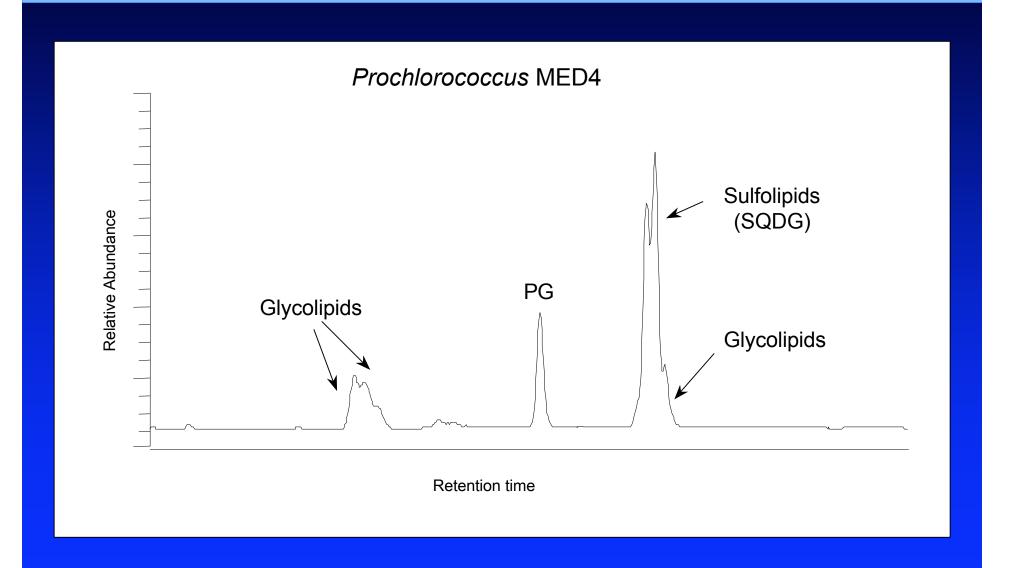
Phospholipids and some alternatives

PNAS



Seawater was filtered on $0.2\mu m$ anodisc membrane, which was then extracted by Bligh and Dyer. Intact polar lipids were identified and quantified by HPLC/MS.

Phospholipid alternatives in *Prochlorococcus*

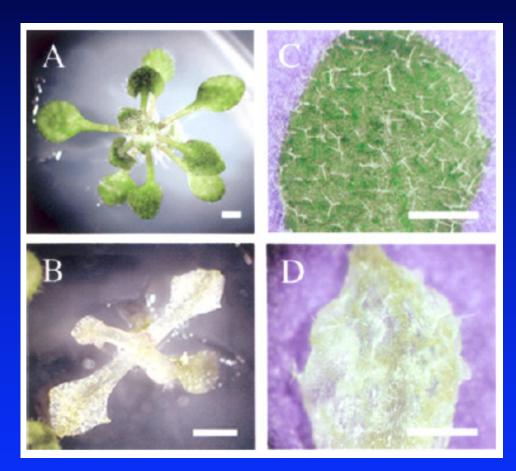


SQDG can be used as a "substitute" for PG...

SQDG and PG are the only anionic lipids in photosynthetic membranes, and may, in principal, serve the same functions (Benning et al., *Proc. Natl. Acad. Sci.*, 1993).

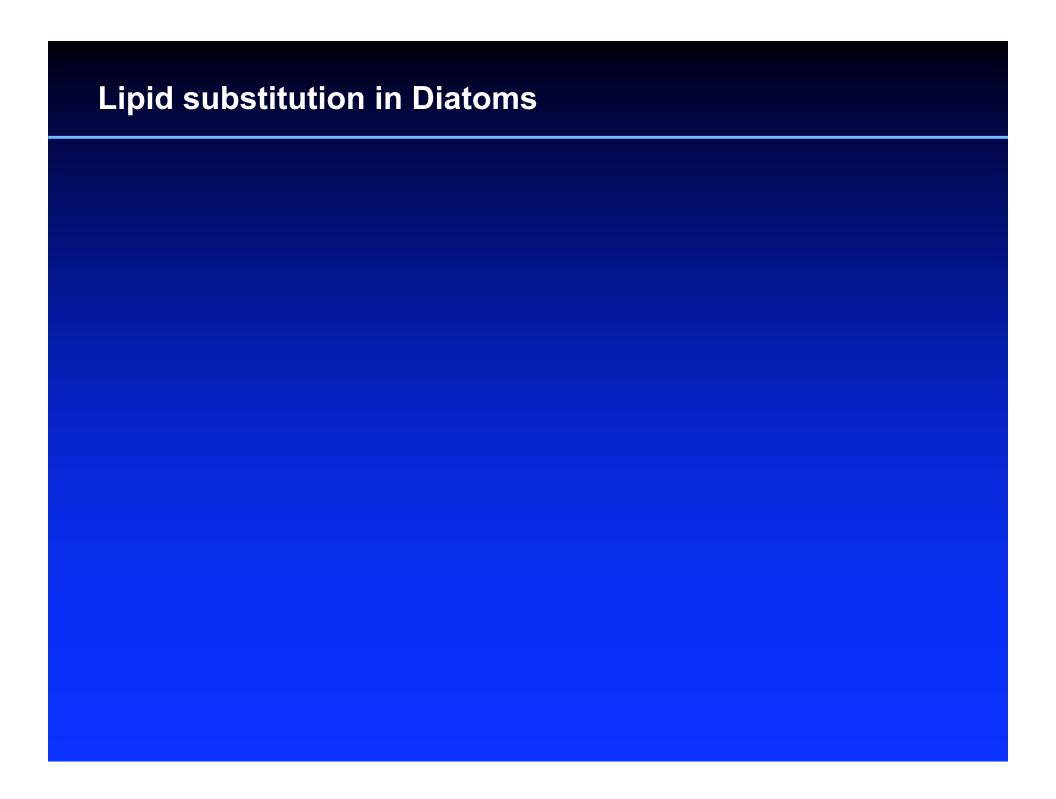
...but there are costs for using SQDG instead of PG

- The recognition of PG/SQDG interchangeability led to burst of basic research with cyanobacteria and plants in the 1990's.
- •PG composes >30% of total P in corn.
- •10-20 papers describing numerous mutants defective in PG and/or SQDG.
- Substituting SQDG for PG negatively impacts both PS I and PS II because PG is slightly more effective at stabilizing chlorophyll and pigments than SQDG (e.g. the substitution incurs costs).



Hagio et al., Plant Cell Physiol., 2002.

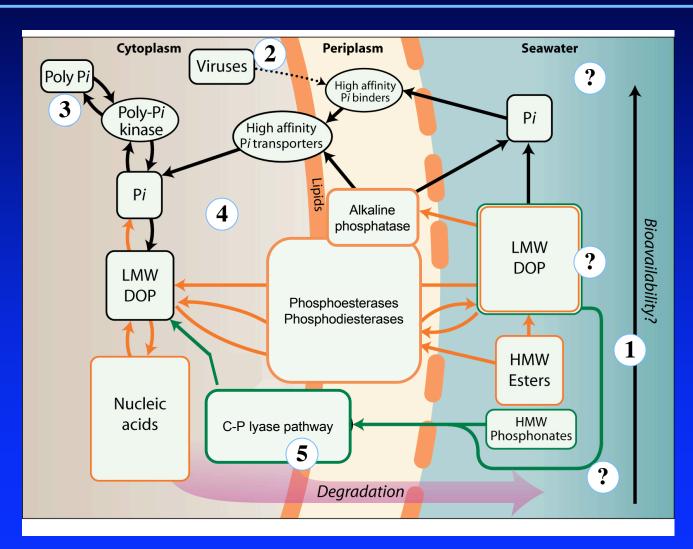




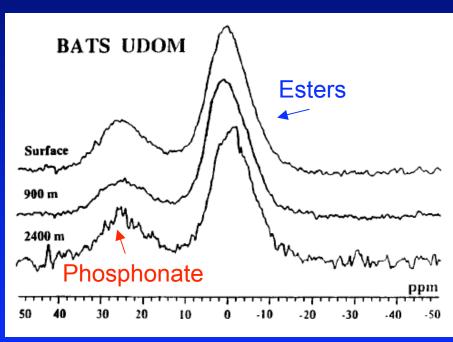




A conceptual diagram of marine microbial interactions with phosphorus

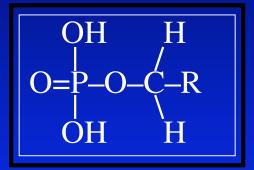


³¹P NMR of UDOM in the Sargasso Sea



Kolowith et al. 2001

Phosphomonoesters

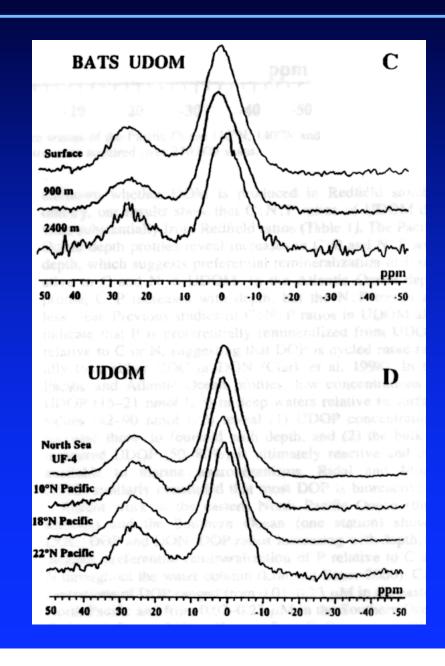


Phosphonates

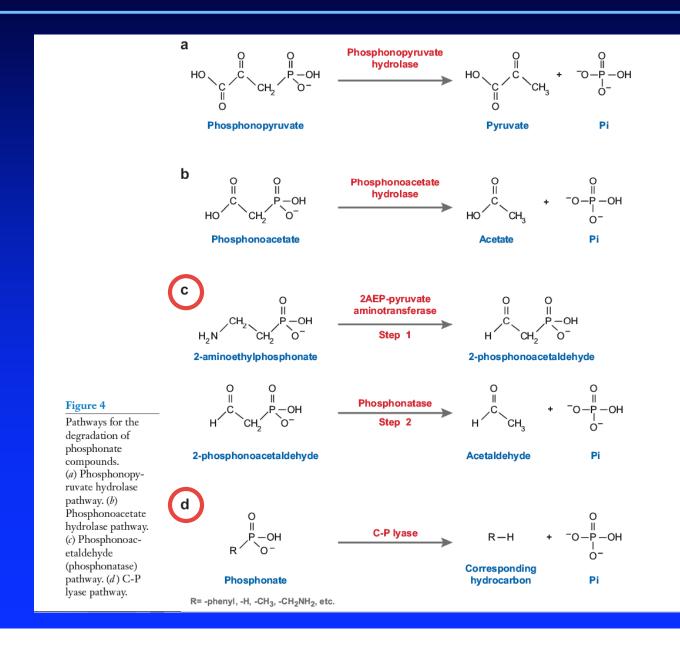


P bond class distribution

- DOP by <u>solid</u> state ³¹P NMR of UDOM (high molecular weight only)
 - DOP is composed of primarily of phosphoesters (75%).
 - Also, phosphonates (25%).
 - The ratio of phosphoesters to phosphonates is fairly constant regardless or environment or depth
 - Esters (C-O-P) are thought to be much more labile then phosphonates (C-P)



Pathways of phosphonate metabolism

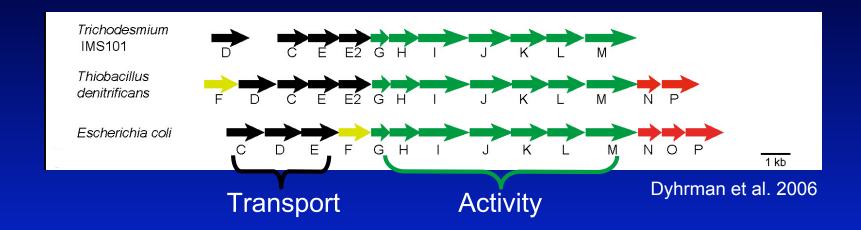


phnA

phnX/W

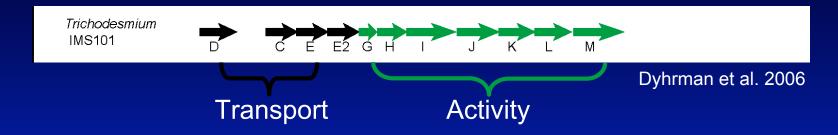
phnG-M

P acquisition genes in Trichodesmuim erythraeum



 All phn genes required for C-P lyase mediated phosphonate metabolism are present in the IMS101 genome

P acquisition genes in *Trichodesmuim*



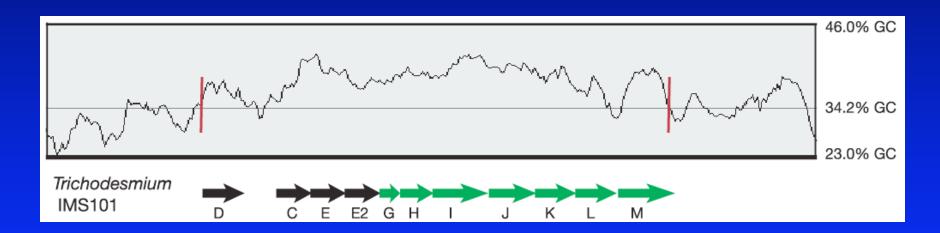
 The phnD and phnJ genes are present in other Trichodesmium species.

Species	pstS	phoA	Xqq	phnD	phnJ
T. erythraeum	X	Y	Y	Y	Y
T. theibautii	X	X	Y	Y	Y
T. tenue	V	V	V	V V	\ \ \ \ \ \
* 98-100% identic		<u> </u>		^	

• C-P lyase gene cluster is not represented in any of the marine *Syn*, *Pro* or *Croco* genomes examined to date.

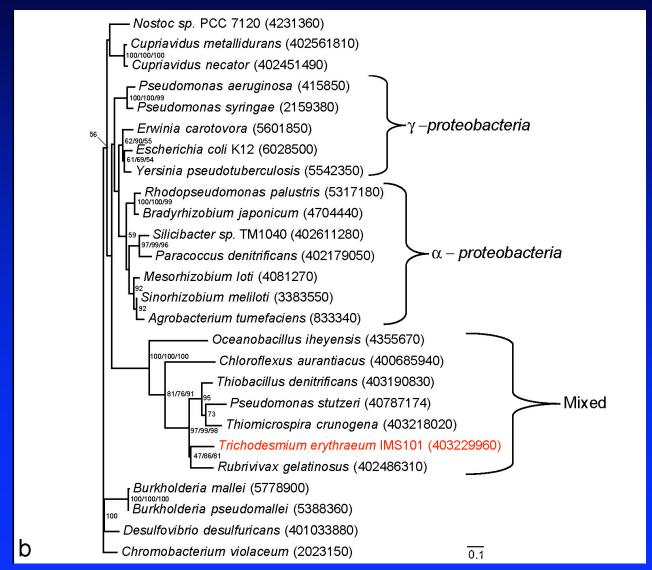
Possible gene transfer...?

 The GC content is elevated in the phn cluster relative to the rest of the genome



The phnJ gene clusters with distantly related bacteria.

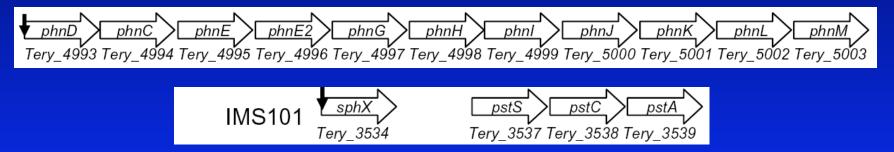
phnJ phylogeny



(Dyhrman et al. 2006 - Thanks to Dr. Eric Webb!)

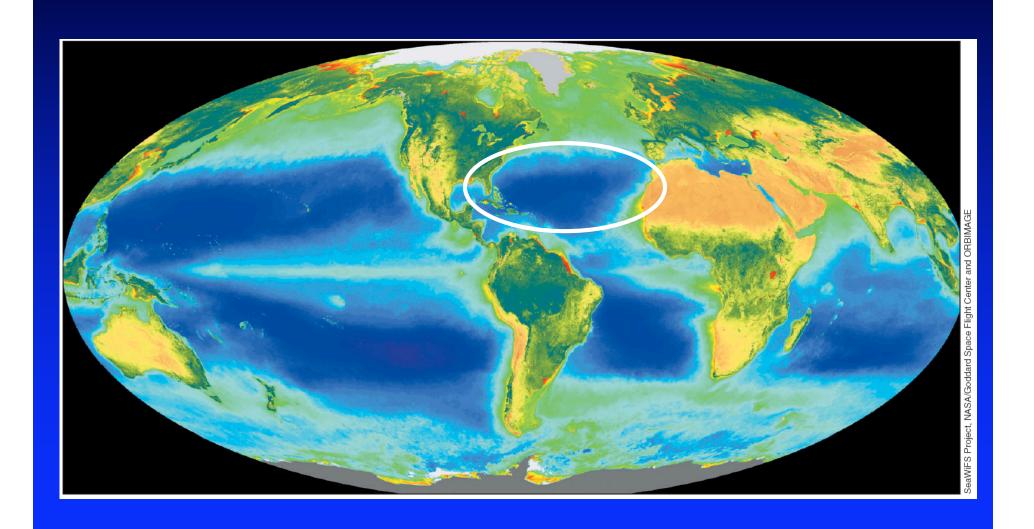
P regulation

- Expression studies
 - Su et al. 2007 identify pho boxes for the phn and pstS gene clusters.

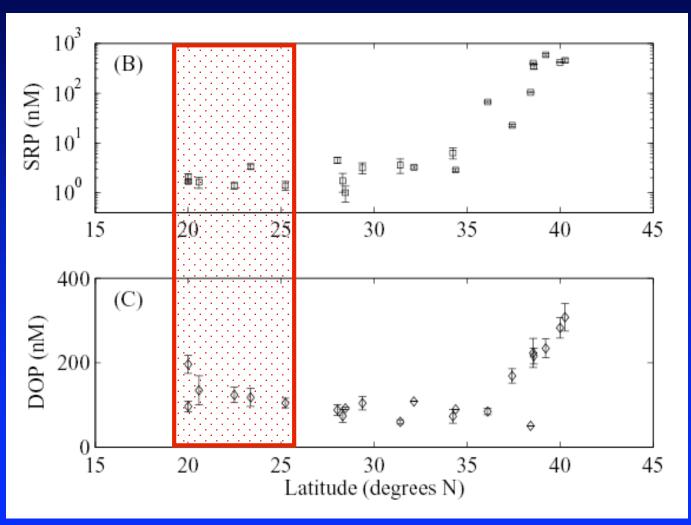


- Target genes appear Pregulated.
- Quantitative approaches under development to examine the timing of the response.

Sargasso Sea: Trichodesmium P acquisition

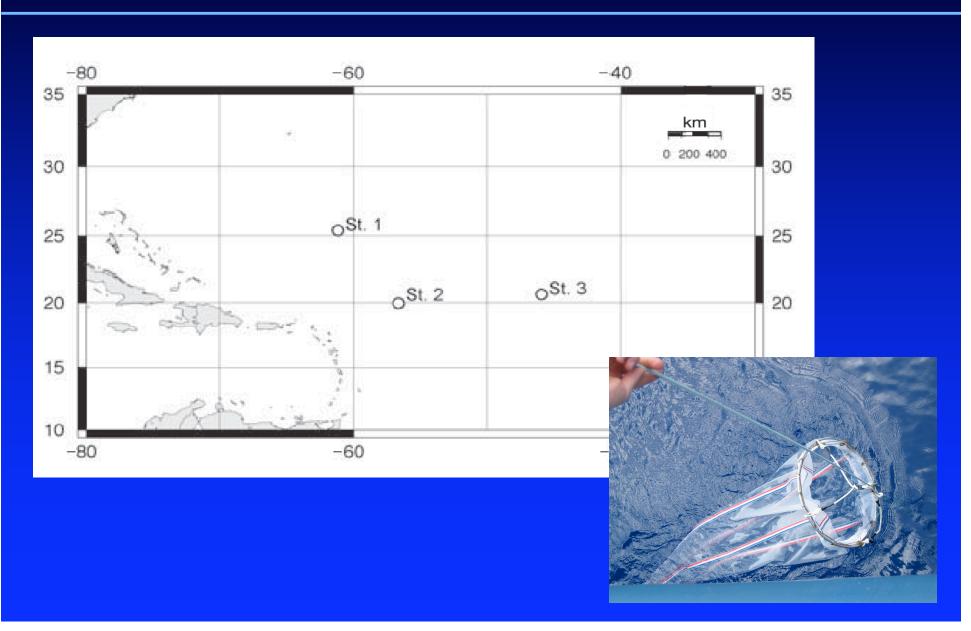


Sargasso Sea: P biogeochemistry

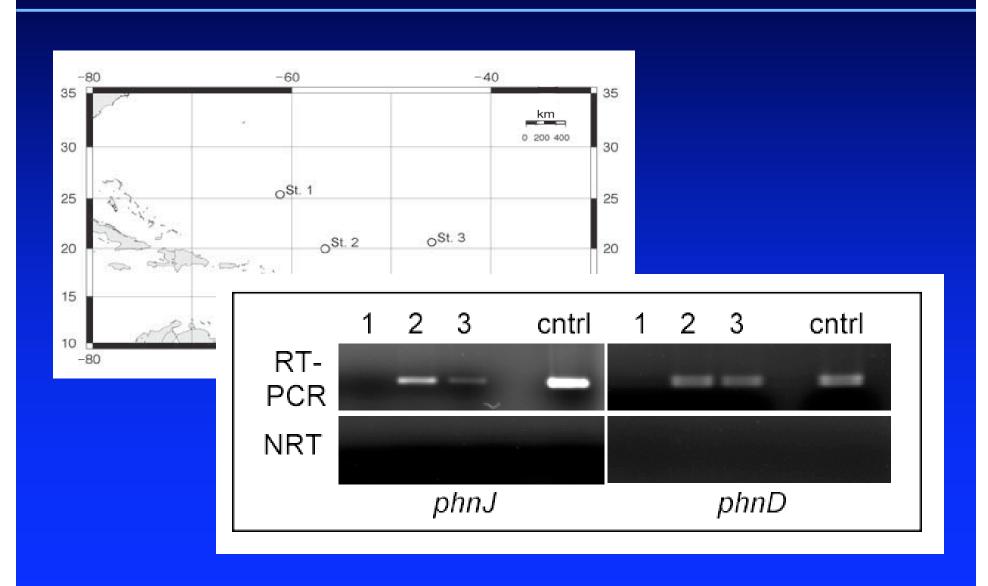


Wisniewski Jakuba et al. In press

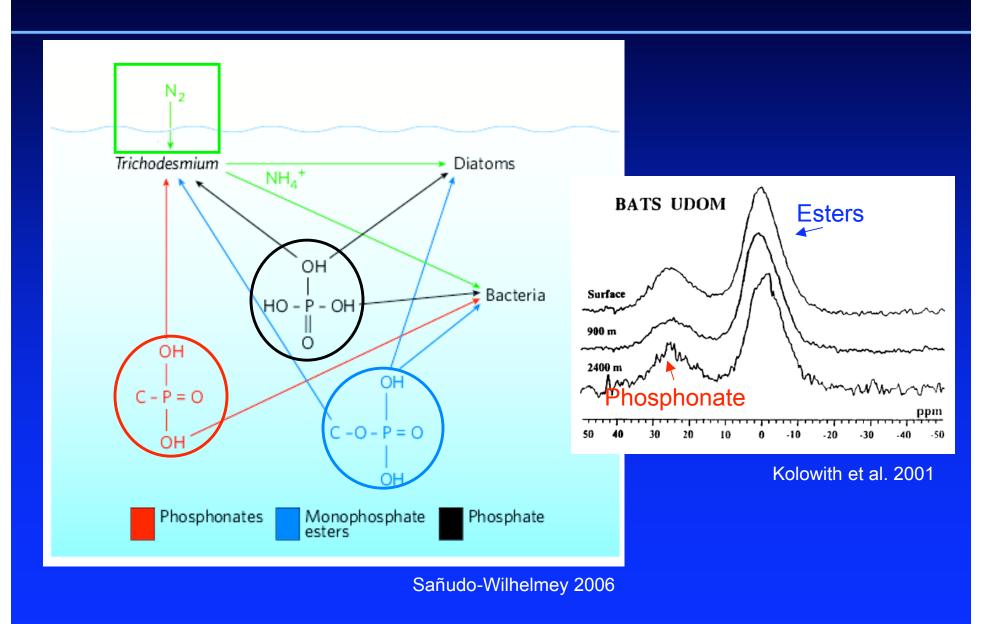
Does *Trichodesmium* express phosphonate metabolism genes in the field?



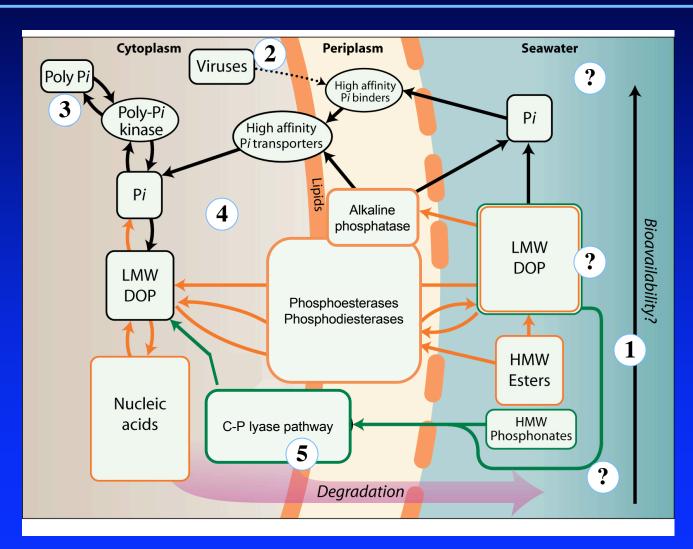
phnJ and phnD are differentially expressed in field populations



Geochemical niche adaptation in Trichodesmium



A conceptual diagram of marine microbial interactions with phosphorus



The composition of low molecular weight DOP?

SNP compound class	%Composition (method) ^a	Reference			
Bulk SNP					
Monophosphate Esters	10-100% (Enzymatic Assays)	Strickland and Solorzano, 1966;			
		Kobori and Taga, 1978; Taft et al.,			
		1977; Chrost et al., 1986			
	55-77% (0-100 m; Modified UV oxidation)	Karl and Yanagi, 1997			
	50% (> 100 m; Modified UV oxidation)				
Nucleotides and	23-45% (0-100 m; ≈ Persulfate — modified UV)	Karl and Yanagi, 1997			
nucleic acids	50% (> 100 m; ≈ Persulfate — modifiedUV)	_			
	10-100% (Enzymatic Assays)	Strickland and Solorzano, 1966;			
		Kobori and Taga, 1978; Taft et al.,			
		1977; Chrost et al., 1986			
	ATP: < 1% (Firefly bioluminescence)	Azam and Hodson, 1977; Azam et			
		al., 1979; Hodson et al., 1981;			
		Nawrocki and Karl, 1989			
	DNA/RNA: < 5% (Multiple methods)	DeFlaun et al., 1986; Paul et al.,			
		1986; Karl and Bailiff, 1989			
Phospholipids	3–11% (Cross Flow Filtration (CFF) and	Suzumura et al., 1998			
	Polymyxin B treatment)				
Phosphonates	5-10% (31P NMR)	Clark et al., 1998			
Polyphosphates	0-50% (≈ Acid Reflux — UV oxidation)	Armstrong and Tibbets, 1968;			
		Solorzano and Strickland, 1968;			
		Solorzano, 1978			
Size fractionated SNP					
LMW (< 10 kDa)	50-80% (CFF)	Matsuda 1985; Ridal and Moore, 1990;			
		Suzumura et al., 1998			
HMW (> 10 kDa)	20-50% (CFF)	Matsuda et al., 1985; Ridal and Moore, 1990;			
HMW (> 50 kDa)	15% (CFF)	Suzumura et al., 1998			
Monophosphate esters	10% (Alkaline phosphatase treatment)	Suzumura et al., 1998			
	75% (³¹ P NMR)	Clark et al., 1998			
Nucleic acids	25% (Phosphodisterase treatment)	Suzumura et al., 1998			
Phospholipids	38-46% (Polymyxin B treatment)	Suzumura et al., 1998			
Phosphonates	onates 25% (³¹ P NMR) Clark et al., 1998				

The composition of low molecular weight DOP?

Viral pstS?

Table 5. Summary Table of Unique Features of Prochlorococcus Cyanophage Genomes That Are Uncommon among Known Phages

Three Prochic Signature Feat Interpretation

Matthew B. Sullivan¹, Maureen L

Joint Program in Biological Oceanography, W. America, 2 Department of Civil and Environme of Biology, Massachusetts Institute of Technok California. United States of America

The oceanic cyanobacteria Proch that their viruses (phages) med present an analysis of genomes t overall genome features, and ge T4-like (P-SSM2 and P-SSM4) pl genome sequences to establish T7-like genes and the two myov each genome contains a signific genes found in cyanobacteria. So three phage genomes contain pl activity during infection, as wel metabolism during infection. Tl suggest it is capable of integrat phages or marine cyanophages host. Further, both myoviruses phage and host responses to pl cyanophages appear to be varia reflect adaptations for infection

Citation: Sullivan MB, Coleman ML, Weigele P, F PLoS Biol 3(5): e144.

Fu	nctional Category	Genes	Putative Function	P-SSP7	e-Value	Marine T7-Likes	P-SSM2	e-Value	P-SSM4	e-Value	Marine T4-Likes
Pho	osphate	pstS	Phosphate uptake				322	e ⁻¹³⁶	322	e ⁻¹³⁷	
		phoH	Phosphate-stress-induced			+	251	e ⁻²⁴	258	e ⁻²⁰	+
Car	rbon mobilization	talC	MipB/TalC family ************************************	215	e ⁻⁴³		216	e ⁻⁴⁷	218	e ⁻⁵³	+
Lys	ogeny	int	Phage integration	291	e ⁻¹³						
Nu	cleotide	mazG	pyrophosphohydrolase/				139	e ⁻¹¹	134	e ⁻²⁷	
me	tabolism		pyrophosphatase								
		pyrE	Orotate phosphoribosyltransferase				215	e ⁻⁴⁴			
		purH	Phosphoribosylformyl glycinamide synthase				108	e-7			
		purL	Phosphoribosyl formyl glycinamide cyclo-ligase				223	e ⁻⁸⁰			
		purM	AICARFT/IMPCHase bienzyme				314	e ⁻⁹⁷			
		purN	phosphoribosyl glycinamide formyltransferase				175	e ⁻³³			
		nrd	RNR domain	469	e-11	+	universal a	mong T4-lik	e phages		
Pho	otosynthesis-	psbA	D1 protein, PSII	360	e=0		361	e = 0	366	e = 0	+
rela	ated genes										
		hli	Thylakoid-associated proteins	×1 hli ger	ne		× 6 hli ger	nes	×4 hli gen	es	+
		petE	Plastocyanin, PET				115	e ⁻²¹			
		petF	Ferredoxin, PET				98	e ⁻²⁹			
		pebA	Phycoerythrobilin biosynthesis				234	e ⁻¹²			
		ho1	Heme biosynthesis				234	e ⁻⁶³			
		psbD	D2 protein, PSII						359	e = 0	+
		speD	Polyamine biosynthesis						102	e ⁻¹⁷	
		рсуА	Phycocyanobilin biosynthesis						230	e ⁻²⁶	
Oth	her functions	cobS	Vitamin B12 biosynthesis				365	e ⁻²¹	365	e ⁻²³	
		prnA	Bacterial tryptophan halogenase					e ⁻⁵²			
		nol	Carbomoyltransferase				572	e ⁻⁵²			
		hn	HN						158	e ⁻³³	
		LPS	Epimerases, transferases, phospholipases				×24 genes	5			

Non-marine T7-like/T4-like phages completely lack these genes. The size (amino acids) and best BLASTp e-value of each predicted coding region are presented using gene names and function assignments according to their function in cellular organisms. The hil genes were assigned using e-value and a signature sequence as reported in Lindell et al. [14]. A plus sign indicates that the feature is present in the phage group, otherwise the feature is absent or is yet to be identified. PET, photosynthetic electron transport: PSIL, photosystem II reaction center.

DOI: 10.1371/journal.pbio.0030144.005

- The composition of low molecular weight DOP?
- Viral pstS?
- A P redox cycle ?

Microbial Metabolism of Reduced Phosphorus Compounds

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Annu. Rev. Microbiol. 2007. 61:379-400

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Metabolism of reduced P

Compounds with P valence of +5 O-P-OH O-P-OR Phosphate Phosphate ester Phosphoethanolamine Compounds with P valence of +3 O-P-R Ō **Phosphite** Phosphonate Aminoethylphosphonate Compounds with P valence of +1 -O-P-R Hypophosphite Phosphinate Phosphinothricin P valence -1 P valence -3

Phosphine

Phosphine oxide

Figure 1

The chemical structures of phosphorus compounds at various redox levels. Inorganic compounds, a generalized formula, and a known biological example are shown in the left, middle, and right columns, respectively, for compounds in the +5, +3, and +1valence states.

Growth on hypophosphite

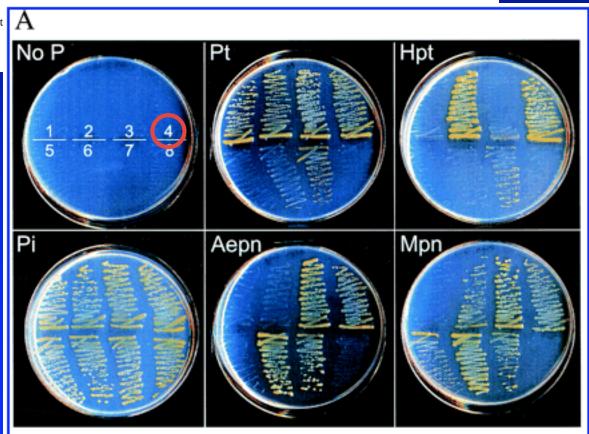
Journal of Bacteriology, July 2004, p. 4730-4739, Vol. 186, No. 14 0021-9193/04/\$08.00+0 DOI: 10.1128/JB.186.14.4730-4739.2004 Copyright © 2004, American Society for Microbiology. All Rights Reserved.

Two C—P Lyase Operons in *Pseudomonas stutzeri* and Their Roles in the Oxidation of Phosphonates, Phosphite, and Hypophosphite

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Received 30 January 2004/ Accepted 5 April 2004



Growth on hypophosphite

SUMMARY POINTS

- 1. A wide variety of reduced P compounds are produced and consumed by microbes.
- Bacterial synthesis of many medically important reduced P antibiotics involves the production of reduced P phosphonate and phosphinate intermediates that could provide an important source of P in the environment.
- 3. Four bacterial pathways for the oxidation of the inorganic reduced P compound, phosphite, have been identified: C-P lyase, BAP, NAD:phosphite oxidoreductase, and a novel but as yet uncharacterized pathway for deriving energy from phosphite oxidation in D. phosphitoxidans

With the variety of pathways to oxidize phosphite to phosphate...elucidation of a biological P redox cycle may not be far off.

FUTURE ISSUES

- 1. Attaining in vitro C-P bond cleavage activity by C-P lyase will allow the reaction sequence and mechanism to be fully described.
- 2. In vitro characterization of hypophosphite oxidation via htxXY in X. flavus will provide insight into this novel reaction.
- 3. Determining the source(s) of phosphine, hypophosphite, and phosphite in the environment would significantly add to our understanding of P availability to organisms.

- The composition of low molecular weight DOP?
- Viral pstS?
- A P redox cycle ?
- Sources and sinks of marine phosphonates?
 - Phosphonates have been identified in a few marine invertebrates, but but the source of phosphonates to the upper water column remains a mystery

The take home message

- The basics:
 - P is important
 - It is rapidly cycled
 - It comes in organic forms that are poorly characterized
- Tech advances
 - Magic tells us that the inorganic form is at very low levels
 - Distribution of bond classes in high molecular weight DOP
- Adaptation to low P is common
 - Frequency and expression of pstS
 - The emerging importance of polyphosphate
 - Losing your phospholipids
 - Metabolism of phosphonates
- Enduring mysteries
 - Low molecular weight DOP composition
 - P machinery in the agal viruses
 - Microbial metabolism of reduced P

